

## **Synthesis, characterization of a new series of thiohydantoin derivatives and study their biological activity against prostate cancer and microbes.**

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### **ABSTRACT**

Prostate cancer is one of the most common malignant diseases in men, and the design of new chemical compounds against prostate cancer is one of the challenges faced by modern medicine. In this study, we prepared and characterized a new series of thiohydantoin derivatives by reacting maleimide derivatives (1C-4) with cyclohexyl isothiocyanate. The prepared compounds were diagnosed using (FT-IR ,<sup>1</sup>H-NMR ,<sup>13</sup>C-NMR ). MTT assay was used to determine the activity of the prepared compounds against prostate cancer cell lines, as the data indicated, based on IC<sub>50</sub> values, that the prepared compounds do not have anti-prostate cancer activity. The prepared compounds were also studied against selected types of bacteria and fungi, as the data indicated that some of the prepared series compounds had anti-bacterial and anti-fungal activity under study, as the compound 4C showed a high anti-*staph* activity.

**Key words:** Thiohydantoin, antibacterial activity, antifungal activity, prostate cancer.

### **الخلاصة :**

يعد سرطان البروستات من أكثر الأمراض الخبيثة شيوعاً لدى الرجال كما ان تصميم مركبات كيميائية جديدة مضادة لسرطان البروستات أحدى التحديات التي يواجهها الطب الحديث . في هذه الدراسة قمنا بتحضير وتشخيص سلسلة جديدة من مشتقات الثيوهيدانتوين وذلك من خلال تفاعل مشتقات الماليمايد (1C-4) مع cyclohexyl isothiocyanate .

تم تشخيص المركبات المحضرة باستخدام (FT-IR ,<sup>1</sup>H-NMR ,<sup>13</sup>C-NMR ) .

تم استخدام فحص MTT لتحديد نشاط المركبات المحضرة ضد خطوط خلايا سرطان البروستات حيث أشارت البيانات وبالاعتماد على قيم IC<sub>50</sub> أن المركبات المحضرة ليس لها فعالية مضادة لخطوط خلايا سرطان البروستات . كذلك تم دراسة المركبات المحضرة ضد أنواع مختارة من البكتيريا والفطريات حيث تشير البيانات أن بعض مركبات السلسلة المحضرة فعالية مضادة للبكتيريا والفطريات قيد الدراسة حيث أظهر المركب 4C فعالية عالية مضادة لبكتيريا Staph .

## **INTRODUCTION**

Hydantoin was first isolated by Bayer in 1861 in the course of her research on uric acid [1]. The data indicate that the nucleus of thiohydantoin is one of the important tributaries of the medical and pharmaceutical aspects, and the first compound that was prepared for this type of compound was by Peter Klason in 1890[2]. 2-thiohydantoin is a biologically active molecule with diverse activities to support the medical and pharmaceutical aspects of facing the challenges faced by the scientific community in combating drug-resistant diseases that are spreading alarmingly where it has been used as an anti-cancer [3-8],anti-mutagenic [9], an antiviral [10], anti-tuberculosis [11], antibacterial [12] [13][14]also showed anti-HIV activity [15] and anti-Alzheimer's disease [16]. One of the most important things that drew the attention of researchers to the thiohydantoin nucleus recently is that it is the nucleus of the parent compound in the anti-prostate cancer drug (enzalutamide, apalutamide ) [17]. Recently, some prostate cancer patients have acquired resistance to the drugs prescribed for the treatment of prostate cancer patients (enzalutamide, apalutamide), and the reason is attributed to the emergence of the F876L mutation[18]. Data show that many major corporations have abandoned their support for the discovery of new antibiotics over the last two decades, owing to the economic challenges that come with this aspect[19] [20]. In this study, we prepared a new series of thiohydantoin derivatives and was studied against prostate cancer cell lines. The prepared compounds were also studied against specific types of bacteria and fungi, as the data indicated that some of the prepared compounds had anti-bacterial and anti-fungal activity under study.

## **Experimental**

### **Chemistry**

Infrared spectra of the prepared compounds were recorded using a Japanese-made Shimadzu 8400 FT.IR device and in the form of potassium bromide tablets in the (400–4000)  $\text{cm}^{-1}$  region at room temperature in the laboratories of the Department of Chemistry - College of Education for Pure Sciences - University of Basra.  $^1\text{H}$  and  $^{13}\text{C}$ -NMR spectra were recorded using a BRUCKER-500MHz and 125MHz instrument, DMSO- d6 as a solvent , and TMS as an internal reference. All chemical displacements were measured in ppm.  $^1\text{H}$  and  $^{13}\text{C}$ -NMR spectra were recorded at University of Tehran-Iran

### **Synthesis of compounds (1C-4) [21] [22]**

0.01mol of maleimide (1A-4) with 0.01mol of Isoniazid(isonicotinohydrazide) (B) was dissolved in 30ml of absolute ethanol in a round flask equipped with a magnetic stirrer and leaving the mixture under reflux

for (6-32)hrs . The reaction was monitored by TLC( thin layer chromatography). After the reaction is completed, the filtrate is separated and recrystallized or washed in absolute ethanol.

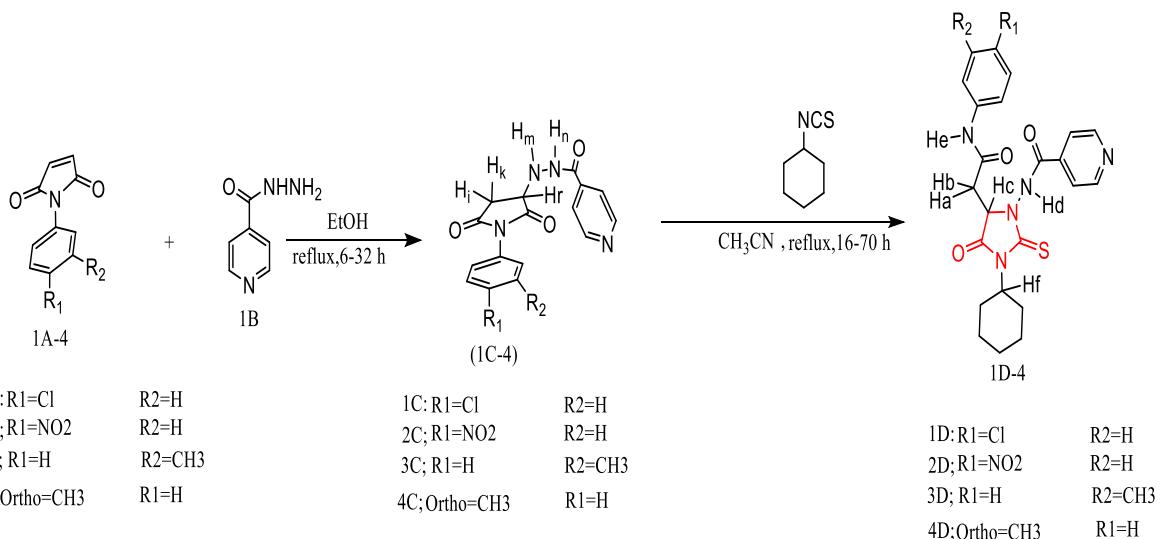
### **Synthesis of compounds (1D-4) [21] [22]**

0.02 mol of maleimide derivatives (1C-4) were reacted with 0.02 mol of cyclohexyl isothiocyanate in the presence few drops of glacial acetic acid as a catalyst in 30 ml of acetonitrile in a round flask equipped with a magnetic stirrer. The mixture was heated by reflux for 18-70 hrs. The reaction was monitored by TLC( thin layer chromatography). After the reaction is completed, the solvent is evaporated and then recrystallized by a mixture of acetone and hexane.

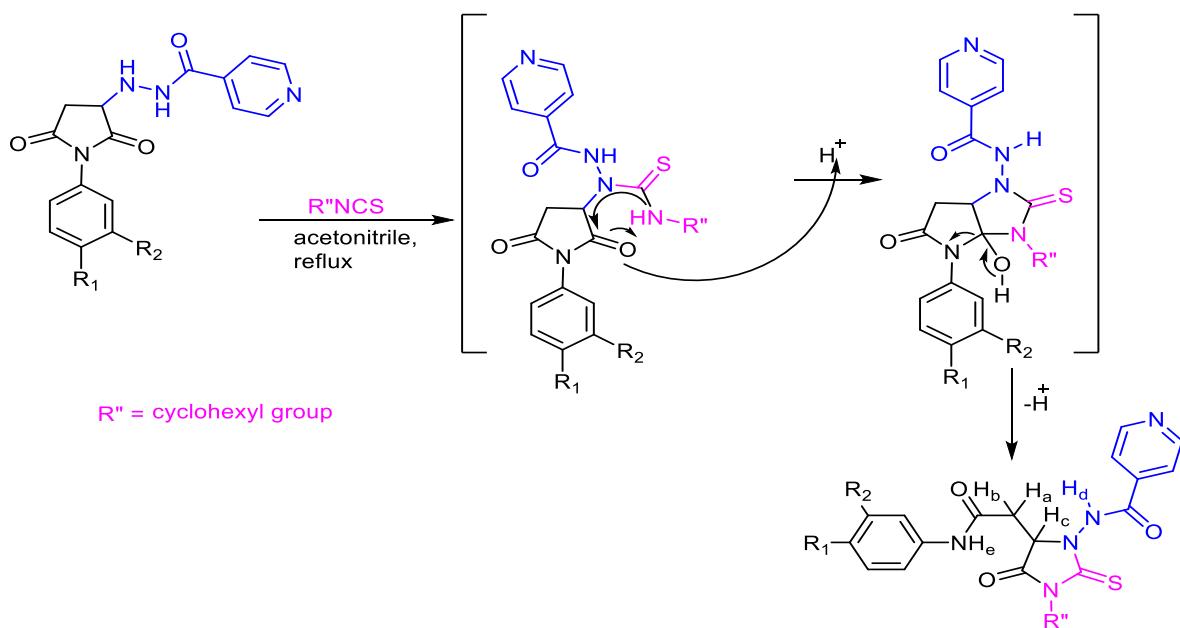
## **RESULTS AND DISCUSSION**

In this study, we prepared a new series of thiohydantoin derivatives, where the reaction included two steps. In the first step, the maleimide derivatives (1C-4) were prepared by reacting different maleimides with Isonazid (isonicotinohydrazide) (B) in absolute ethanol, and the time required for the reaction to occur was (6-32) hrs. The second step included the preparation of thiohydantoin derivatives (1D-4) through the reaction of maleamide derivatives (1C-4) with cyclohexyl isothiocyanate in acetonitrile as a solvent and in the presence of drops of glacial acetic acid as a catalyst,Scheme 1. The prepared series compounds were diagnosed by (FT-IR,  $^1\text{H}$ NMR,  $^{13}\text{C}$ -NMR). The obtained spectra data indicate that the structures of the assumed compounds are correct. The most important bands shown by the FT-IR spectrum, wide band at 3300-3200 belong to the (NH group) It is also noticed that a wide band appears at 3479-3300 belongs to the (NH amide group). The weak band at 3186-3000 is due to (CH arom.). Another band appears at 2951-2900 due to (CH-aliph). The strong and distinct sharp band at 1766-1650 is attributed to the (C=O) group, while the group of bands that appear at 1615-1450 belongs to (C=C arom). . Also, the recorded spectrum of the prepared compounds showed two bands, one at 1449-1400 attributed to the (C = S) group, and the other at 1350-1300 belonged to the (C-N) group. Among the most important signals that were shown by the recorded  $^1\text{H}$ NMR spectrum of maleimide derivatives (1C-4). doublet of doublet signal at 3.0 ppm with a coupling constant  $J = (18.0, 4.1 \text{ Hz})$  attributed to Hi. doublet of doublet signal at 2.8 ppm with a coupling constant  $J = (18.0, 4.0 \text{ Hz})$  is attributed to Hk .Also the recorded  $^1\text{H}$ NMR spectrum of thiohydantoin derivatives (1D-4) showed a triplet signal at 4.63ppm with a coupling constant  $J = 4.3 \text{ Hz}$  due to Hc, multiple signal at 4.63ppm

attributed to Hf, multiple signal at (3-2.7)ppm attributed to HaHb. The recorded  $^{13}\text{C}$ -NMR spectrum of maleimide derivatives (1C-4) and thiohydantoin (1D-4) derivatives showed signs consistent with the structures of the putative compounds. The most important signals showed by the  $^{13}\text{C}$ -NMR spectrum, a signal at 174 ppm is attributed to the carbonyl group in (imidazolidine-2,4-dione), a signal at 164 ppm is attributed to a carbonyl group in (Isonicotinohydrazide), a signal at 171 ppm belongs to a carbonyl group in (thioimidazole), a signal at 174 ppm attributed to the thiocarbonyl group (C=S). A signal at 164 ppm attributed to the carbonyl group appears in (NNHC=O) 'signal at 166 ppm is attributed to the carbonyl group in (ArNHC=O). Figures 1 to 14 shows the spectra that were recorded for the prepared compounds.



**Scheme (1)** shows the general method for preparing the prepared compounds



**Scheme 2 shows the Mechanism of preparing thiohydantoins (1D-4).**

**Structure analysis data of the prepared compounds;**

**N'-(1-(4-chlorophenyl)-2,5-dioxopyrrolidin-3-yl)isonicotinohydrazide(1C):**

White solid powder; yield 89%; mp= 218-222 °C; IR (KBr, cm<sup>-1</sup>) 3408 (NH amide), 3317(NH), 3061 (CH arom), 2993,2937 (CH-aliph), 1784 (C=O imidazolidine-2,4-dione ), 1710 (C=O imidazolidine-2,4-dione), 1639(C=O Isonicotinohydrazide), 1546, 1498 (C=C arom), 1406 ( C-N).<sup>1</sup>H NMR (500 MHz, DMSO) δ 10.46 (s, 1H,H<sub>n</sub>), 8.74(d , J = 5.0,2H,H-Ar), 7.73(d , J = 5.0 Hz ,2H,H-Ar), 7.58 (d, J = 8.6 Hz , 2H,H-Ar), 7.33 (d, J = 8.5 Hz, 2H,H-Ar) ,6.14 (s, 1H,H<sub>m</sub>), 4.31-4.21 (m, 1H,H<sub>r</sub>), 3.09 (dd, J = 17.9, 8.6Hz, 1H,H<sub>i</sub>), 2.80 (dd, J = 18.0, 3.9 Hz, 1H,H<sub>k</sub>), <sup>13</sup>C NMR (126 MHz, DMSO) δ175 (C=O imidazolidine-2,4-dione), 174 (C=O imidazolidine-2,4-dione), 164.39(C=O Isonicotinohydrazide), 150, 139, 132, 131, 128.95, 128.63,121 (C-Ar), 57(CH<sub>r</sub>), 34(CH<sub>i</sub>H<sub>k</sub>).

**N'-(1-(4-nitrophenyl)-2,5-dioxopyrrolidin-3-yl)isonicotinohydrazide(2C):**

White solid powder; yield 49%; mp= 203-205 °C ; IR (KBr, cm-1 ) 3415 (NH amide), 3304(NH), 1788(C=O imidazolidine-2,4-dione), 1720 (C=O imidazolidine-2,4-dione), 1641, 1527,1469, (C=C arom), 1350 ( C-N).<sup>1</sup>H NMR (500 MHz, DMSO) δ 10.53 (d, J = 6.1 Hz , 1H,H<sub>n</sub>), 8.74(d , J = 5.1,2H,H-Ar), 8.38(dd , J = 7.6 ,4.5Hz ,2H,H-Ar), 7.75 (d, J = 5.0 Hz , 2H,H-Ar), 7.64 (d, J = 8.5 Hz, 2H,H-Ar) ,6.24 (d, J = 6.0 Hz, 1H,H<sub>m</sub>), 4.32 (dt, J = 5.9, 2.9 Hz, 1H,H<sub>r</sub>), 3.16 (ddd, J = 17.9, 8.6,1.9 Hz, 1H,H<sub>i</sub>), 2.80 (dt, J = 18.0, 3.0 Hz, 1H,H<sub>k</sub>) <sup>13</sup>C NMR (126 MHz, DMSO) δ 174.75(C=O imidazolidine-2,4-dione), 174.24(C=O

imidazolidine-2,4-dione), 164(C=O Isonicotinohydrazide), 150, 146, 139, 137, 127, 124, 121 ,124 , 121 (C-Ar), 57(CH<sub>r</sub>), 34(CH<sub>i</sub>H<sub>k</sub>).

**N'-(2,5-dioxo-1-(m-tolyl)pyrrolidin-3-yl)isonicotinohydrazide(3C):**

White solid powder; yield 31%; mp= 200-202 °C;IR (KBr, cm-1 ) 3477 (NH amide), 3325(NH), 1708(C=O imidazolidine-2,4-dione), 1643 (C=O Isonicotinohydrazide), 1537, 1492,1471, (C=C arom), 1340 (C-N), <sup>1</sup>H NMR (500 MHz, DMSO) δ 10.47 (d, *J* = 6.0 Hz , 1H,H<sub>n</sub>), 8.81 – 8.69 (m , 2H,H-Ar), 7.74 (dd, *J* = 4.5, 1.8 Hz , 2H,H-Ar), 7.37 (t, *J* = 7.6 Hz, 1H,H-Ar), 7.23 (d, *J* = 7.6 Hz, 1H , H-Ar), 7.07 (d, *J* = 8.9 Hz, 2H, H-Ar) 6.12 (t, *J* = 5.5 Hz, 1H,H<sub>m</sub>), 4.28 (dt, *J* = 8.7, 4.3 Hz, 1H,H<sub>r</sub>), 3.09 (dd, *J* = 17.9, 8.5 Hz, 1H,H<sub>i</sub>), 2.80 (dd, *J* = 17.9, 3.7 Hz, 1H,H<sub>k</sub> ), 2.33 (s, 3H,CH<sub>3</sub>).

**N'-(2,5-dioxo-1-(o-tolyl)pyrrolidin-3-yl)isonicotinohydrazide(4C):**

White solid powder; yield 47%; mp= 184-186 °C;IR (KBr, cm-1 ) IR (KBr, cm-1 ) 3410 (NH amide), 3327 (NH) ,3041 (CH arom), 2987,2933 (CH-aliph), 1784(C=O imidazolidine-2,4-dione), 1712 (C=O imidazolidine-2,4-dione), 1643, 1552,1529,1469, (C=C arom.), 1390 ( C-N).<sup>1</sup>H NMR (500 MHz, DMSO) δ 10.47 (d, *J* = 6.2 Hz , 1H,H<sub>n</sub>), 8.74 (d , *J* = 4.4 Hz, 2H,H-Ar), 7.76-7.75 (m, *J* = 2H,H-Ar), 7.34 (dd, *J* = 4.5,1.2 Hz, 2H,H-Ar), 7.34 (dd, *J* = 4.5,1.2 Hz, 2H , H-Ar), 7.30-7.28 (m, 1H, H-Ar) 7.15 (dd, *J* = 8.2,2.7 Hz, 1H,H-Ar), 7.76-7.75 (m, *J* = 2H,H-Ar), 6.15-6.13 (m, 1H,Hm), ,4.38-4.35 (m,1H,H<sub>r</sub>), 3.20 (dd, *J* = 17.8, 8.6 Hz, 1H,H<sub>i</sub>), 3.11 (dd, *J* = 18.0, 8.6 Hz, 1H,H<sub>k</sub> ), 2.04 (s,CH<sub>3</sub>).

**N-(5-((4-chlorophenyl)amino)-2-oxoethyl)-3-cyclohexyl-4-oxo-2-thioxoimidazolidin-1-yl)isonicotinamide(1D):**

White solid powder; yield 42 %, mp.= 140-143 °C; IR (KBr, cm-1 ): 3479 (NH amide), 3415 (NH), 3118 ,3053 (CH arom.), 2995 (CH-aliph.), 1749 (C=O thioimidazole), 1710 (C=O amide),1678, 1616,1533 (C=C arom.), 1490 (C=S), 1311 (C-N); <sup>1</sup>H NMR (500 MHz, DMSO-d6) δ 11.65(s, 1H, Hd), 10.24 (s, 1H, He), 8.78 (d, *J* = 6.1 Hz, 2H, H-Ar), 7.78 (d, *J* = 6.2 Hz, 2H, H-Ar), 7.55 (d,*J*=8.8Hz, 2H, H-Ar),7.35 (d, *J* = 8.6 Hz, 2H, H-Ar), 4.64 (t, *J* = 4.3 Hz, 1H, Hc), 4.50-4.45 (m, 1H, Hf), 3.20- 2.98 (m, 2H, Ha, Hb), 2.18 (dt, *J*=23.9,12.0Hz ,2H, H-cyclohexyl), 1.83(d, *J*=13.1Hz ,2H, H-cyclohexyl), 1.74 (t,*J*=15.6 2H, H-cyclohexyl),1.65 (d, *J*=12.8 Hz ,1H, H-cyclohexyl)1.29 (s, 2H, H-cyclohexyl), 1.16(t,*J*=13.0,1H, H-cyclohexyl) ; <sup>13</sup>C NMR (126 MHz, DMSO-d6) δ 184 (C=S), 171 (C=O thioimidazole), 166 (ArNHC=O), 164 (NNHC=O), 150, 138, 137, 128, 126, 121,120 (C-Ar), 59 (CHc) , 55 (CHf), 34 (CHaHb), 28, 27, 25.58, 25, 24 (C-cyclohexyl).

**N-(3-cyclohexyl-5-(2-((4-nitrophenyl)amino)-2-oxoethyl)-4-oxo-2-thioxoimidazolidin-1-yl)isonicotinamide(2D):**

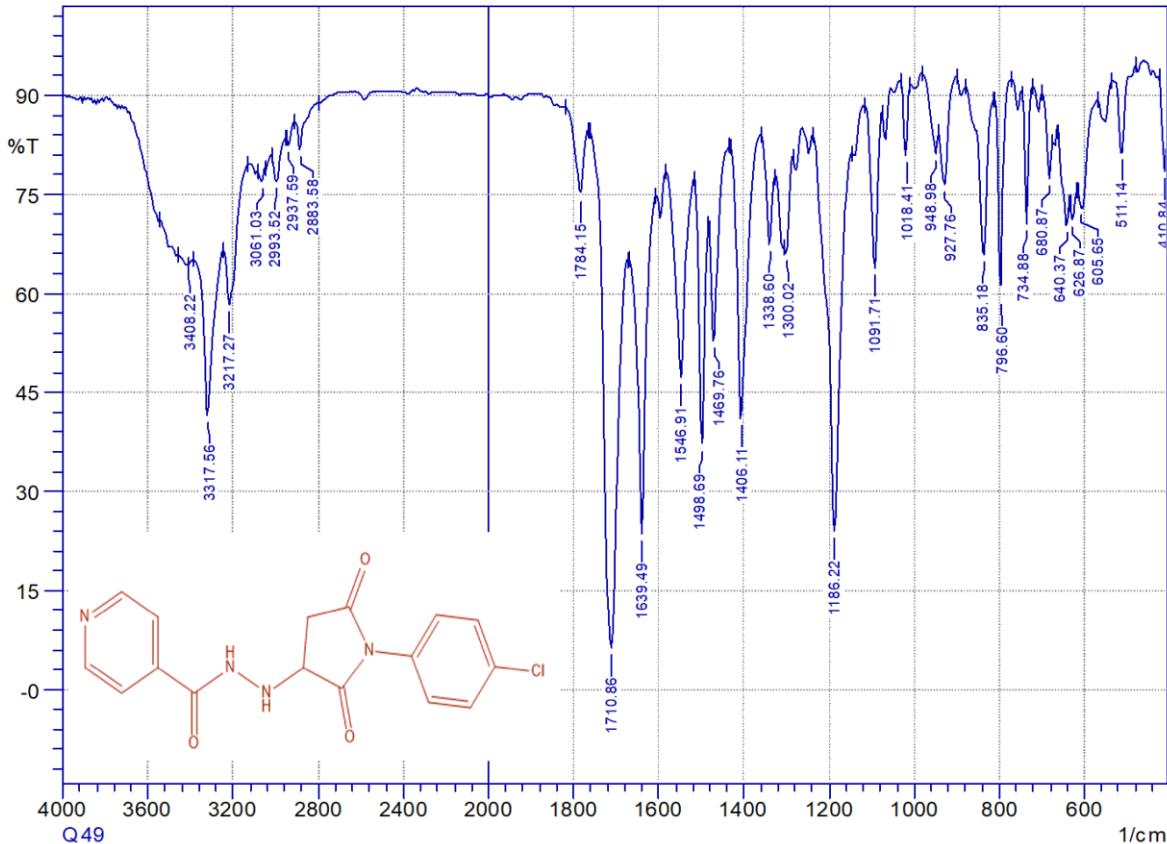
White solid powder; yield 60%; mp= 210-213 °C ; IR (KBr, cm<sup>-1</sup>) 3304 (NH amide), 3151 (NH) ,2929 (CH arom.), 2856 (CH-aliph.), 1766(C=O thioimidazole), 1749 (C=O thioimidazole), 1730 (C=O amide), 1597, 1552,1512, (C=C arom.), 1332 ( C-N). ; <sup>1</sup>H NMR (500 MHz, DMSO-d6) δ 11.66 (s, 1H, Hd), 10.75 (s, 1H, He), 8.81-8.76 (m, 2H, H-Ar), 8.25-8.19 (m, 2H, H-Ar), 7.82-7.74 (m, 4H, H-Ar), 4.68 (t, J = 4.2 Hz, 1H, Hc), 4.49 (ddt, J = 12.2, 7.2, 3.8 Hz ,1H, Hf), 3.12 (d, J = 4.2 Hz, 2H, Ha, Hb), 2.28-2.13 (m, 2H, H-cyclohexyl), 1.84 (d, J=15.0 Hz ,3H, H-cyclohexyl), 1.79-1.71 (m, 2H, H-cyclohexyl),1.66 (d, J=12.4 Hz ,1H, H-cyclohexyl),1.29(ddd, J = 16.0, 10.9, 3.2 Hz,1H, H-cyclohexyl), 1.17 (ddd, J = 16.0, 8.3, 3.3 Hz, 1H,H- cyclohexyl).

**N-(3-cyclohexyl-4-oxo-5-(2-oxo-2-(m-tolylamino)ethyl)-2-thioxoimidazolidin-1-yl)isonicotinamide(3D):**

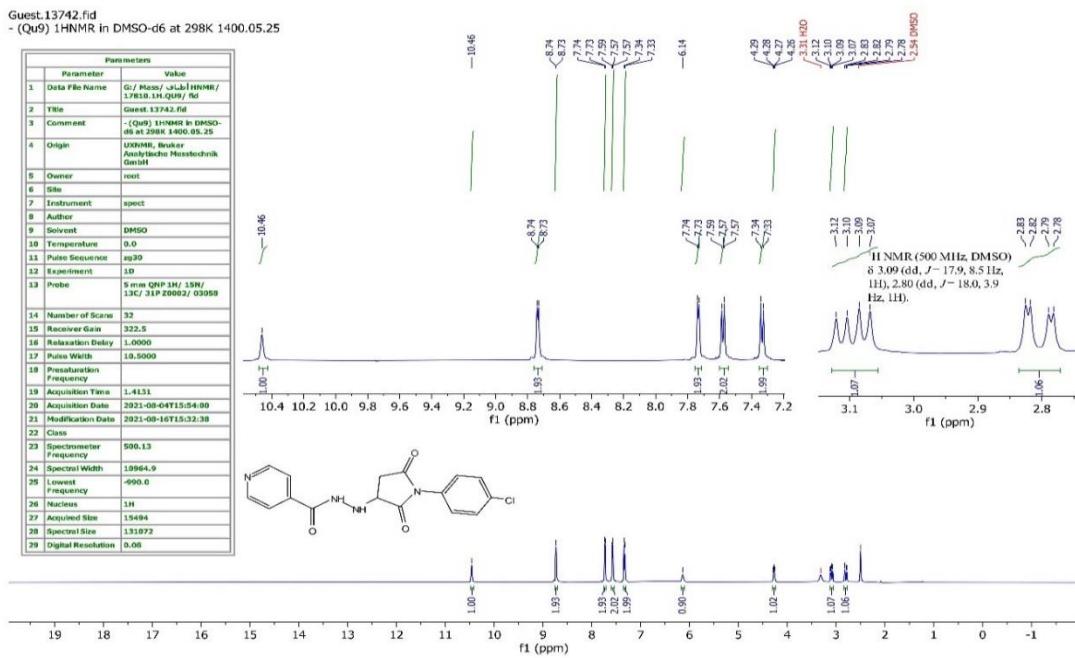
White solid powder; yield 53%; mp= 235-238 °C; <sup>1</sup>H NMR (500 MHz, DMSO-d6) δ 11.62 (s, 1H, Hd), 9.96 (s, 1H, He), 8.79-8.74 (m, 2H, H-Ar), 7.80-7.75 (m, 2H, H-Ar), 7.35-7.28 (m, 2H, H-Ar), 7.16 (t, J = 7.8 Hz, 1H), 6.85 (d, J = 7.5 Hz, 1H), 4.64 (t, J = 4.4 Hz, 1H, Hc), 4.52 – 4.43 (m, 1H, Hf), 3.06 – 2.94 (m, 2H, Ha, Hb), 2.25 (s, 3H,CH<sub>3</sub>), 2.19 (d, J = 18.1 Hz, 2H, H-cyclohexyl), 1.83 (d, J = 13.2 Hz, 2H, H-cyclohexyl), 1.74 (t, J = 14.4 Hz, 2H,H-cyclohexyl),1.65 (d, J=13.2 Hz ,1H, H-cyclohexyl), 1.30 – 1.28 (m, 3H, H-cyclohexyl), 1.17 (t, J = 13.0 Hz, 1H, H-cyclohexyl) .

**N-(3-cyclohexyl-4-oxo-5-(2-oxo-2-(o-tolylamino)ethyl)-2-thioxoimidazolidin-1-yl)isonicotinamide (4D):**

White solid powder; yield 58%; mp= 60-62 °C; <sup>1</sup>H NMR (500 MHz, DMSO-d6) δ 11.62 (s, 1H, Hd), 9.37 (s, 1H, He), 8.84-8.72 (m, 2H, H-Ar), 7.87-7.71 (m, 2H, H-Ar), 7.37 (t, J = 11.0 Hz, 1H,H-Ar), 7.18 (d, J = 7.4 Hz, 1H, H-Ar), 7.12 (t, J = 7.5 Hz, 1H, H-Ar), 7.05 (t, J = 7.4 Hz, 1H, H-Ar), 4.67 (t, J = 4.5 Hz, 1H, Hc), 4.46 (tt, J = 12.6, 3.8 Hz, 1H, Hf), 3.05 (qd, J = 17.1, 4.5 Hz, 2H, Ha, Hb), 2.27-2.12 (m, 4H, H-cyclohexyl), 1.82 (s, 2H, H-cyclohexyl), 1.72 (s, 3H,CH<sub>3</sub>), 1.64 (d, J=13.1 Hz ,1H, H-cyclohexyl), 1.32 – 1.23 (m, 1H, H-cyclohexyl), 1.16 (dd, J = 14.3, 10.9 Hz, 1H) ; <sup>13</sup>C NMR (126 MHz, DMSO-d6) δ184 (C=S), 171(C=O thioimidazole), 166(ArNHC=O), 164 (NNHC=O), 150, 138, 136, 131, 130, 125,124,121, (C-Ar), 59 (CHc) , 55 (CHf), 34 (CHaHb), 28, 27, 25.56, 25.42, 24 (C-cyclohexyl),17(CH<sub>3</sub>).



**Figure (1) FT-IR spectrum of compound 1C**



**Figure (2)  $^1\text{H-NMR}$  spectrum of compound 1C**

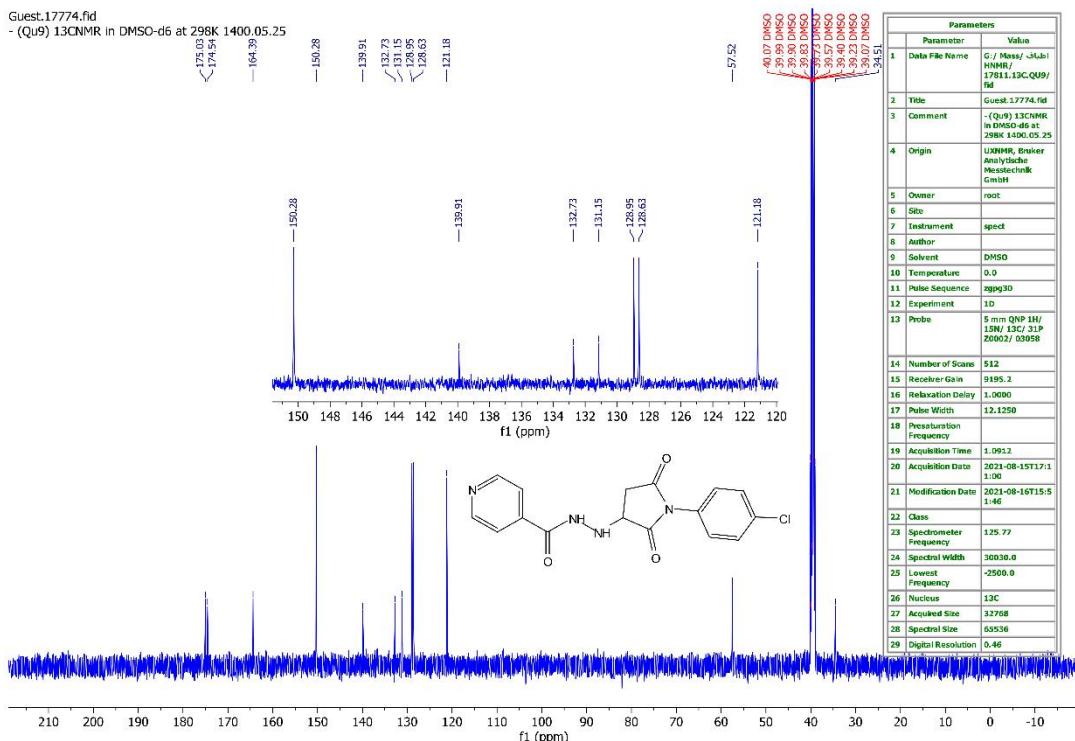


Figure (3)  $^{13}\text{C}$ -NMR spectrum of compound 1C

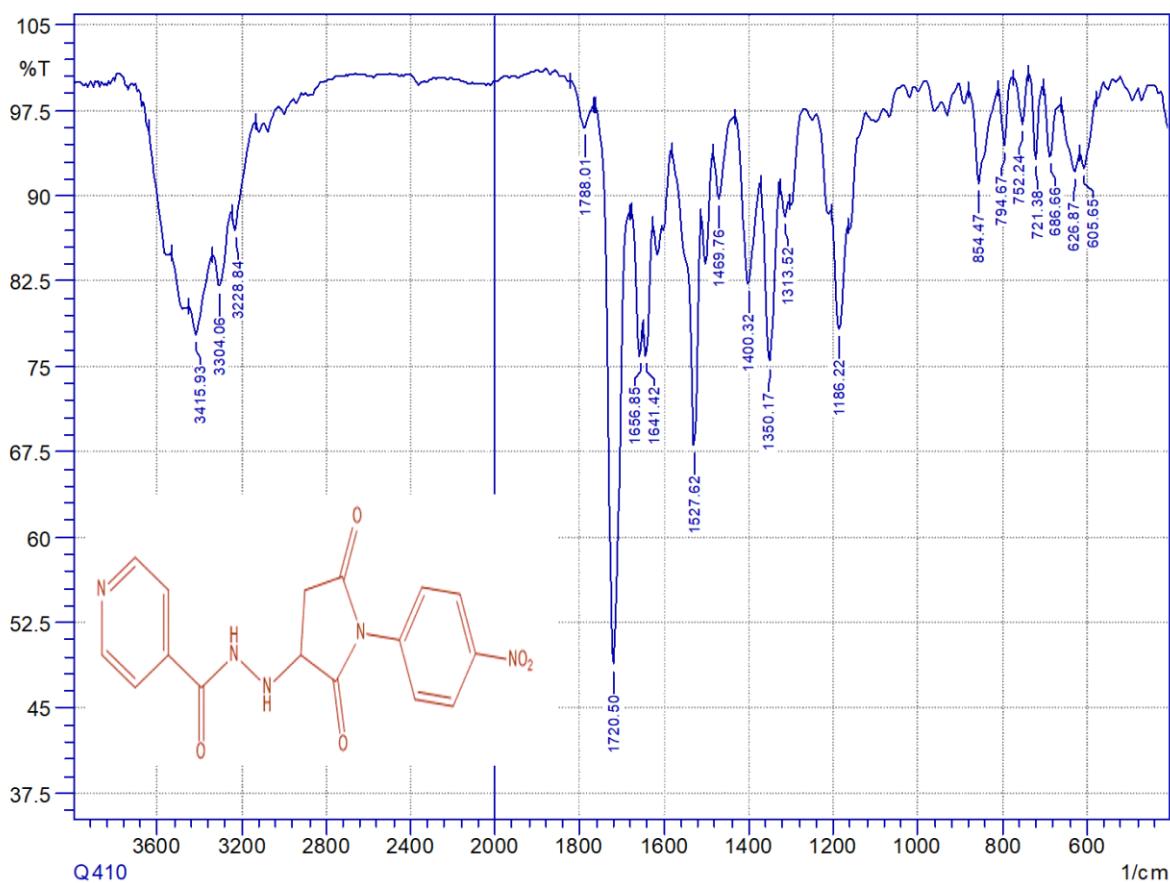


Figure (4) FT-IR spectrum of compound 2C

Guest.13742.fid  
- (Qu10) 1H NMR in DMSO-d6 at 298K 1400.05.25

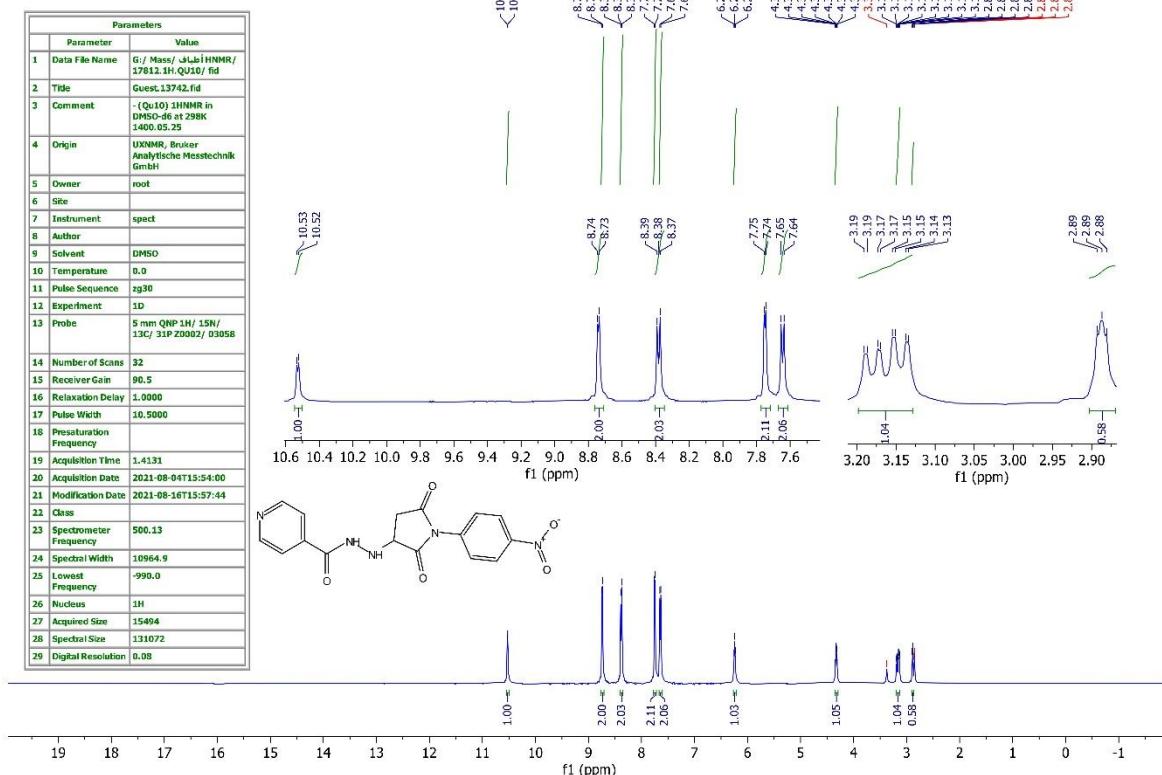


Figure (5) <sup>1</sup>H-NMR spectrum of compound 2C

Guest.17774.fid  
- (Qu10) 13CNMR in DMSO-d6 at 298K 1400.05.25

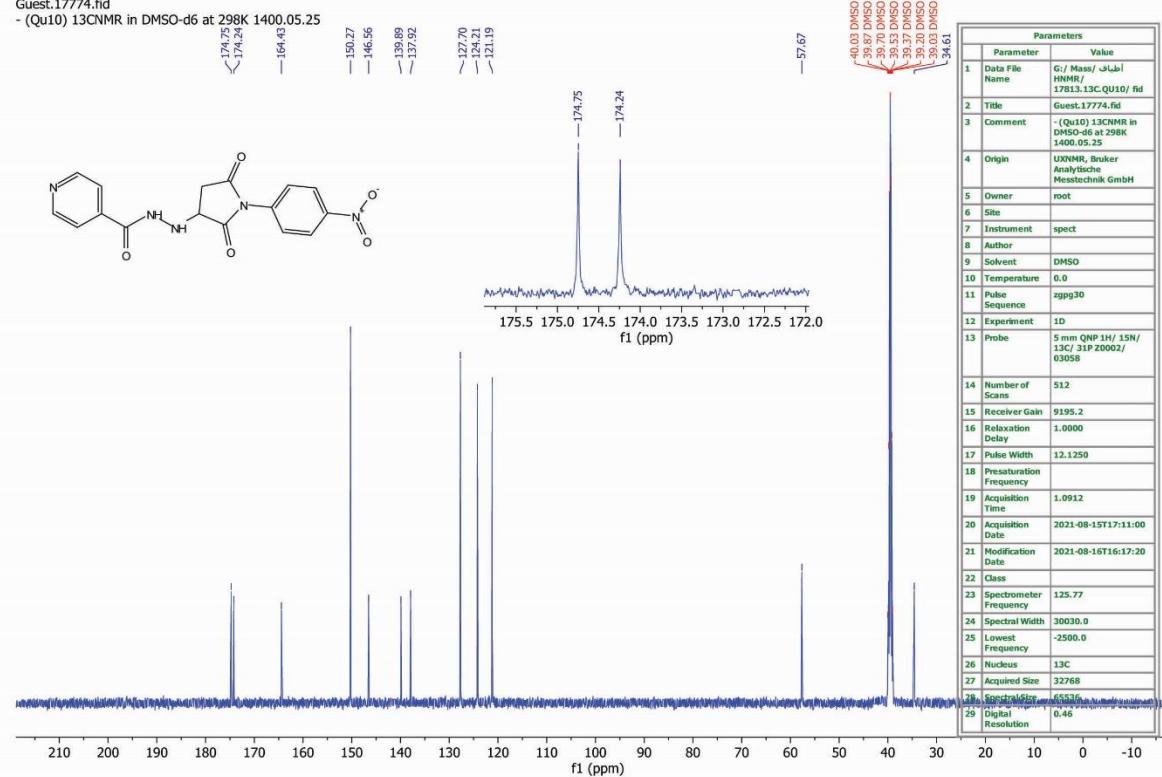
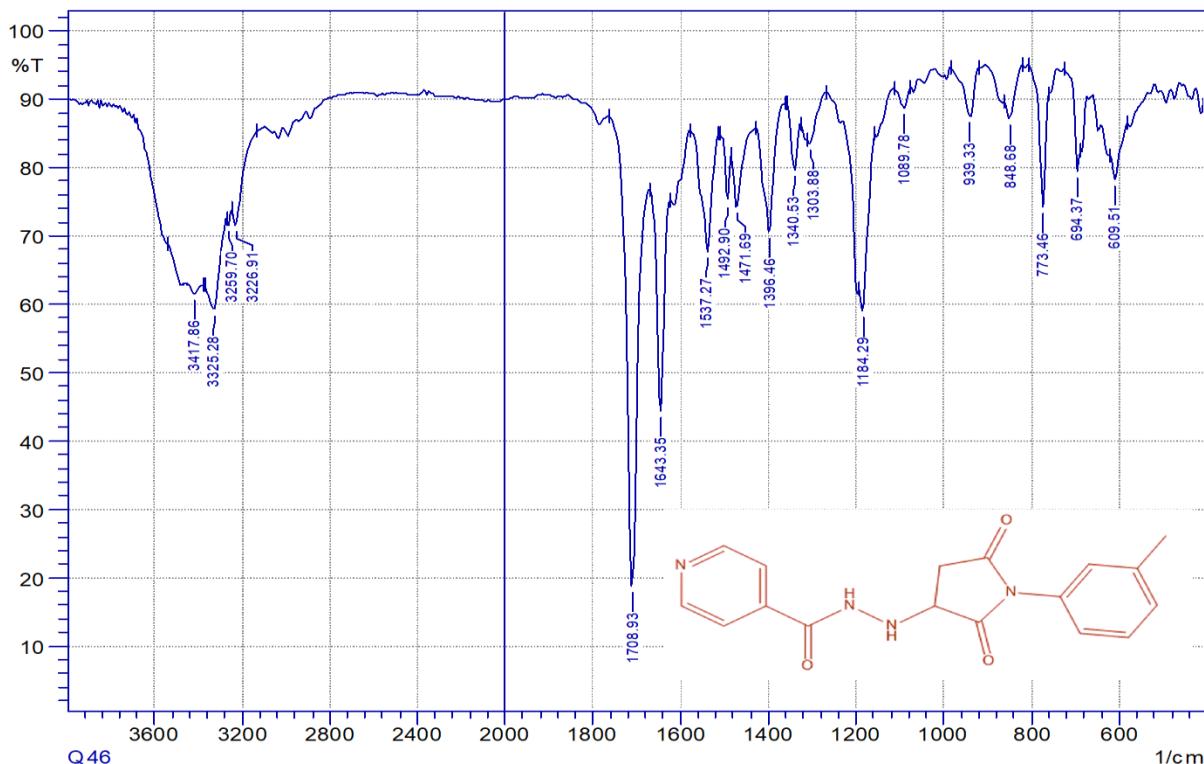


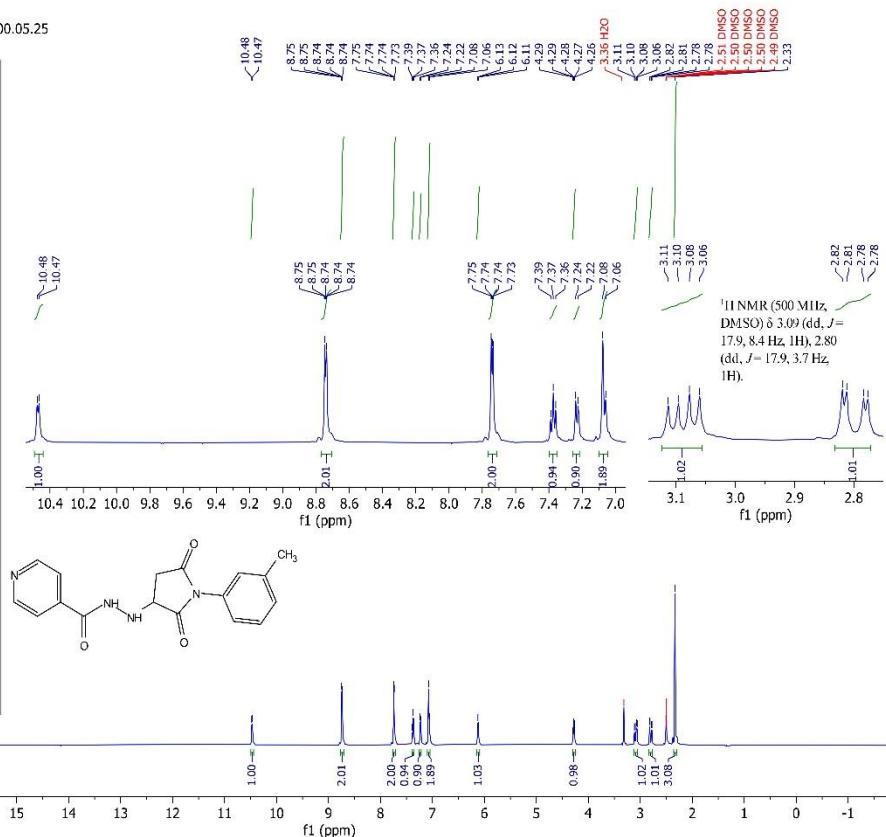
Figure (6) <sup>13</sup>C-NMR spectrum of compound 2C



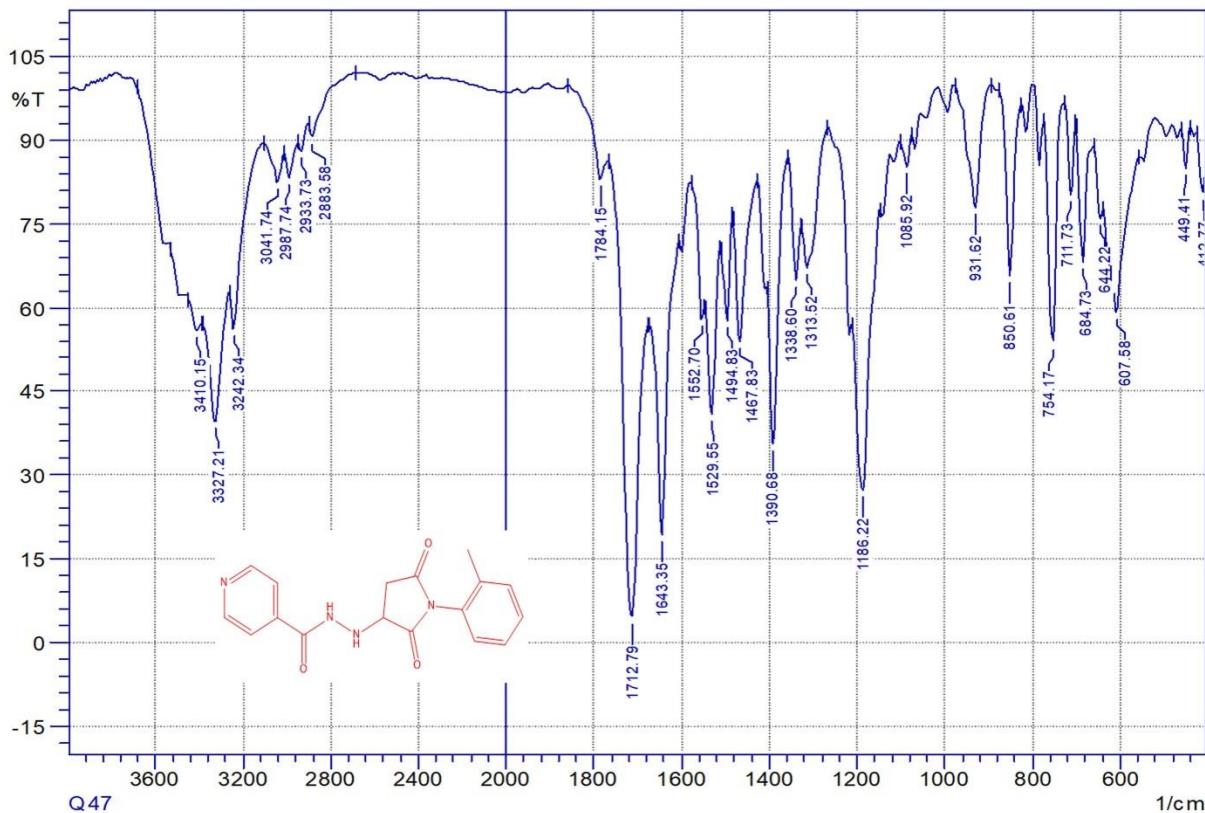
**Figure (7) FT-IR spectrum of compound 3C**

Guest.13742.fid  
- (Qu6) 1HNMR in DMSO-d6 at 298K 1400.05.25

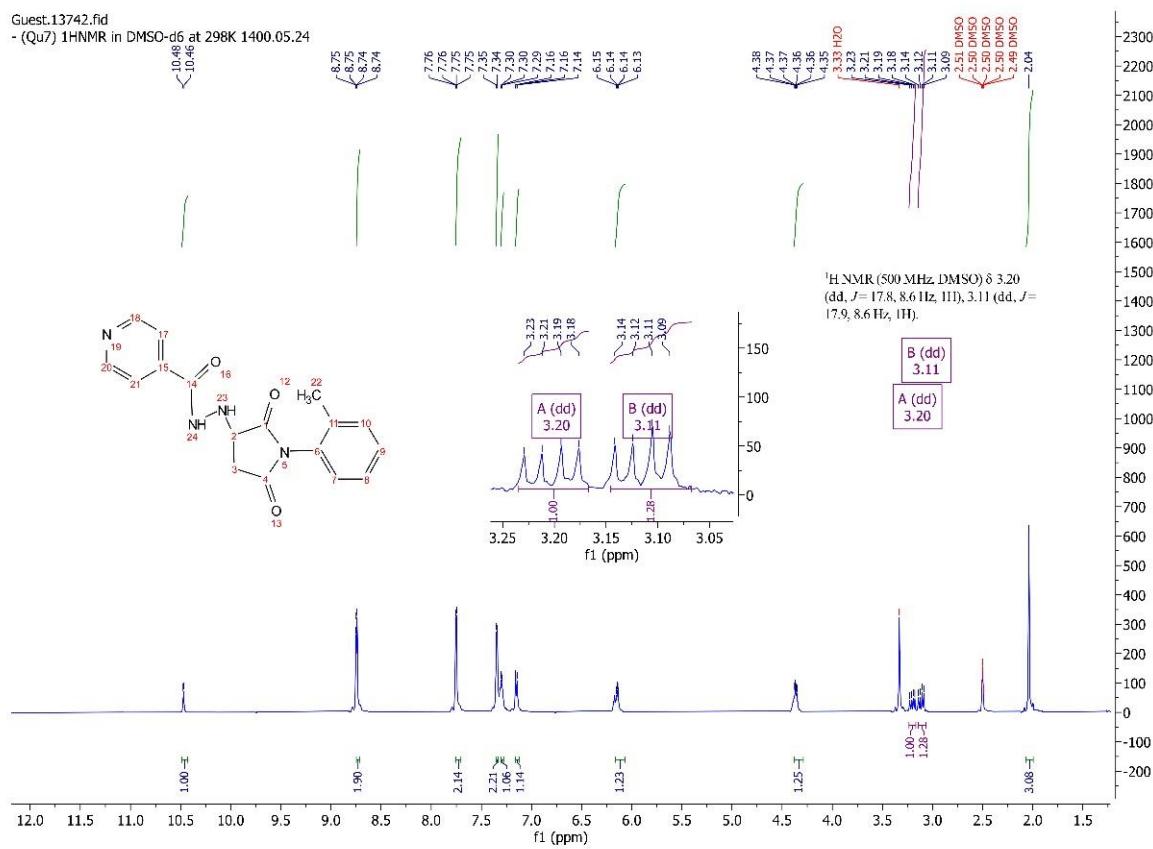
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5	Owner	root
6	Site	
7	Instrument	spect
8	Author	
9	Solvent	DMSO
10	Temperature	0.0
11	Pulse Sequence	zg30
12	Experiment	1D
13	Probe	5 mm QNP 1H/ 15N/ 13C/ 31P 2000/ 03958
14	Number of Scans	32
15	Receiver Gain	181.0
16	Relaxation Delay	1.0000
17	Pulse Width	10.5000
18	Prestimulation Frequency	
19	Acquisition Time	1.4131
20	Acquisition Date	2021-08-04T15:54:00
21	Modification Date	2021-08-16T13:34:40
22	Class	
23	Specrometric Frequency	500.13
24	Spectral Width	10964.9
25	Lowest Frequency	-999.0
26	Nucleus	1H
27	Acquired Size	15494
28	Spectral Size	131072
29	Digital Resolution	0.08



**Figure (8 )**  $^1\text{H}$ -NMR spectrum of compound 3C



**Figure (9) FT-IR spectrum of compound 4C**



**Figure (10 )  $^1\text{H-NMR}$  spectrum of compound 4C**

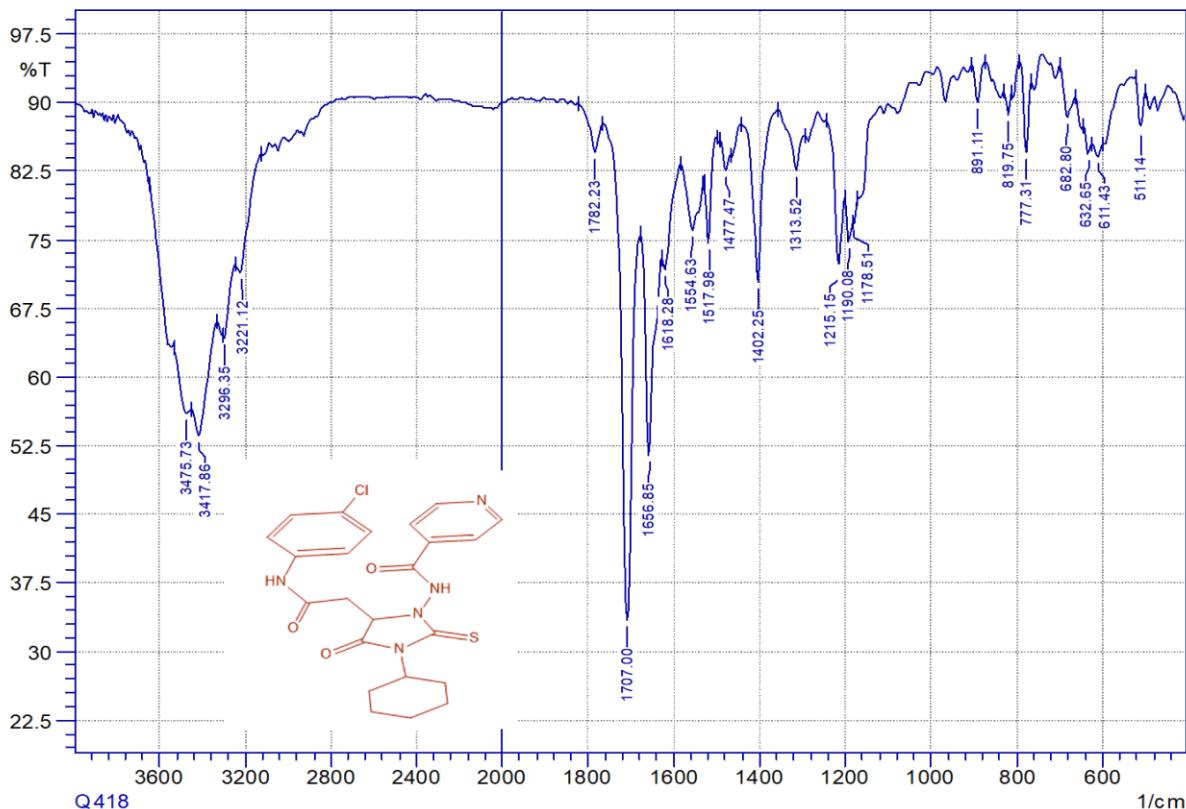


Figure (11) FT-IR spectrum of compound 1D

Guest.12656.fid  
Q6 1H NMR in DMSO-d<sub>6</sub> at 298K 1400.01.21

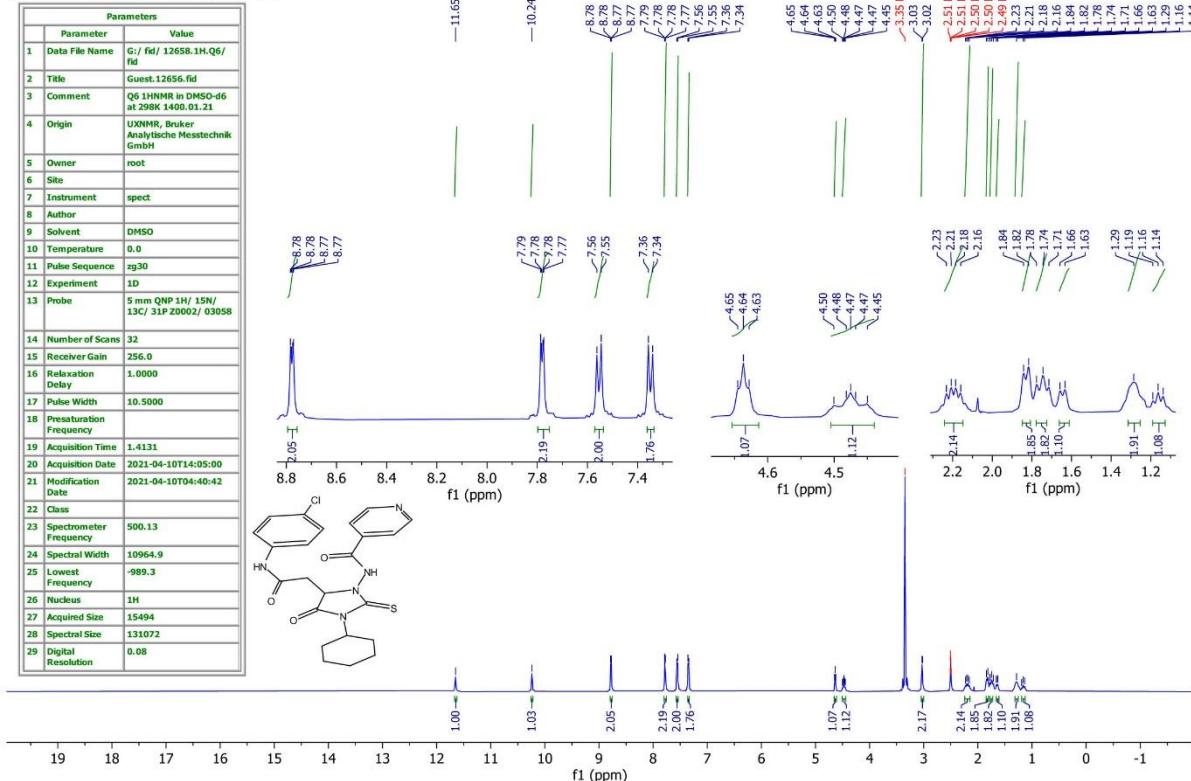


Figure (12 ) <sup>1</sup>H-NMR spectrum of compound 1D

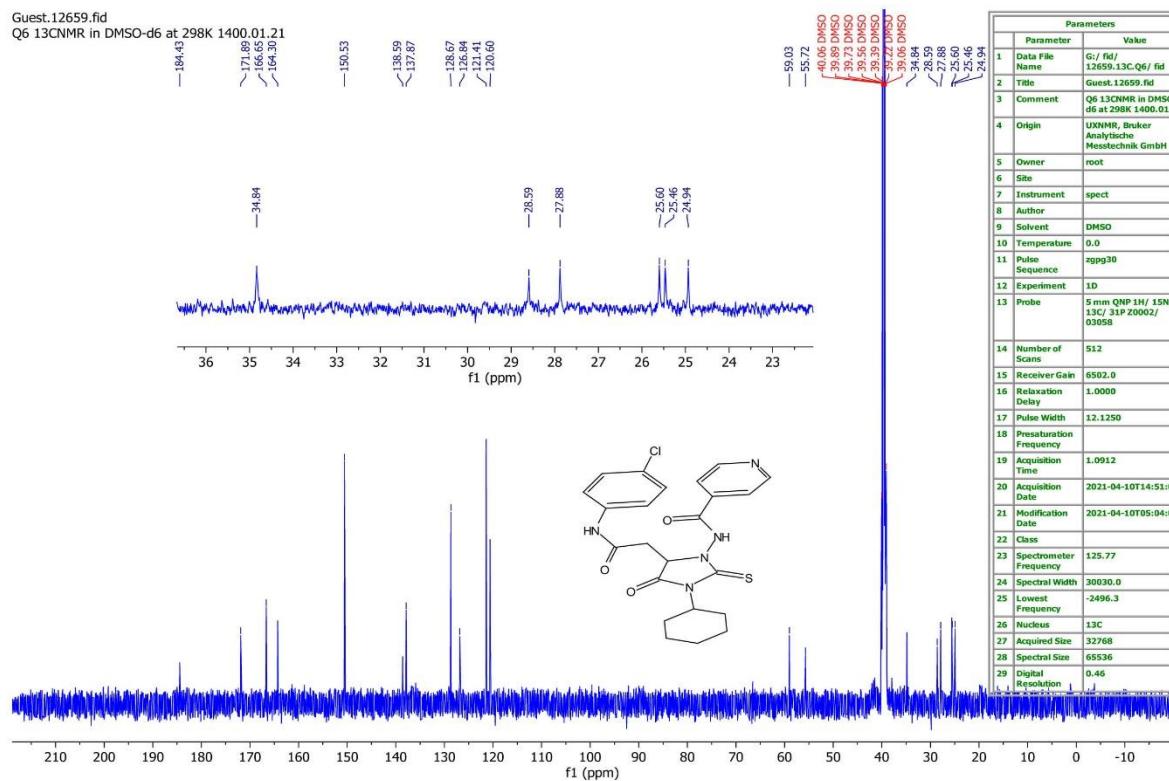


Figure (13)  $^{13}\text{C}$ -NMR spectrum of compound 1D

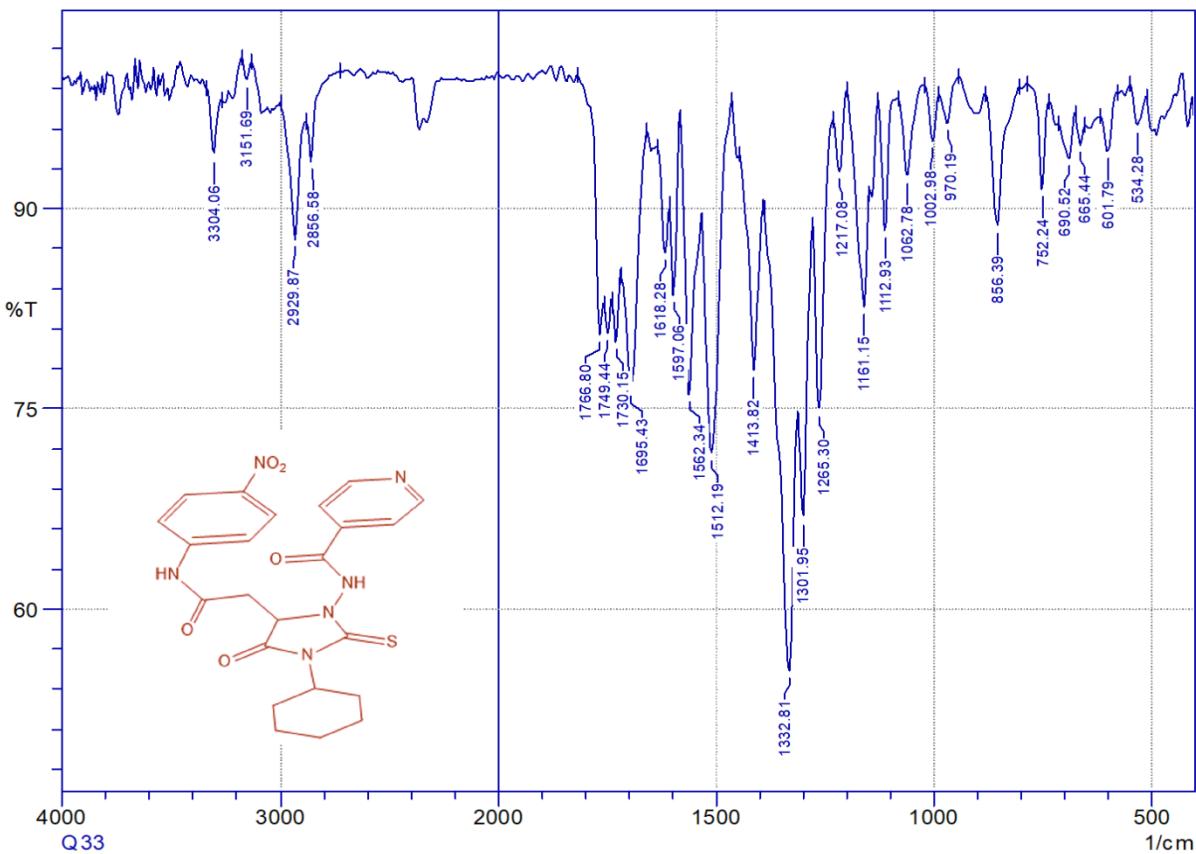


Figure (14) FT-IR spectrum of compound 2D

QUSQ17.10.fid

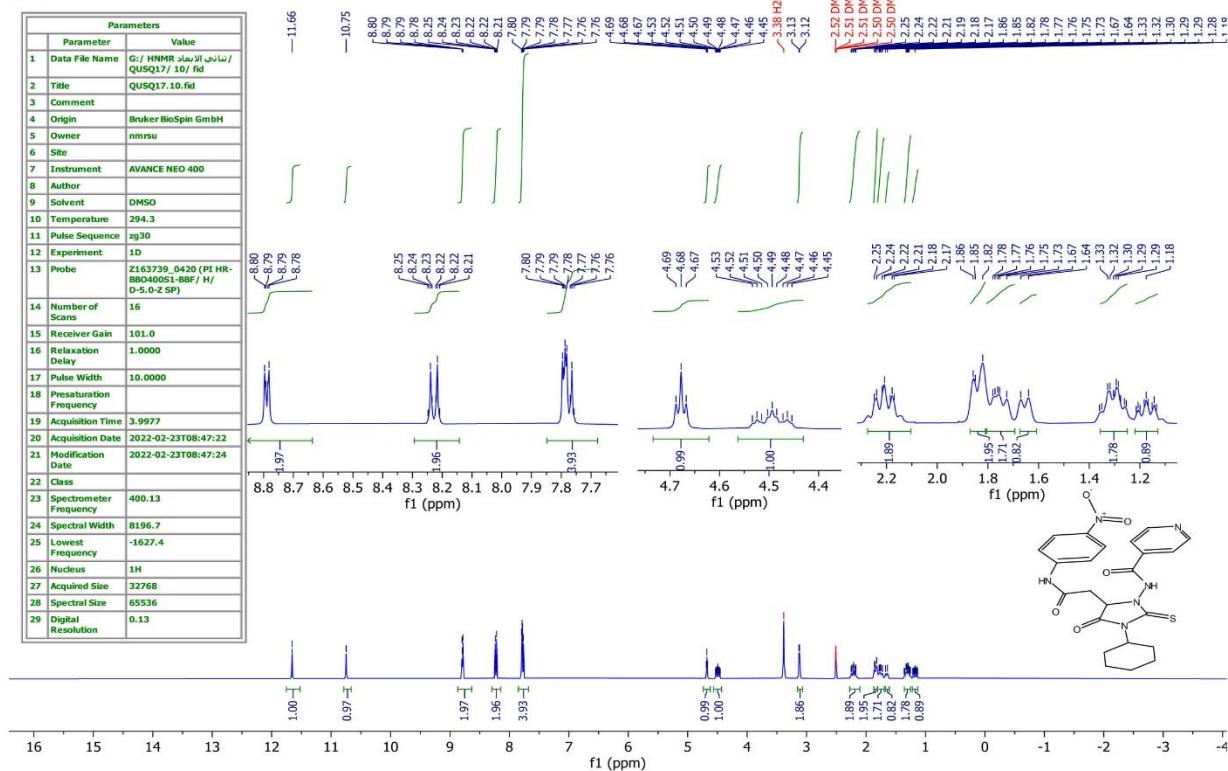


Figure (15)  $^1\text{H}$ -NMR spectrum of compound 2D

Guest.13792.fid

- (Q17) 1HNMR in DMSO-d6 at 298K 1400.06.06

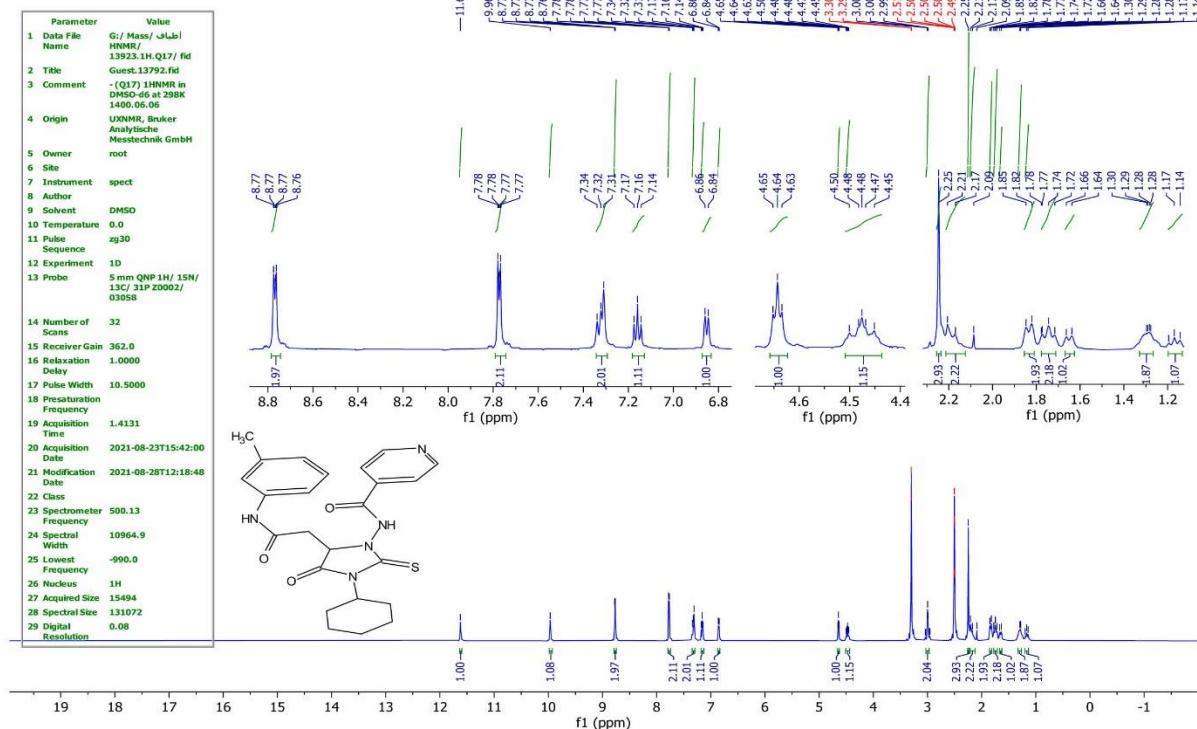
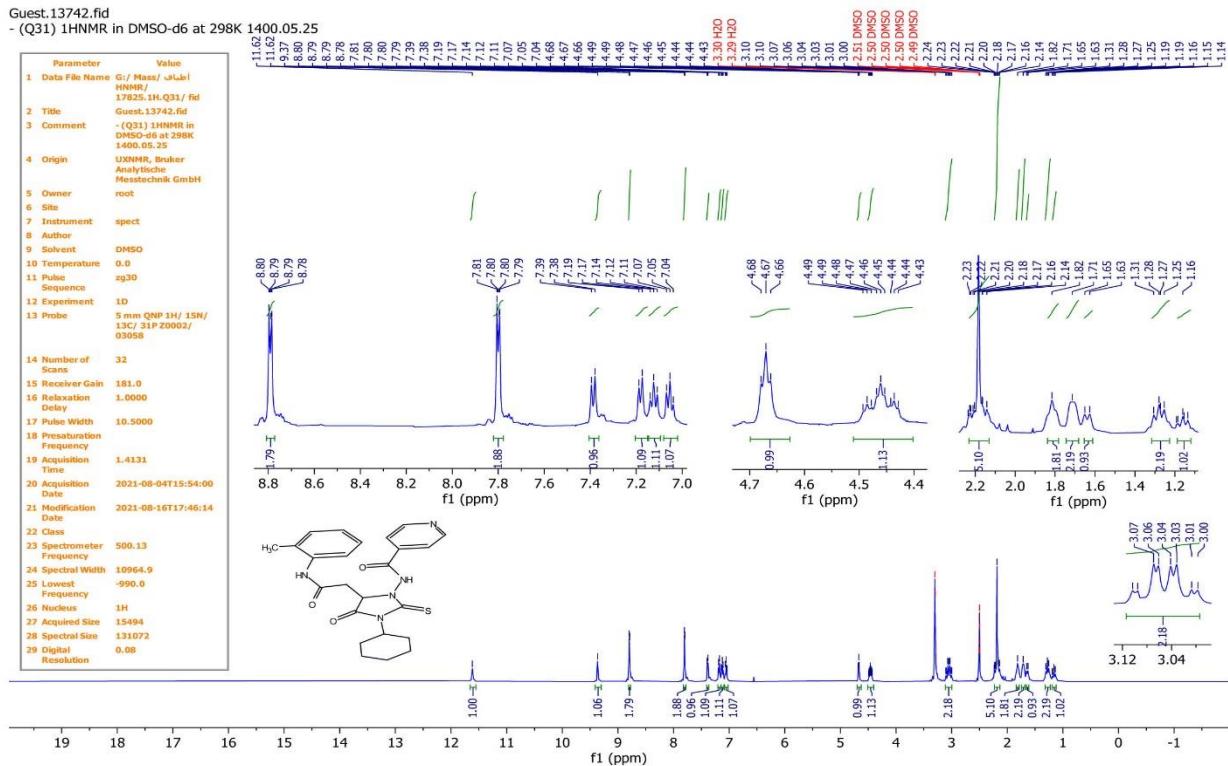
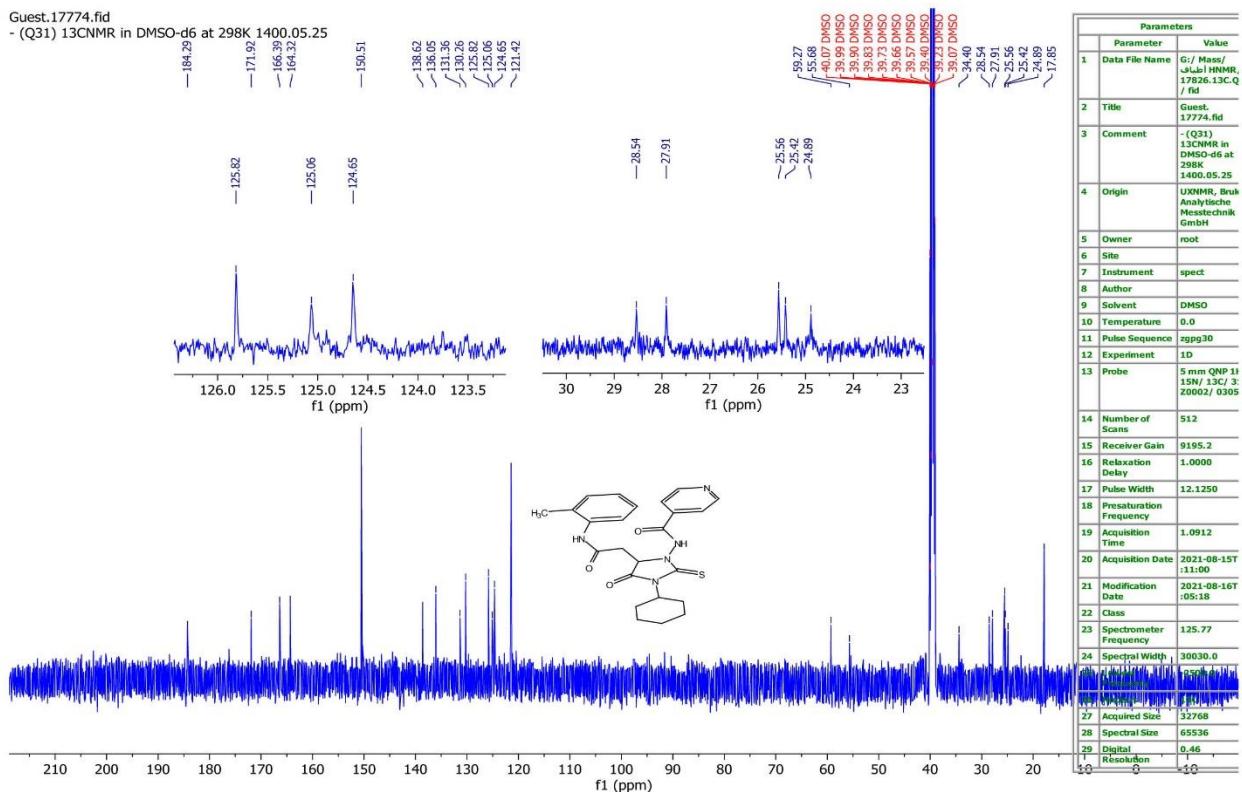


Figure (16 )  $^1\text{H}$ -NMR spectrum of compound 3D



**Figure (17)**  $^1\text{H}$ -NMR spectrum of compound 4D



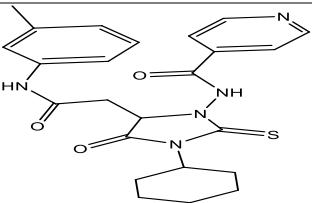
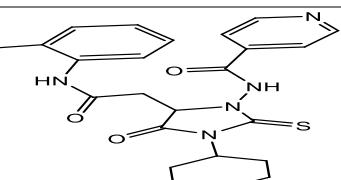
**Figure (18 )**  $^{13}\text{C}$ -NMR spectrum of compound 4D

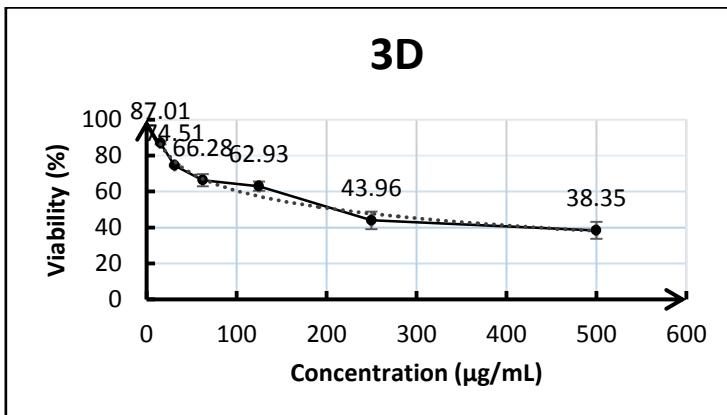
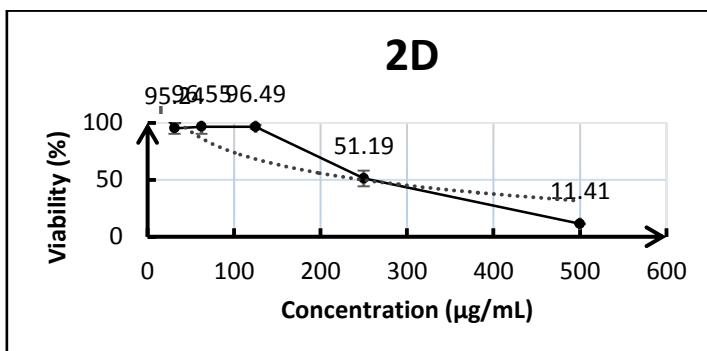
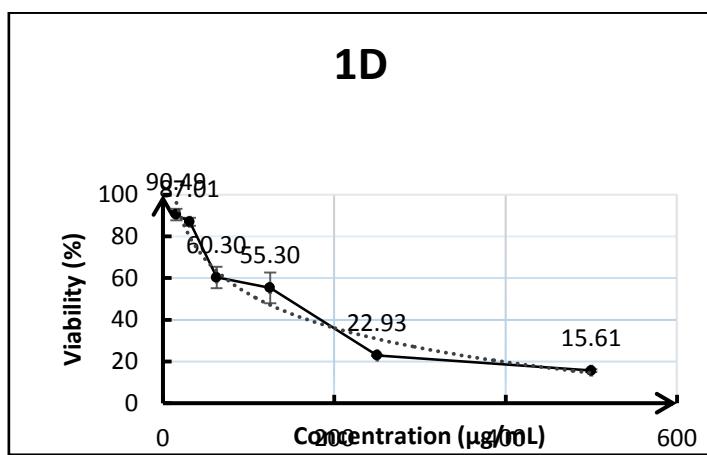
## The activity of the prepared compounds against prostate cancer. [23]

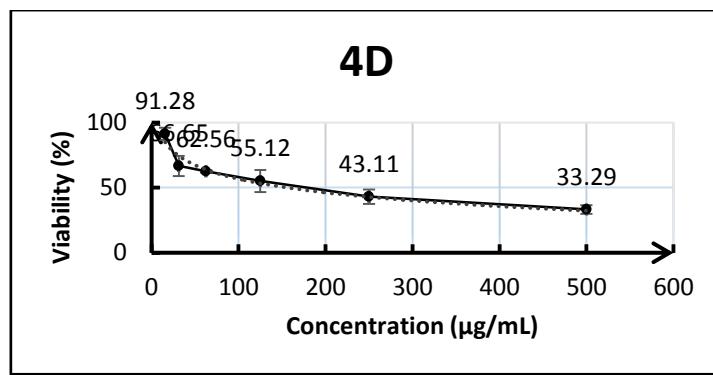
Compound symbol	Compound structure	PC <sub>3</sub> cells IC <sub>50</sub> in µg/mL	
1D		110.41	Inactive
2D		243.053	Inactive

The discovery of anti-prostate cancer drugs is one of the most challenging challenges faced by modern medicine, and the reason is due to the large number of mutations that occur in the Ar (Androgen receptor). It has been found that the thiohydantoin nucleus is an excellent drug carrier for the treatment of prostate cancer. In this study, we examined a series of thiohydantoin derivatives prepared as anti-prostate cancer . Depending on the IC<sub>50</sub> value of the compounds of the prepared series, we note that there is a noticeable difference in the activity of the compounds against prostate cancer cells. This discrepancy is attributed to the difference of functional groups on the aromatic ring in the prepared compounds. Depending on the IC<sub>50</sub> value, none of the compounds under study showed anti-prostate cancer activity, Table 1.

**Table (1) shows the IC<sub>50</sub> values of the compounds under study versus PC3 cells.**

<b>3D</b>		211.281	Inactive
<b>4D</b>		156.56	Inactive





**Figure 1** the graphs of the response of the compounds under study against prostate cancer cells

### The activity of the prepared compounds against bacteria and fungi [24] [25]

Thiohydantoin (1D-4) and maleimide derivatives (1C-4) are found in a wide range of heterocyclic compounds, and the biological activity of this type of compound usually depends on the nature of the substituting groups on the ring. Studies confirm that the compounds that contain in their structure the nucleus of the thiohydantoin have anti-bacterial and anti-fungal properties [26] [27]. In this study, we evaluated the prepared series compounds against selected strains of bacteria and fungi. The prepared compounds were studied as antibacterials *E.Coli* and *Staph*, compared with Spiromycin as a positive control, and they were evaluated as against fungi on two types of fungi, *Candida albican* and *Aspargilus*.

The compounds (4C,1D,3D,4D) showed *anti-staph* activity, as the 4C compound showed high *anti-staph* activity. None of the studied compounds showed *anti-E.Coli* activity except for the compound 4C, which showed medium activity. The compounds (1C,4C,2D,3D) showed *anti-aspargilus* activity, while the rest of the compounds under study were not activity. The data indicate that all the compounds under study have antifungal activity for *Candida albican*, Table 2.

Table (2) shows the activity of the compounds under study against selected types of bacteria and fungi.

Compound symbol	The biological activity of the bacteria under study		The biological activity of the fungi under study	
	<i>Staph</i>	<i>E.Coli</i>	<i>Aspargilus</i>	<i>Candida albican</i>
1C	-	-	<b>12</b>	19
2C	-	-	-	16
3C	-	-	-	13
4C	<b>20</b>	<b>15</b>	<b>12</b>	20
1D	16	-	-	21
2D	-	-	11	25
3D	15	-	15	32
4D	15	-	-	37

Spiromycin	23	23	No tested	No tested	Negative :-
<b>No activity</b>					

## CONCLUSION

The advantage of preparing this series of compounds is easy access to the product and ease of purification. The graph of the response of the compounds under study against prostate cancer cells indicates that there is a significant effect of the aromatic ring functional groups in their activity against prostate cancer cell lines. Depending on the IC50 values of the compounds under study, none of the compounds showed anti-prostate cancer activity, and we need further study to increase the activity of the prepared series compounds against prostate cancer cells. The data of the compounds under study indicate that there is a significant effect of the functional groups in the aromatic ring in its anti-bacterial and anti-fungal activity. Compound 4C showed high activity against *Staph* bacteria due to the presence of the methyl group in the ortho position.

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