

Effective ZnO Doping by pulsed laser deposition to Improve the efficiency of the solar cells of TiO₂ at 200 C⁰

تأثير تطعيم اوكسيد الخارصين بواسطة ترسيب الليزر النبضي لتحسين كفاءة الخلية الشمسية للأوكسيد التيتانيوم عند درجة حرارة C⁰ (200)

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Abstract

In this research was prepared TiO₂: ZnO have been deposited on the glass substrates by the pulsed laser technique, the use of Nd:YAG laser with wavelength (1064)nm and energy 600mJ and have been deposited with different concentration (0.03, 0.05, 0.07, 0.1)% of ZnO at annealing temperature 200 C⁰ with thickness (400 nm). The TiO₂ thin films showed n-type conductivity, according to Hall measurements, the carrier concentration of TiO₂ thin films had increased from $(18.5 \times 10^{10}$ to $45.16 \times 10^{10} \text{ cm}^{-3}$). The average absorbance value of TiO₂ thin films was improved from $(3.412 \times 10^{-6}$ to 8.654×10^{-6}) by the doping of ZnO. These improved properties should contribute to the efficiency of the photovoltaic effect of the photogenerated charged carriers. The (I-V) characteristic of (Al / TiO₂: ZnO / n-Si / Al) heterojunction show that the solar cell which is produced at X=0.1% by efficiency (η) = 8.769

Keywords: Titanium dioxide (TiO₂) Zinc oxide (ZnO), Silicon wafer, Pulsed laser deposition

الخلاصة

تم تحضير اغشية اوكسيد التيتانيوم على قواعد من الزجاج باستخدام تقنية الترسيب بالليزر النبضي حيث تم استخدام ليزر النديميوم- ياك النبضي ذو الطول الموجي (1064nm) وبطاقة (600 mj) ومن ثم تم ترسيب هذه الاغشية بأوكسيد الخارصين بتراكيز مختلفة (0.03 , 0.05 , 0.07 , 0.1) ودرجة حرارة (200 C⁰) وبسمك (400nm). وبينت قياسات هول ان الاغشية المحضرة تكون ذات توصيلية سالبة حيث نلاحظ كلما زاد نسبة التراكيز تزداد تراكيز الحاملات من (18.5×10^{10}) الى $(45.16 \times 10^{10} \text{ cm}^{-3})$ وكذلك تزداد التوصيلية من $(3.412 \times 10^{-6} \text{ } \Omega \cdot \text{cm})$ الى (8.654×10^{-6}) وهذه الخصائص ساهمت في تحسين كفاءة الخلية الشمسية بينت خواص تيار- فولتية للمفروق الهجين (Al / TiO₂: ZnO / n-Si / Al) الحصول على خلية شمسية حيث عند التركيز 0.1% ذات كفاءة عالية 8.769

Introduction

The titanium dioxide (TiO₂) is important material for a many device such as solar cell , optical detector and sensor. The Physical and chemical properties of TiO₂ (direct optical band gap 3.06eV, n-type , structure is tetragonal, with two TiO₂ recipe units (six atoms) per primitive cell color is white) [1,2]. Lattice parameters are: a, b equal 3.7710 Å and c equal 9.430 Å with c/a proportion of 2.5134 [1] as appeared in Fig. (1), density 4.27(g/cm³) , melting point 1855 (°C), boiling point 2972(°C) and refractive index 2.609. As a champion among the most important direct conductive oxide , TiO₂ and its mixes have been for the most part utilized as a piece of photovoltaic devices, sun oriented cells, clear –gas sensors [3]. Wide range of methods utilized for depositing titanium dioxide films including laser removal, compound vapor deposition (CVD) [4] and template-directed growth [5]. Each of these techniques are hard to dissipate the material titanium dioxide since it has a high dissolving point, the specialists could evaporate materials by the laser pulse [6].

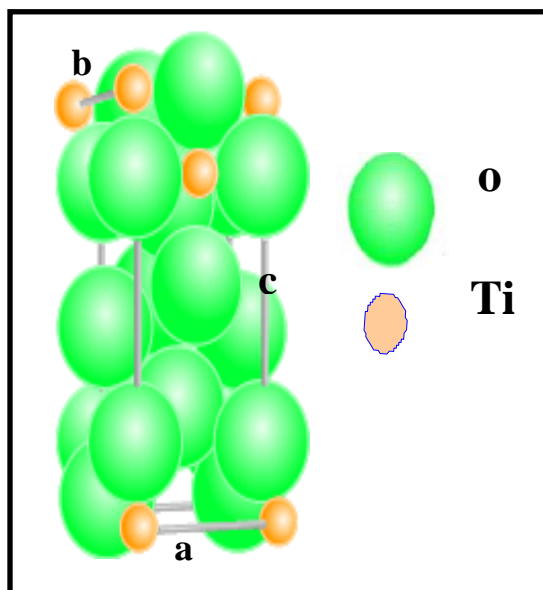


Fig. (1): Ti metastable stage for crystalline TiO₂ [1].

Experimental details

(TiO₂) and (ZnO) carried with purity (99.9%) Used to prepare TiO₂ with different doping of ZnO (0.03, 0.05, 0.07and 0.1) wt.%. The mixed powder has blended mechanically so that the mix is reliably scattered. The resultant powder is ground again and was squeezed under 5 tons press to shape a target with 2.5 cm width and 0.4 cm thickness. The got target was as thick and homogenous as possible to ensure a better than average nature of the store. The goal earlier and after that a while later evacuation is appeared in Fig. (2)

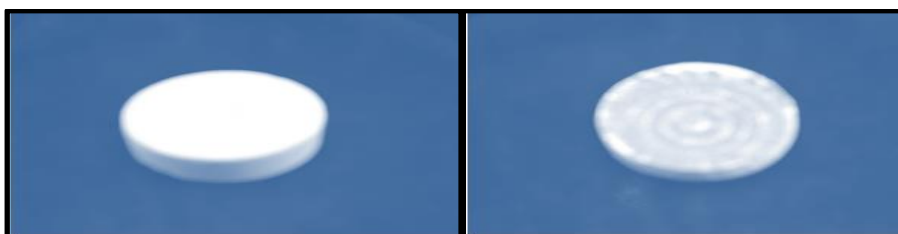


Fig. (2): The objective (a) preceding and (b) in the wake of being removed by laser.

I-V characteristic were has been done for n-TiO₂:ZnO/p-Si heterojunction by using Halogen light source (Philips (120 W)) with forces (100) mW/cm². Keithley Digital Electrometer 616 utilized as voltmeter and ammeter. The electrical circuit for I-V estimation as shown in Fig.(3)

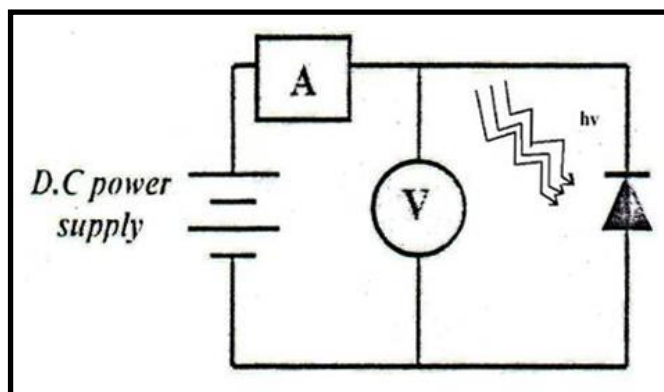


Fig. (3): Circuit chart for I-V estimation of heterojunction under light [7].

Results and Discussion

Some electrical properties of TiO₂ thin films were measured for doping of ZnO into TiO₂ thin films by Nd-Yag laser. The TiO₂ thin films in this experiment showed n-type conductivity, according to Hall measurements, after annealing at 200 °C for 2 hour as shown in table (1). The carrier concentration and absorbance of TiO₂ thin films significantly increased with increasing of concentration ZnO , as well as the decreased resistivity, would contribute to the efficiency of the photovoltaic effects, including generation, transport to the built-in field region, and separation of the photogenerated charged carriers[8]

Table (1): The result of electrical measurement for ZnO -doping TiO₂ at different concentration

ZnO-doped TiO ₂ (wt.%)	R _H (m ² /C)*10 ⁹	Carrier type	n(cm) ⁻³ *10 ¹⁰	σ (1/Ω.cm)*10 ⁻⁶	μ _H (cm ² /V.s) *10 ³
0.03	3.670	n	18.5	3.412	3.245
0.05	3.450	n	22.89	4.654	3.124
0.07	3.012	n	32.21	5.789	2.654
0.1	2.215	n	45.16	8.654	2.012

The V_{oc} is the maximum voltage that can be gotten from a solar cell ; when the current is zero, The equation for V_{oc} is given by [9]:

$$V_{oc} = \frac{kT}{q} \ln \left(\frac{I_{Light}}{I_0} + 1 \right) \dots \dots \dots (1)$$

Where

(k ,I_o ,q ,T) are Boltzmann's consistent , dark saturation current. electron charge and total temperature

The light produced current short circuit current can be approximated by [9]:

$$J_{sc} = qG(L_n + L_p) \dots \dots \dots (2)$$

Where

(G, q, L_n, L_p) are the generation rate , electron charge, electron diffusion length, hole diffusion length

The fillfactor (This relation is given by [8]:

$$FF = \frac{P_m}{V_{oc}J_{sc}} = \frac{V_m J_m}{V_{oc} J_{sc}} [\%] \dots \dots \dots (3)$$

Where: (V_m and J_m) are the voltage and the current density corresponding to P_m

The conversion efficiency (η) calculate equation (4),[10].

$$efficiency = \eta = \frac{P_m}{P_s} = \frac{V_{oc} J_{sc} FF}{P_s} [\%] \dots \dots \dots (4)$$

Where: (P_m, P_s):are the maximum power generated by the cell ,power of the incident light

Figures.((4) to (9)) show that the photocurrent increments with expanding the predisposition voltage, additionally it can be seen that the photocurrent in the invert inclination is bigger than that in the forward inclination. This can be ascribed to the way that the width of the exhaustion locale increments with the expansion of the connected turn around predisposition voltage which prompts to the partition of the electron–hole sets [11] .We can observe that efficiency increases with increasing of ZnO -doping , from 1.077 to 8.769, we can notice that from Table(2)

since the expanding of doping causes a reworking of the interface iotas and diminish the dangling bond which prompts to change of the junction qualities [12].

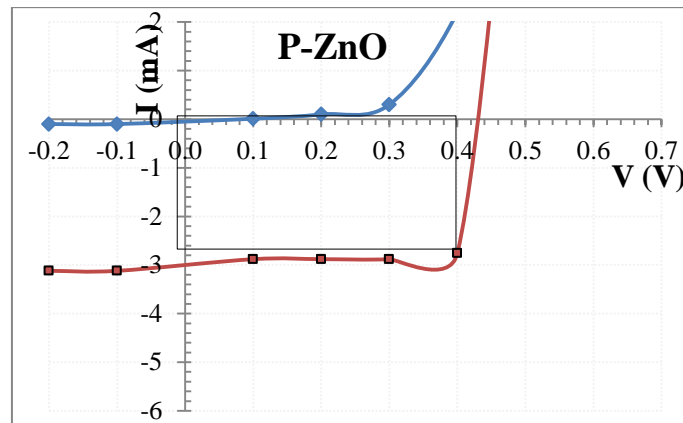


Fig.(4) I-V characteristics under dark and (100) mW/cm²illumination for P-ZnO.

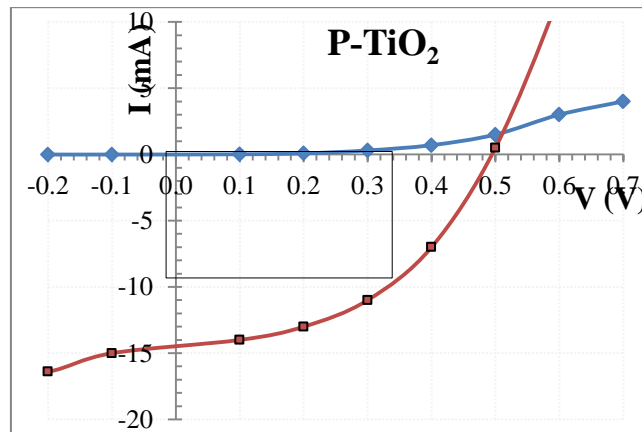


Fig.(5) I-V characteristics under dark and (100) mW/cm²illumination for P-TiO₂.

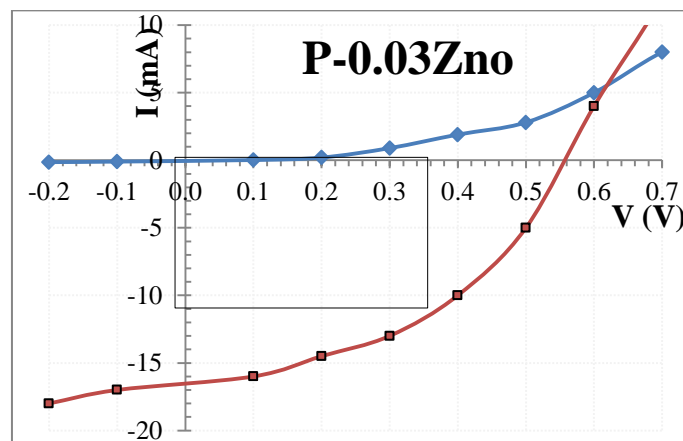


Fig.(6) I-V characteristics under dark and (100) mW/cm² illumination for P- 0.03% ZnO-doped .

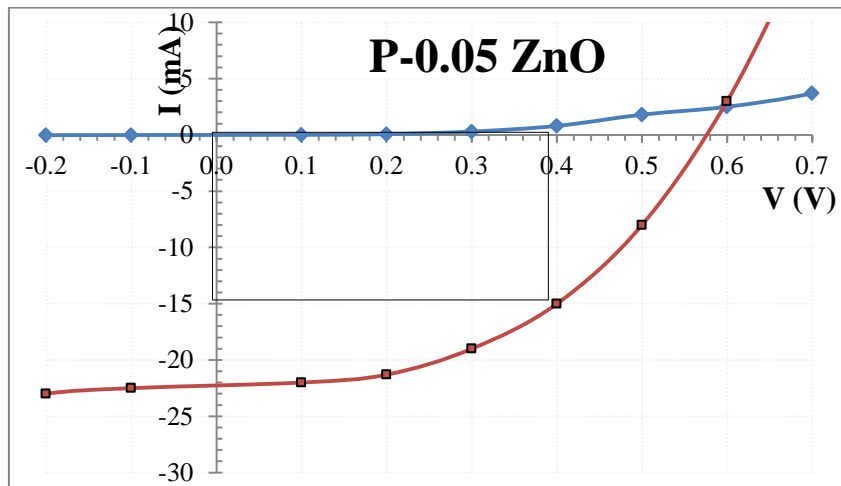


Fig.(7) I-V characteristics under dark and (100) mW/cm² illumination for P- 0.05% ZnO-doped

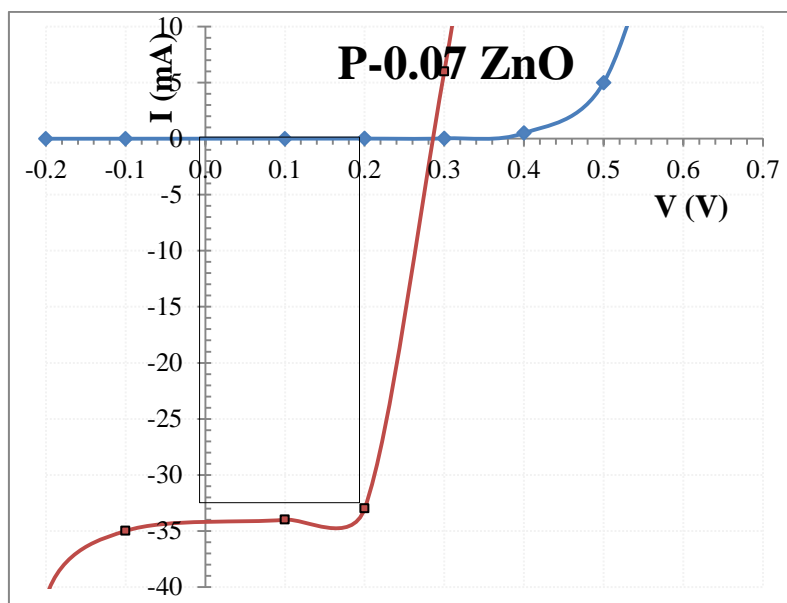


Fig.(8) I-V characteristics under dark and (100) mW/cm² illumination for P- 0.07% ZnO-doped .

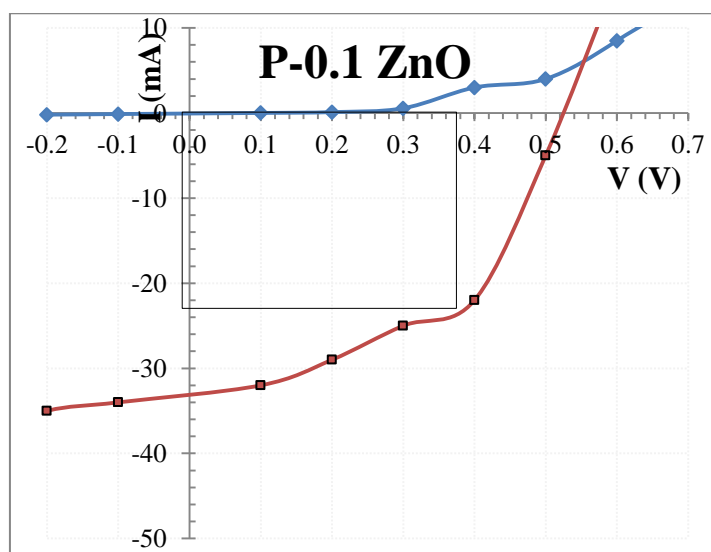


Fig.(9) I-V characteristics under dark and (100) mW/cm² illumination for P- 0.1% ZnO-doped .

Table(2)shows the efficiency values obtained for the various concentration.

Ratio of Composition (X)%	V _{oc} (Volt)	I _{sc} (mA/cm ²)	V _{max} (Volt)	I _{smax} (mA/cm ²)	FF %	η %
ZnO	0.430	3.000	0.400	2.800	0.868	1.077
TiO ₂	0.500	14.500	0.340	10.000	0.469	3.269
0.03	0.560	16.500	0.360	11.000	0.429	3.808
0.05	0.580	22.000	0.400	15.000	0.470	5.769
0.07	0.280	34.000	0.200	33.000	0.693	6.346
0.1	0.520	33.000	0.380	24.000	0.531	8.769

Conclusion

From I-V can conclude that the best efficiency of the solar cell is obtained when the concentration of the (0.1%) obtained at (8.769) . That's where the concentration helps to improve the structure of the solar cell and also helps to increase the current density and the ones we get the best efficiency. The results obtained from Hall effect of ZnO doping TiO₂ substrate at annealing temperature (200C⁰) Oxygen pressure 10⁻³ were (n-type) .The results show that the increase in doping concentration lead to increase in the value of electrical conductivity and charge carriers but the mobility and Hall coefficient have been decreased. Hall coefficient show that the type of TiO₂ with increase in ZnO concentration. That mean the electrons charge carriers are dominant in increased the electrical conductivity.

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