

A Review of Modern Methods of Synthesis 1, 3, 4-Oxadiazole as a Bioactive Compounds

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ABSTRACT: Oxadiazole ring is a heterocyclic molecule with an oxygen and two nitrogen atoms spread throughout its five-membered structure. There are four different isomers that have been discovered, Because of their wide applications in a range of sectors, including medications. Some of these biological activities are; anticonvulsant capacity, anticancer as well, antibacterial, antiviral, antifungal, antimalarial, antitubercular, anti-asthmatic, antidepressant, antidiabetic, antioxidant, antiparkinsonian, analgesic and anti-inflammatory, are just some of the therapeutic uses that have drawn attention to drug candidates containing an oxadiazole moiety. This review, we will examine the various methods of oxadiazole synthesis. The molecular docking of some oxadiazole compounds has been studied to investigate the active derivatives and to evaluate their activity. The synthesis of the oxadiazole ring has sparked a lot of attention since then. A large number of oxadiazole derivatives, as well as methods, were reported New antimicrobial drugs have been developed from a number of different areas in recent years in an effort to reduce the prevalence of drug-resistant bacteria. Furthermore, this review touches upon the importance of structural modification in fine-tuning the biological activities of 1,3,4-oxadiazole derivatives. By altering the substituents and the position of functional groups, researchers can tailor the pharmacological properties to target specific diseases or conditions, making them highly versatile and attractive in drug discovery.

Keywords: 1,3,4-Oxadiazole, Molecular Docking, biological activities.



1. INTRODUCTION

The five-member ring heterocyclic's exhibited various biological activities and among them are the 1,3,4-oxadiazole. furthermore, The oxadiazole and their 2,5-disubstituted derivatives are good for development materials due their large spectrum of various biological activity like, anti-inflammatory[1], anticancer , antibacterial [2][3] Antifungal [4][5] antihypertensive activities[6] , antitubercular [7] anti-HIV[8] , antimicrobial[9]. Insecticidal [10][11], Anesthetic Activity[12], Chronic rhinosinusitis [13]Molecular Docking [14]and antioxidant activities[15]moreover, bisoxadiazole compounds was reported as a biological active compounds [16] ,also it show some physical property like electrochemical properties[17], electro-optical properties [18] and Luminescence property[19]. Inhibition of breast cancer cell growth, photosensitization of TiO₂nanoparticles, control of rice sheath blight and sorghum anthracnose, and inhibition of *Xanthomonas oryzae* pv. *oryzae* (Xoo) using an in vitro turbidimeter assay[20], anti-tubercular agent[21], nematocidal activities, nematocidal activities[22].

The 1,3,4-oxadiazole heterocyclic scaffold has emerged as a fascinating and versatile class of compounds with a wide range of biological activities, making them highly sought-after in the realm of medicinal chemistry. These compounds have displayed remarkable potential in various therapeutic applications, such as anti-inflammatory, anti-cancer, anti-microbial, and anti-viral activities. The diverse biological activities exhibited by 1,3,4-oxadiazoles have spurred extensive research into developing efficient methods for their synthesis. This multitude of synthetic pathways reflects the urgency and significance of harnessing the unique pharmacological properties of 1,3,4-oxadiazoles for the development of novel pharmaceuticals. In this context, numerous innovative synthetic methods have been devised to access these compounds, each tailored to enhance yield, selectivity, and sustainability. This introduction sets the stage for an exploration of the

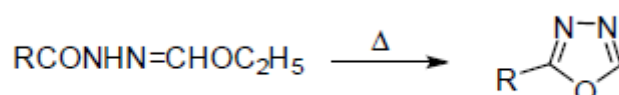
diverse biological activities of 1,3,4-oxadiazoles and the various synthetic methodologies that have emerged to harness their potential for drug discovery and development.

HIV (human immunodeficiency virus)
XXO (*Xanthomonas oryzae* pv. *Oryzae*)

1.1 MATERIALS AND METHODS

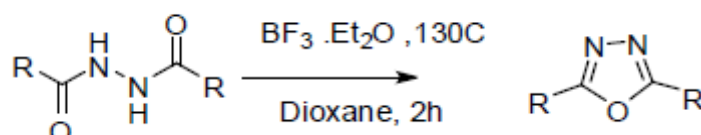
Literatures display several variety methods to synthesize 2,5 di-substitute -1,3,4-oxadiazole. Despite the wide range of approaches, the majority of these studies focused on the condensation of carboxylic acid and acid hydrazide. The first condensation to make heterocyclic reported in 1943 from R. A. Labriola and A. Felitte and in 1955 two individual literatures report the synthesis of 1,3,4-Oxadiazole, below demonstrated for some these methods.

- Ainsworth and Hackler synthesised alkyl oxadiazole by heating 1-Acyl-2-ethoxymethylene hydrazines under ambient conditions [23].



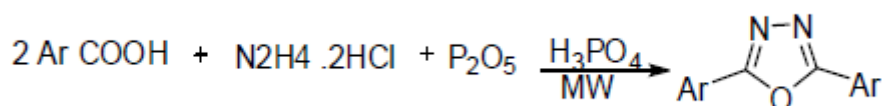
Scheme 1. Synthesis of oxadiazole

- Reaction of 1,2-diacyl and 1,2-diaroyl hydrazines with $\text{BF}_3 \cdot \text{Et}_2\text{O}$, as cyclodehydration reagent [24].



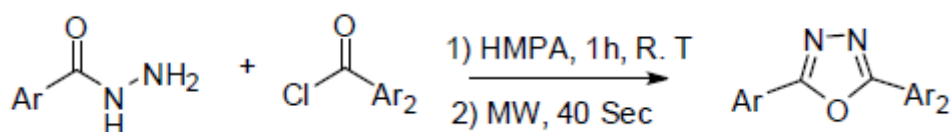
Scheme 2. Synthesis of 2,5-dialkyl oxadiazole

- Reaction of two mole of Aryl acid with Hydrazine dihydrochloride in presence of phosphorus pentoxide and phosphoric acid, 2-5 minutes (60 W) at 130 °C in a microwave oven with a reflux condenser [25].



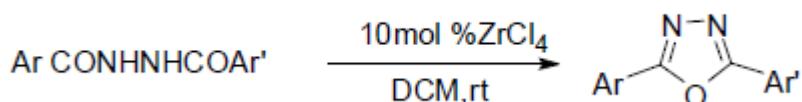
Scheme 3. Route of synthesis diaryl oxadiazole

- single aryl-hydrazides were condensed with acid chlorides in the presence of HMPA and microwaved to produce specific 2,5-disubstituted-1,3,4-oxadiazoles [26].



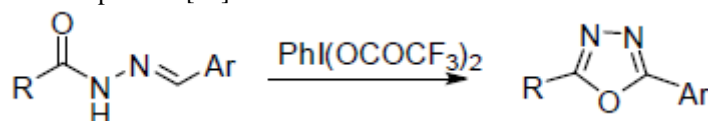
Scheme 4. Route of synthesis 2,5-disubstituted-oxadiazoles

- For the cyclodehydration of di-acyl-hydrazine, by mild catalyst that has been shown to be effective is zirconium(IV) chloride (10 mole%) [27].



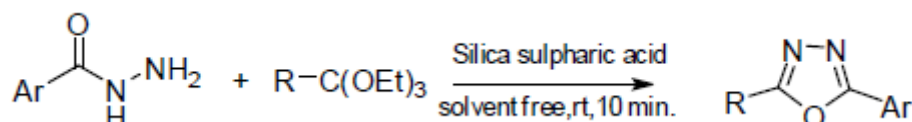
Scheme 5. Synthesis of oxadiazole

- Utilizing oxidizing Factor such as bis-(tri-fluoro acetoxy) iodobenzene with hydrazone may be cyclized in CHCl_3 or DMSO at room temperature [28].



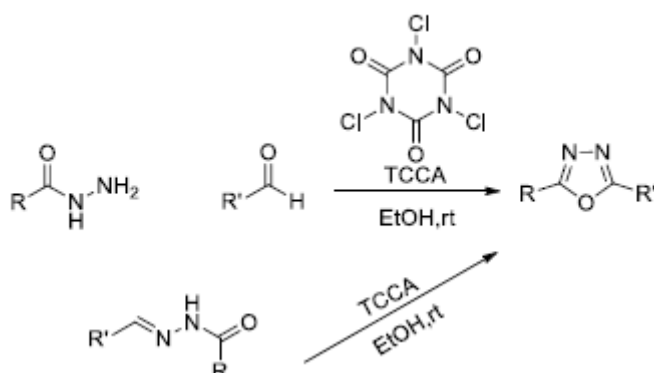
Scheme 6. Synthesis of oxadiazole by oxidised agent

- Solvent-free acyl hydrazone-orthoester condensation in existence of silica and H_2SO_4 [29].



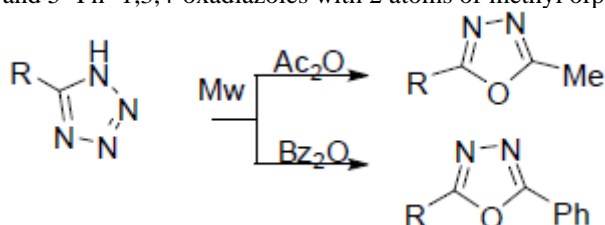
Scheme 7. Synthesis of oxadiazole by silica-sulfuric acid

- Trichloroisocyanuric acid (TCCA) is used to cyclize acylhydrazones and is also utilized to generate disubstituted-1,3,5-oxadiazole at room thermal reading by reacting acid hydrazide with aldehyde [30].



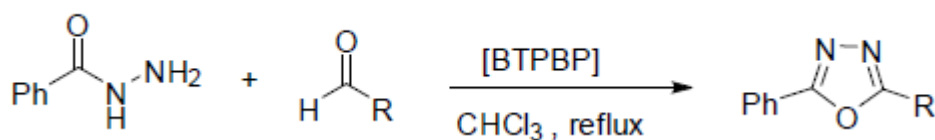
Scheme 8. Synthesis of oxadiazole by TCCA

- Under microwave irradiation, acetic or benzoic anhydride adding acyl group of 5-aryl(hetaryl)tetrazoles yielded the corresponding 5-Me and 5-Ph-1,3,4-oxadiazoles with 2 atoms of methyl or phenyl substituted [31].



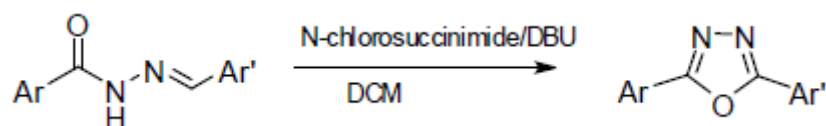
Scheme 9. Synthesis of 2,5-disubstituted-1,3,4-oxa diazoles Under microwave irradiation

- By combining aromatic aldehydes, acylhydrazide, and BTPBP [1,4-bis (triphenyl-phosphonium)-2-butene peroxodisulfate], one may create the oxadiazoles [32][33].



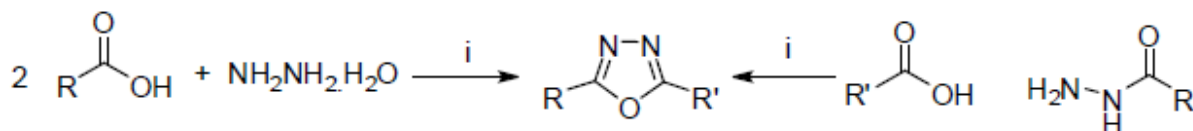
Scheme 10. Synthesis of 1,3,4-oxadiazole by BTPBP

- In order to get the necessary 2,5-disubstituted-1,3,5-oxadiazole, acyl hydrazone is oxidatively cyclized with the help of NCS and DBU as an efficient oxidant [34].



Scheme 11. Synthesis of 1,3,4-oxadiazole by DBU

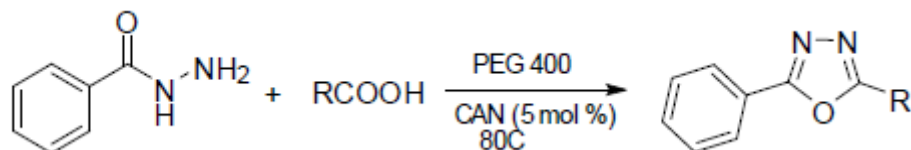
- M. Kidwai & R. Mohanin in 2003 reported synthesis symmetrical and un symmetrical oxadiazole from reaction carboxylic acid and hydrazine or acid hydrazide with carboxylic acid using acidic alumina or monomorillonite K10 clay under microwave irradiation condition as catalyst[35].



R=R' or R≠R', i) acidic alumina or monomorillonite K10 clay, MWI

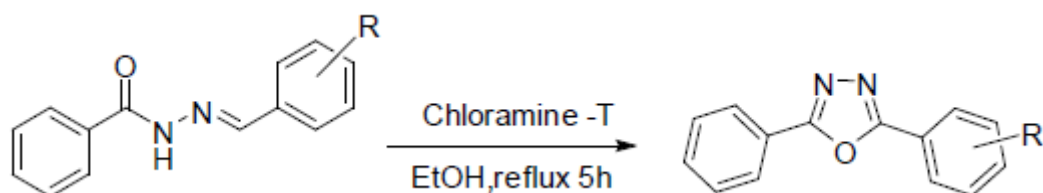
Scheme 12. Synthesis of symmetrical and un symmetrical oxadiazole

- Condensation of acid hydrazide and carboxylic acid is catalysed using ceric ammonium nitrate (CAN) and Polyethyleneglycol (PEG) as catalysts in the synthesis of di-substituted oxadiazoles[36].



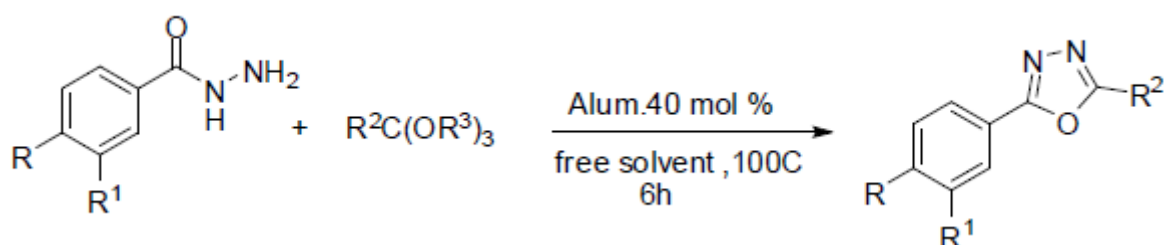
Scheme 13. Synthesis of oxadiazole by CAN & PEG

- Antibacterial study for synthesized oxadiazole had been reported, and this cyclization occurred by using Chloramine-T as a cyclization agent toward the hydrazone [37].



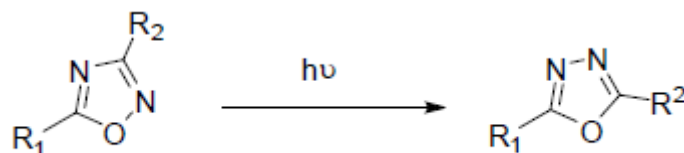
Scheme 14. Synthesis of oxadiazole by Chloramine-T

- Condensation of acyl hydrazides with orthoesters under solvent-free conditions at 100°C catalyzed by g Alum (KAl(SO₄)₂ · 12H₂O)[38].



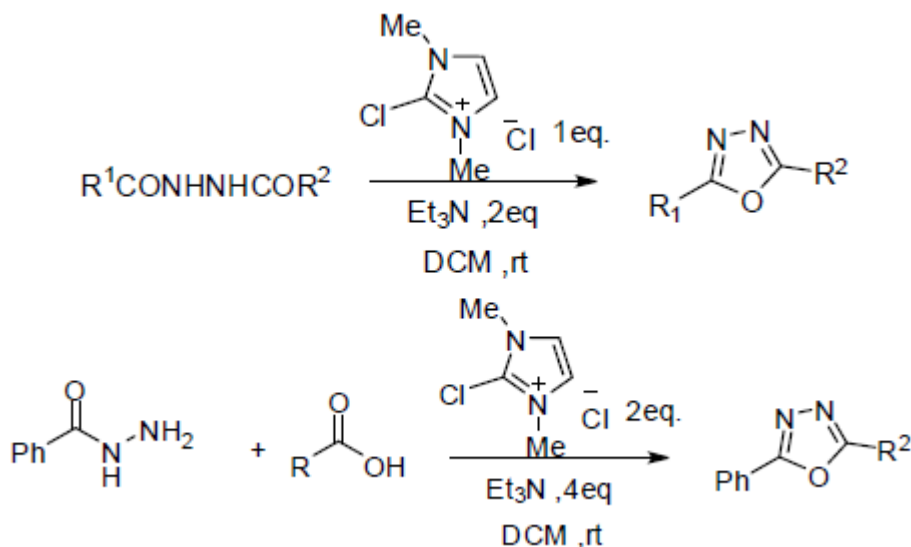
Scheme 15. Synthesis of oxadiazole by Alum

- 1,3,4 oxadiazole was synthesized from Photorearrangement of 1,2,4-Oxadiazoles[39].



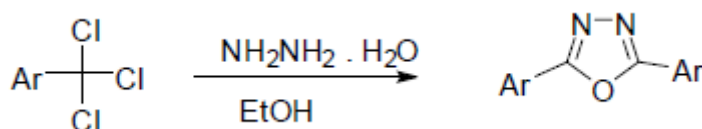
Scheme 16. Synthesis of 1,3,4-oxadiazole by Photorearrangement

- Using an ionic liquid 2-Chloro-1,3-dimethylimidazolium Chloride as catalyst in presence of triethyl amine as catalyst to cyclization of Diacylhydrazines, also with Reaction of Acylhydrazines and Carboxylic Acids .



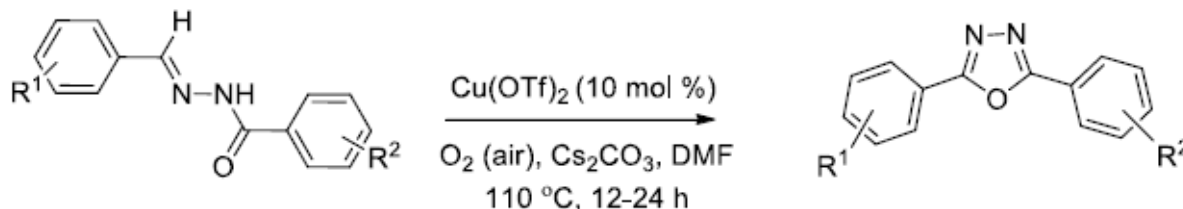
Scheme 17. Synthesis of oxadiazole by ionic liquid

- Symmetrical 2,5-diaryl-1,3,4-oxadiazole and 1,4-phenylene Bis oxadiazole were synthesized from reaction of trichloromethylarenes with excess of hydrazine hydrate in ethanol.



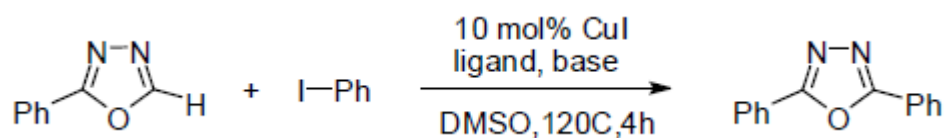
Scheme 18. Synthesis of Symmetrical 1,3,4-oxadiazole

- A catalytic amount of $\text{Cu}(\text{OTf})_2$ was used to amine C-H functionalize N-arylidene arylhydrazide, yielding asymmetric and unsymmetric disubstituted 1,3,4-oxadiazoles [40].



Scheme 19. Synthesis of 1,3,4-oxadiazole by $\text{Cu}(\text{OTf})_2$

- Tsuyoshi Kawano et al [41] used new method for synthesis di substituted oxadiazole from reaction 2-phenyl 1,3,5-oxadiazole with phenyl iodide in presence of ligand and base with copper iodide as a catalyst. In this investigation they used different base such as K_2CO_3 , Li_2CO_3 , Na_2CO_3 , CsCO_3 , and LiO-t-Bu , with various ligand like triphenyl phosphene (PPh_3), 1,10 phenanthroline (phen), 2,2'-bipyridine (bpy), N,N,N',N'-tetramethylethylenediamine (TMEDA) and N,N'- dimethylethylenediamine (DMEDA). The yield percentage demonstrated in table 2-1

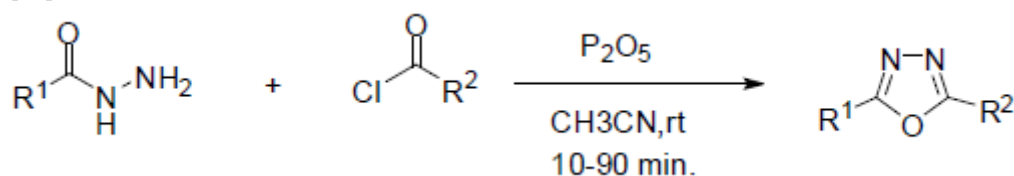


Scheme 20 Synthesis of 1,3,4-oxadiazole by using phenyl iodide

Table 1. -Some Physical Properties of Catalyst.

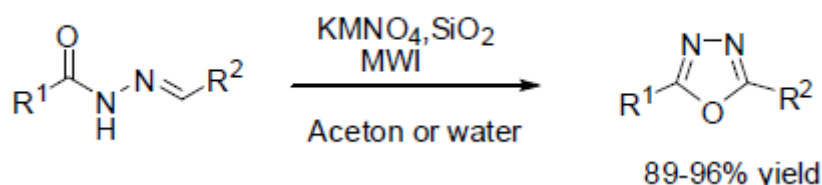
No.	Ligand	Base	Yield %	No.	ligand	Base	Yield%
1	<i>p</i> ph ₃	K ₂ CO ₃	37	6	phen	Li ₂ CO ₃	7
2	TMEDA	K ₂ CO ₃	18	7	phen	Na ₂ CO ₃	31
3	DMEDA	K ₂ CO ₃	73	8	phen	CS ₂ CO ₃	83 at 100 ^o C
4	Bpy	K ₂ CO ₃	33	9	phen	LiO-t-Bu	0
5	Phen	K ₂ CO ₃	91				

- Di substituted oxadiazoles were synthesized via one pot at room temperature, from reaction of different of acid hydrazide and acyl halides in presence of phosphorus pentoxide in acetonitrile, this method shows excellent yield[42].



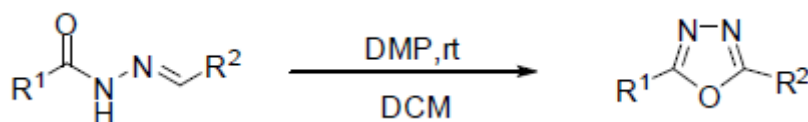
Scheme 21 Synthesis of oxadiazole by P₂O₅

- Treatment of aryl hydrazones with potassium permanganate on the surface of SiO₂ with acetone or water under microwave irradiation; lead to yield between 89%-97% from di substituted oxadiazole.



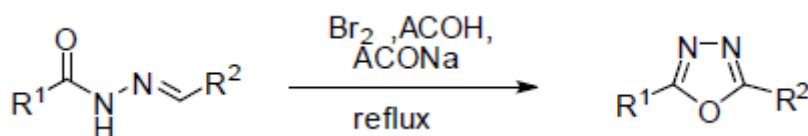
Scheme 22 Synthesis of oxadiazole by using potassium permanganate on the surface of SiO₂

- The oxidative reagent activity of Dess-martin periodinane (DMP) in dicyclohexane (DCM) at room temperature on cyclic N-acylhydrazones [43].



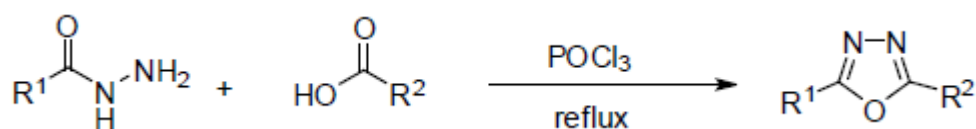
Scheme 23 Synthesis of 1,3,4-oxa diazole by using DMP

- Cyclization of N-Aryl hydrazones using Bromine in glacial acetic acid and sodium acetate to form di-substituted oxadiazole[44].



Scheme 24 Synthesis of oxa diazole by using Bromine

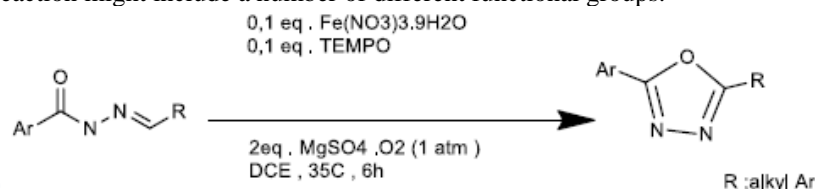
- Cyclization of acid hydrazide and carboxylic acid in the presence of Phosphorus oxychloride (POCl₃) [45][46][47] as a popular dehydration agent to attain 2,5-disubstituted-1,3,4-oxadiazole.



Scheme25 Synthesis of 1,3,4-oxadiazole by using POCl₃

This method was used to achieve our goal; eleven of 2, 6-di-*tert*-butyl-4- (5-Aryl-1,3,4-oxadiazol-2-yl) phenol and seven of 4,4'-(5,5'-(Substitute) bis (1,3,4-oxadiazole-5,2-diyl)) bis (2,6-di-*tert*-butylphenol) were synthesized according to this method.

- Synthesis of 2,5-disubstituted 1,3,4-oxadiazole derivatives by oxidative cyclization of aroyl hydrazones in the presence of oxygen catalyzed by cationic Fe (III)/TEMPO yielded high to excellent percentages in 2017. The reaction might include a number of different functional groups.

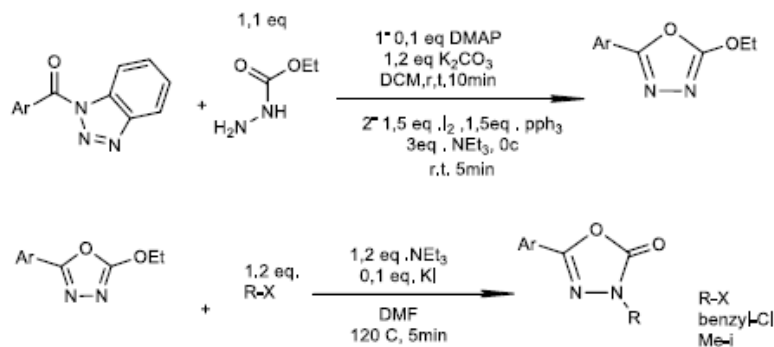


R1 = aryl, thienyl, 2-naphthyl, phenethyl, *t*-Bu, *i*-Bu, *n*-Pr, *n*-pentyl

R2 = H, Me, OMe, OH, CF₃, I, Br, Cl, NO₂

Scheme26 Synthesis of 1,3,4-oxadiazole by using Fe(III)/TEMPO

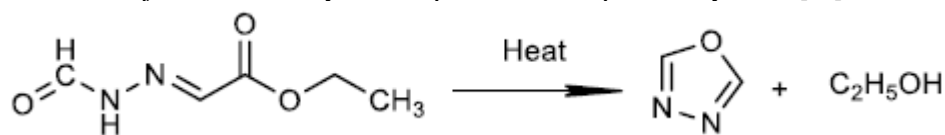
- Using sequential N-acylation/dehydrative cyclization of ethyl carbamate and N-acylbenzotriazoles with Ph₃P-I₂ as a dehydrating agent, it is now possible to quickly and efficiently synthesize 5-substituted-2-ethoxy-1,3,4-oxadiazoles in 2017. After reacting with stoichiometric quantities of alkyl halides, a large number of 3,5-disubstituted 1,3,4-oxadiazol-2(3H)-ones is produced. high above.



Scheme27 Synthesis of oxadiazole by using N-acylation

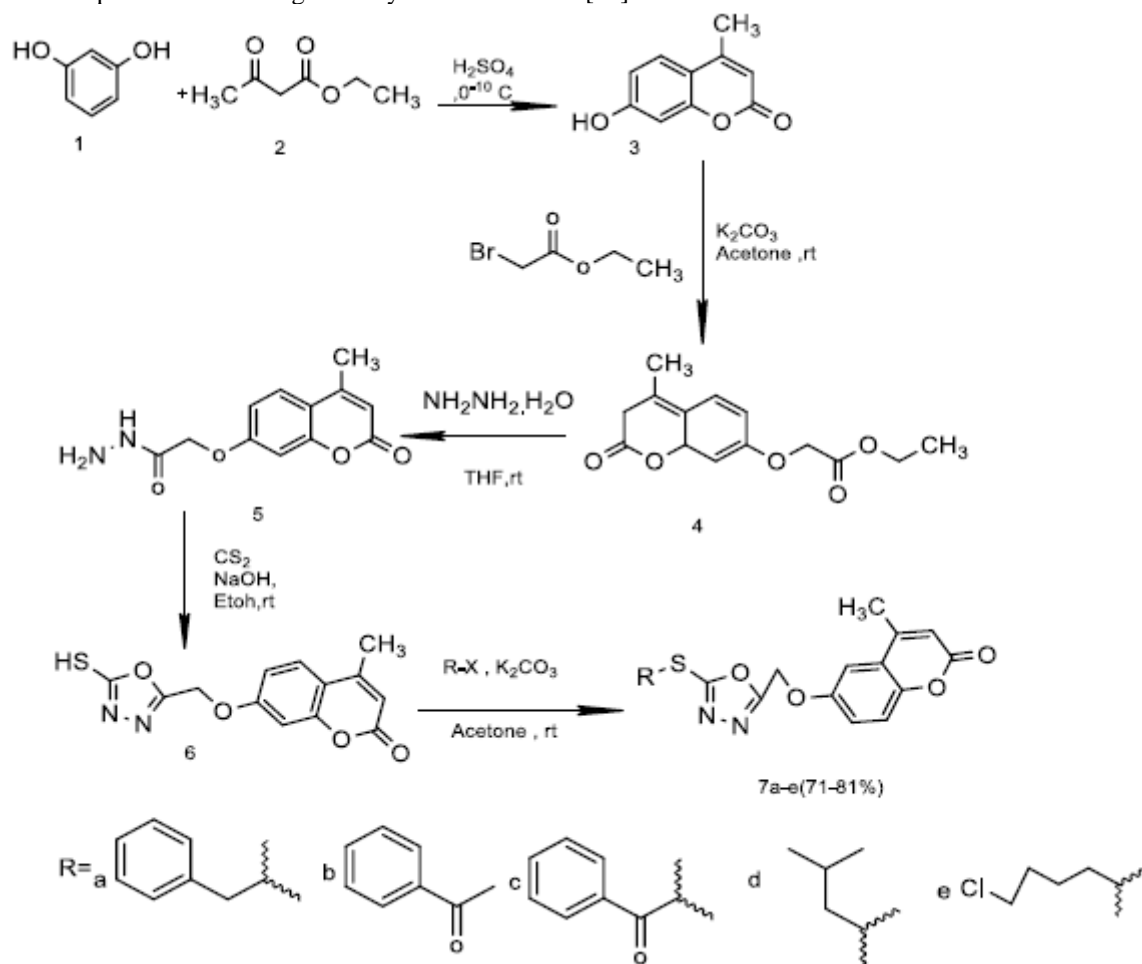
product	yield (% , isol.)	product	yield (% , isol.)
	86		75
	63		74
	80		86

- In 2022, Ainsworth was the first to detail the process for making unsubstituted 1,3,4-oxadiazole. Formylhydrazone ethylformate was subjected to thermolysis at air pressure to complete the synthesis[48].



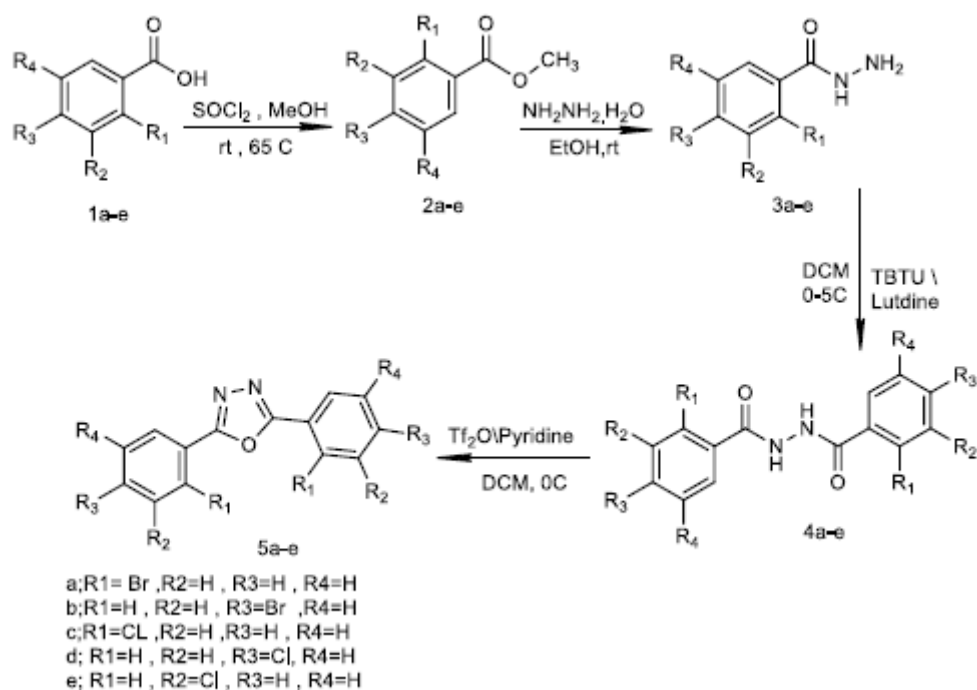
Scheme 28 Synthesis of unsubstituted oxadiazole

- To create 2-aryl-1,3,4-oxadiazole with a p-bromophenylaminomethyl group substituted at position 5, Bhat et al. developed a method using mercury oxide and iodine[49].



Scheme 29 Synthesis of oxadiazole by mercury oxide and iodine

- According to the strategy proposed by Zabiulla et al., 2,5-bisphenyl-1,3,4-oxadiazoles may be synthesised by the cyclization of N,N'-diacylhydrazines using tri fluoromethanesulfonic anhydride as the main reagent[49].



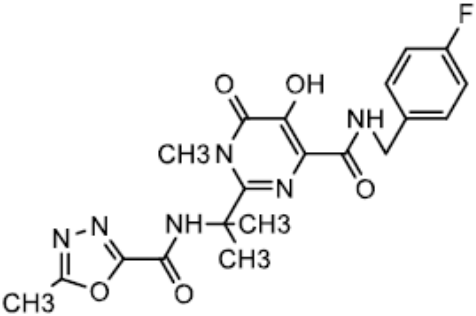
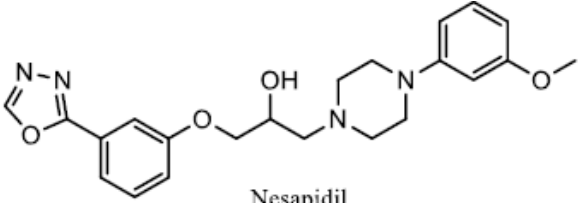
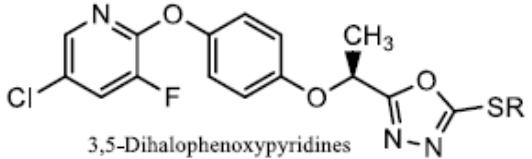
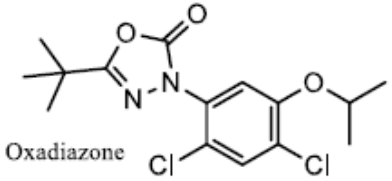
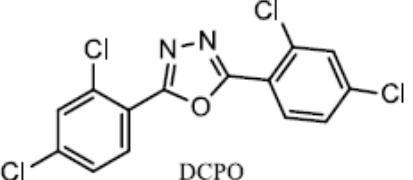
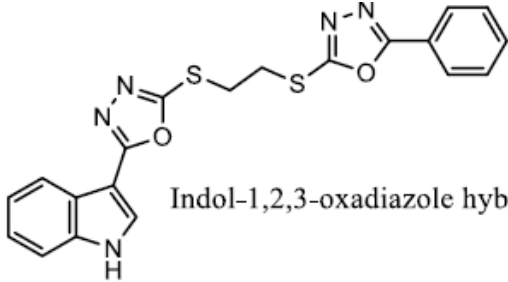
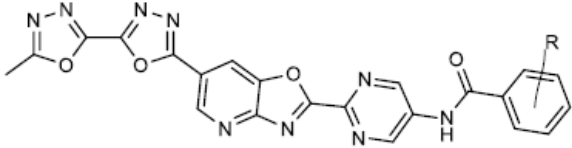
Scheme 30 Synthesis of oxadiazole by using trifluoromethanesulfonic anhydride

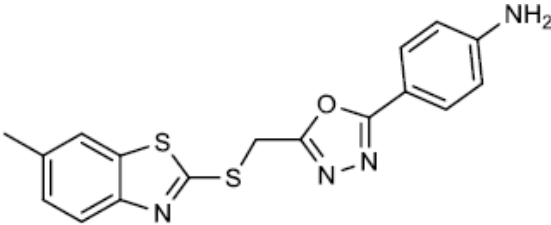
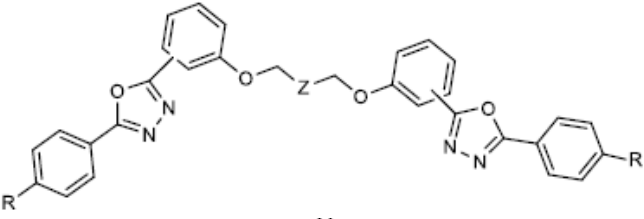
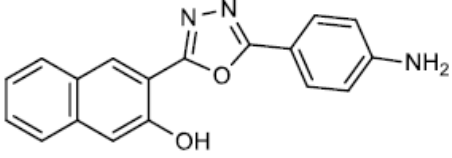
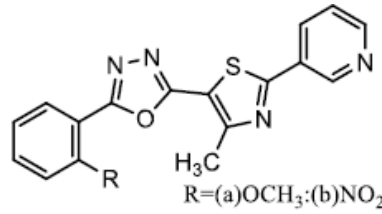
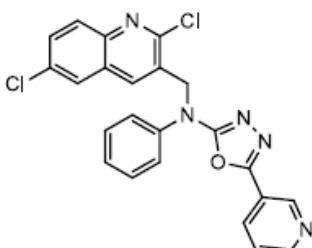
1.2 TABLES

Table 2 showed a wide range of biological activity for many oxadiazole derivatives due to the presence of the oxadiazole nucleus, which is involved in wide fields such as the pharmaceutical industry, dyes, materials science, and most organic compounds, in addition to the discovery of various biological agents such as anti-cancers, antihypertensive drugs, and in soothing sinuses, as well as in the physical field such as photoluminescence and others.

Table 2. - Some of Oxadiazole derivative with bioactive

No.	Structure	Biological activity	References
1	<p>Tiodazosin</p>	Antihypertensive	[50]
2	<p>Furamizole</p>	Antibacterial and anti-influenza	[51]

3	 <p>Raltegravir(MK-0518)</p>	Antiretroviral and HIV	[52]
4	 <p>Nesapidil</p>	Vasodilator	[53]
5	 <p>3,5-Dihalophenoxy pyridines</p>	Herbicide	[49]
6	 <p>Oxadiazone</p>	Herbicide	[54]
7	 <p>DCPO</p>	DNA and protein synthesis in insects are stymied.	[55]
8	 <p>Indol-1,2,3-oxadiazole hybrid</p>	Effective against <i>Xanthomonas oryzae</i> and <i>Ralstonia solanacearum</i>	[56]
9	 <p>1a-j</p> <p>a;R=3,4,5-trimethoxy, b;R=3,5-dimethoxy, c;R=4-methoxy, d;R=4-nitro, e;R=3,5-dinitro, f;R=4-chloro, g;R=4-bromo, h;R=4-methyl, i;R=4-(dimethyl amino), j;R=4-cyano</p>	Anticancer agent	[57][58]

10		Antidiabetic activity	[59]
11		antibacterial activity	[60][61]
12		antifungal activity	[62]
13		ant tubercular	[63]
15		Antimalarial	[64]

2.1 RESULTS AND DISCUSSION

The 1,3,4-oxadiazole derivatives with five atoms of ring structure are the building blocks of several molecules with beneficial biological activity [27] in medicine and agriculture. Evidence from studies evaluating the potency of these biological activities suggests that many systems of this kind have therapeutic potential against a wide range of infections. This has increased the tendency of researchers to study many of these compounds, diagnose them spectroscopically and physically, and search for new ways to prepare them and find raw materials. With the most reasonable prices, suitable environmental conditions, and available auxiliary factors.

Study data were collected to evaluate the effect of the prepared derivatives on different types of drugs, and studies showed their inhibitory effectiveness compared to pharmaceutical compounds on the nervous system as sedatives and anti-inflammatory substances and in many industrial applications, dyes, fluorinated materials, and heat resistance.

2.2 Conclusion

The urgent need to find new antifungal, antibacterial, anti-inflammatory in nature and analgesic molecules has prompted several research teams to investigate novel synthesis techniques of these substances. Research on 1,3,4-oxadiazole medications also focuses heavily on enhancing the structure and functionality of existing pharmaceuticals and insecticides. A molecule's solubility may affect its bioavailability and efficacy, and this can be achieved by the introduction of novel substituents or the development of new hybrid materials with existing physiologically active compounds. Oxadiazole-based chemical designs are becoming more popular, and as a result, there is a growing library of physiologically active compounds that might be used in the future for medicines for humans and animals, as well as for the prevention of plant diseases.

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