

Synthesis and Identification some of heterocyclic compounds from 2-Aminobenzimidazole

Shaimaa Adnan

shemaadnan@yahoo.com

College of Education, University of Al-Qadisiya

Kasim Hassan

College of Science, University of Babylon

Hassan Thamer

College of Education for Women- University of Kufa

(NJC)

(Received on 10/11/2013)

(Accepted for publication 23/2/2014)

Abstract

This research involved heterocyclic compounds such as (Thiazolidine derivatives, imidazolidin derivatives, oxazole derivatives, imidazole derivatives and pyridine derivatives) were prepared by reaction 2-aminobenzaldehyde with benzaldehyde derivatives (4-dimethylaminobenzaldehyde, furfural, vanillin) to get shiff base (1-3), N-(4-(dimethylamino) benzylidene)-1H-benzo[d]imidazol-2-amine(1), N-(furan-2-ylmethylene)-1H-benzo[d]imidazol-2-amine(2), 4-(((1H-benzo[d]imidazol-2-yl) imino) methyl)-2-methoxyphenol (3).

The cyclization of (1-3) with α -aminoacid (glycine) give the corresponding 3-(1H-benzo[d] imidazol-2-yl)-2-(4-(dimethylamino) phenyl) imidazolidin-4-one (4), 3-(1H-benzo[d] imidazol-2-yl)-2-(furan-2-yl) imidazolidin-4-one (5), 3-(1H-benzo[d] imidazol-2-yl)-2-(4-hydroxy-3-methoxyphenyl) imidazolidin-4-one (6), and when cyclization of (1-3) with thioglycollic acid give the corresponding 3-(1H-benzo[d] imidazol-2-yl)-2-(4-(dimethylamino) phenyl) thiazolidin-4-one (7), 3-(1H-benzo[d]imidazol-2-yl)-2-(furan-2-yl) thiazolidin-4-one(8), 3-(1H-benzo[d] imidazol-2-yl) -2-(4-hydroxy-3-methoxyphenyl) thiazolidin-4-one (9).

As well as the reaction of 2 - Amino benzoimidazole with chloro acetyl chloride to get N-(1H-benzo[d] imidazol-2-yl)-2-chloroacetamide which reacts with urea , thiourea , thiosemicarbazide, 2-aminobenzimidazole, 2-aminopyridine give the corresponding N5-(1H-benzo[d] imidazol-2-yl) oxazole-2,5-diamine (11),N5-(1H-benzo[d] imidazol-2-yl) thiazole-2,5-diamine(12),N-(1H-benzo[d] imidazol-2-yl)-2-(2-carbamothioylhydrazinyl) acetamide(13),2-((1H-benzo[d] imidazol-2-yl) amino)-N-(1H-benzo[d] imidazol-2-yl) acetamide (14) , N-(1H-benzo[d] imidazol-2-yl)-2-(pyridin-2-ylamino) acetamide (15) . All this compounds characterized by means of FT- IR ,and some of the compoundesby means $^1\text{H-NMR}$,and $^{13}\text{C-NMR}$ and follow reaction by R_f - TLC and Measure ment melting point .

Key words:- 2-aminobenzimidazole ,Thiazolidin , Imidazolidin, oxazole.

الخلاصة

تضمن البحث تحضير مركبات حلقيه غير متاجنسة مثل (مشقات اميدازولدين، مشقات ثازولدين، مشقات اوكسازول، مشقات اميدازول، مشقات بريدين) وذلك من تفاعل 2- امينوبنزوميدازول مع مشقات البنزليهيد (4-ثنائي مثيل امينو بنزليهيد فورفural (فانلين) لتحضير قواعد شف (3-1) وهي نتروجين-(4-ثنائي مثيل امينو) بنزليدين)-1هيدروجين-بنزوميدازول-2-امين (1)، نتروجين-(فيوران-2-ايل مثيل)-1هيدروجين-بنزوميدازول-2-امين(2) و4-(((1هيدروجين-بنزوميدازول-2-ايل) امينو)مثيل)-2-ميوكسي فينول (3). تم تفاعل هذه المركبات مع الكلايسين مره لنحصل على مشقات الاميدازولدين (4-6) وهي 3-(1هيدروجين-بنزوميدازول-2-ايل)-2-(4-ثنائي مثيل امينو) فنيل)اميدازولدين-4-اون(4)، 3-(1هيدروجين-اميدازول-2-ايل)-2-(فيوران-2-ايل)اميدازولدين-4-اون (5) و3-(1هيدروجين-بنزوميدازول-2-ايل)-2-(4-هيدروكسي-3-ميوكسي فنيل)اميدازولدين-4-اون (6). كذلك تفاعل (1-3) مع حامض الثايوكلاتيكول لنحصل على مشقات الثايزولدين(7-9) وهي 3-(1هيدروجين-بنزوميدازول-2-ايل)-2-(4-ثنائي مثيل امينو) فنيل)ثايزولدين-4-اون(7)، 3-(1هيدروجين-اميدازول-2-ايل)-2-(فيوران-2-ايل)ثايزولدين-4-اون (8) و3-(1هيدروجين-بنزوميدازول-2-ايل)-2-(4-هيدروكسي-3-ميوكسي فنيل)ثايزولدين-4-اون(9). كذلك تفاعل 2- امينو بنزوميدازول مع كلورواسيتايل كلورايد لنحصل على نتروجين-اهيدروجين-بنزوميدازول-2-ايل)-2-كلورواستمايد(10)، الذي يتفاعل مع كل من البيريا والثايوبيوريا وثليوسيماكاربازايد و2-امينوبنزوميدازول و2-امينو بريدين لنحصل بالمقابل على نتروجين(1هيدروجين-بنزوميدازول-2-ايل)اوكسازول-2 و5-ثنائي امين(11)، نتروجين-(1هيدروجين-بنزوميدازول-2-ايل)ثايزول-2 و5-ثنائي امين (12) و نتروجين-(1هيدروجين-بنزوميدازول-2-ايل)-2-(2-كارباميثيل هيدرازيل) استمايد(13)، 2-((هيدروجين-بنزوميدازول-2-ايل)(امينو)-نتروجين-(1هيدروجين-بنزوميدازول-2-ايل)استمايد (14)، نتروجين-(1هيدروجين-بنزوميدازول-2-ايل)-2-(بريدين-2-ايل-امينو)استمايد (15).

كل هذه المركبات تم تشخيصها بوساطة مطيافية الاشعه تحت الحمراء وبعضها بوساطة مطيافية الرنين النووي المغناطيسي و متابعة التفاعل بクロموتوغرافيا الطبقة الرقيقة وقياس درجة الانصهار C-13

الكلمات المفتاحية :- المركبات الحلقيه غير المتاجنسة, 2-امينوبنزوميدازول, ثايزولدين ، اميدازولدين و اوكسازول.

Introduction

Benzimidazole derivatives play important role in medical field with many pharmacological activities such as antimicrobial, antiviral, antidiabetic and anticancer activity⁽¹⁻³⁾. Therefore substituted benzimidazoles have attracted the interest of various research group,

especially since it has been reported that the influence of the substitution at 1,2 and 5-positions is very important for their pharmacological effect^(4,5). the benzimidazole moiety express significant activity against several viruses such as HIV4, Herpes(HSV-1) 5 and influenza6. Bisbenzimidazoleis DNA-

minor groove binding agents possessing anti-tumour activity^(6,7).

imidazolidin-dione are belongs to heterocyclic compound, which has a wide range of biological and pharmacological properties such as antimicrobial activity (antifungal, antibacterial), antitumor, antiinflammatty , anti HIV, anti-hypertensive, hydantoin exhibits diverse biological activities, such as anticonvulsant, antifungal activities, antithyroidal, antiviral, , tuber culosis, anti arrhythmic and anti convulsant⁽⁸⁻¹⁰⁾.

The versatile uses of thiazolidinones as anaesthetics, anti-convulsants, amoebicides, hypotensive and tuberculostatic agents have stimulated a considerable interest to explore the possible synthesis of new potential compounds in which the thiazolidinone ring is fused with another biologically active nucleus. With a view in achieving such a system, the thiazolidine was fused with benzimidazole. Benzimidazole nucleus was chosen because certain 2-amino benzimidazole were found to possess some anti-viral activity⁽¹¹⁻¹⁵⁾.

The oxazole moiety was chosen for conducting property studies due to the significant applications like versatile biological activities , application as important precursors in organic transformations , fluorescent whitening agents , and scintillating compounds⁽¹⁶⁾. A large number of natural products, in particular from the marine environment, contain thiazole, oxazole, heterocycles. In many cases, promising anti-tumor, antibacterial, anti-viral, anti-malaria and anthelmintic activities have been identified forthese compounds⁽¹⁷⁻¹⁹⁾.

Experimental Apparatus

(FTIR)Spectra(4000-400cm⁻¹)in KBr disk were recorded on aSHIMADZU

FTIR-8400S fourier.transform. melting point were measured using Stuart, UK.

¹HNMR and ¹³C-NMR were recorded on fourier transformation bruker spectrometer ,operating at (400MHz) with (DMSO-ds) measurments were made at Department of chemistry ,kashan university.Iran.

General method of synthesis of schiff bases compounds (1-3)⁽²⁰⁾

A mixture of equimolar quantities (0.01mol) of aromatic benzaldehyde and 2-aminobenzoimidazol was refluxed for 20 min in 30 ml of ethanol. The reaction mixture was cooled and kept for (24 hs) . The crystals found was filtered , dried and recrystallized from ethanol to give compounds (1-3) .

General method of synthesis of imidazolidin-4-one derivative (4-6)

A mixture of schiff bases(1-3) (0.001mol) dissolved in THF (15mL) and glycine (0.001mol)was dissolved in THF (15mL)and refluxed for 24 hs.The reaction was then cooled and the resulting final (4-6) ,recrystallized from ethanol.

General method of synthesis of thiazolidinones (7-9)

A mixture of schiff bases (1-3) (0.001mol)and thioglycollic acid (0.001mol) dissolved in 1,4 dioxane (20mL), anhydrous zinc chloride(0.7gm)was added and refluxed for 8 h. The reaction was then cooled and the resulting solid was washed with sodium bicarbonate solution and final (7-9) recrystallized from absolute ethanol.

method of synthesis of (10)⁽²¹⁾

Chloroacetyl chloride (0.01 mol) was added to a solution of (0.01mol) 2-amino benzo imidazol in benzene (20 ml). The

mixture was stirred at room temperature for 2-3 hours and poured onto ice. The separated solid was filtered, washed with water, dried and recrystallized from aqueous ethanol

method of synthesis of (11)⁽²²⁾

To a solution of (10) (3.5 mmol) in absolute ethanol (60 mL), urea (3.5 mmol) was added. The mixture was heated at reflux for 30 min. cooling after neutralized by 10% NaOH, precipitate was collected and recrystallized from ethanol, to give (11) .

method of synthesis of (12)⁽²²⁾

To a solution of (10) (3.5 mmol) in absolute ethanol (60 mL), thiourea (3.5 mmol) was added. The mixture was heated at reflux for 30 min. After cooling, the precipitate was collected and recrystallized from ethanol–10 % sodium hydroxide, to give(12)

method of synthesis of (13)

(0.01 mole) of (10) in 30 ml ethanol was added to (0.01 mole) thiosemicarbazide and Drops of pyridine , The mixture was stirred at room temperature for 24 hours

,left it at room temperature for 24 hs. The precipitate was filtered off and then recrystallized from ethanol.

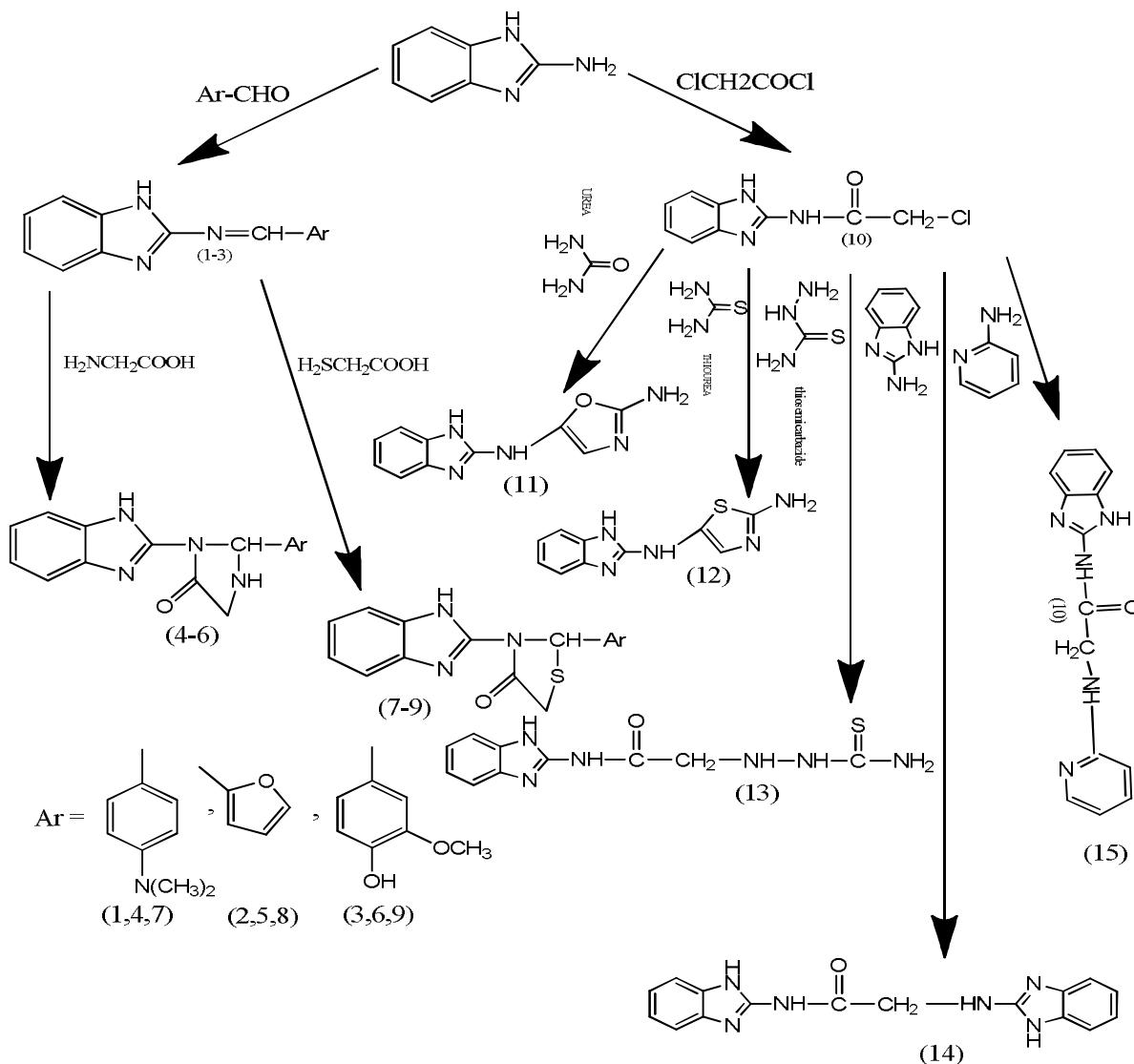
method of synthesis of (14)⁽²³⁾

To compound (10)(0.001 mole) dissolved in 20ml, ethanol. 2-aminobenzimidazole (0.001mole) was added gradually. When addition was complete reaction mixture was reflex for 6 hr. After the reaction excesses of ethanol and aryl amine were recovered distillation. The residue was washed with sodium bicarbonate to remove the acid impurities and finally with water. The product was crystallized from a ethanol .

method of synthesis of (15)⁽²³⁾

To compound (10)(0.001 mole) dissolved in 20ml, ethanol. 2-aminopyridine (0.001mole) was added gradually. When addition was complete reaction mixture was reflex for 6 hr. After the reaction excesses of ethanol and aryl amine were recovered distillation. The residue was washed with sodium bicarbonate to remove the acid impurities and finally with water. The product was crystallized from ethanol .

Schemel



Results and Discussion

compound (1) N-(4-(dimethylamino)benzylidene)-1H-benzo[d]imidazol-2-amine

This compound was obtained as yellow solid yield 90.86%, $R_f = 0.87$, M.P (263) $^{\circ}\text{C}$.

The infrared spectrum data of compound (1) show absorption at (3055) cm^{-1} for (Ar-H), (3309) cm^{-1} (N-H), (1650)

cm^{-1} ($\text{C}=\text{N}$), and show new band at (2923) for ($\text{C}-\text{H}$) CH_3 .

The $^1\text{H-NMR(DMSO)}$ spectrum data of compound (1) show $\delta: 6.17-7.6$ (m, 8H, Ar-H), 9.21 (m, 1H, NH), 7.8 (m, 1H, CH), 3.6 (m, 6H, CH_3).

The $^{13}\text{C-NMR(DMSO)}$ spectrum data of compound (1) show $\delta: 158.6$ (C10), 156.7 (C2), 155.8 (C14), 154.8 (C8,C9), 133.1 (C12,C16), 124.07 (C9),

120.8(C5,C6) , 113.1(C13,C15) , 112.6 (C2,C5) ,22.9 (C17,C18) .

compound (2)

N-(furan-2-ylmethylene)-1H-benzo [d] imidazol-2-amine was obtained as brown solid yield 63% , R_f =0.39 , M.P(174)^oC. The infrared spectrum data of compound (2) show absorption at (3055) cm⁻¹ for (Ar-H),(3379) cm⁻¹ (N-H), (1681) cm⁻¹ (C=N),and show band at (1218) for (C-O-C).

compound (3)

4-(((1H-benzo[d]imidazol-2-yl) imino)methyl)-2-methoxyphenol was obtained as orang solid yield 35%,R_f =0.66 ,M.P (81)^oC. The infrared

spectrum data of compound (3)show absorption at (3062) cm⁻¹ for (Ar-H), (3363)cm⁻¹ (N-H) ,(1681)cm⁻¹ (C=N),and show band at (2877-2931) for (C-H)CH₃,(1234) cm⁻¹ for (C-O) C-OH ,(1272) cm⁻¹ (C-O) Ph-O-CH₃. The¹H-NMR(DMSO) spectrum data of compound (3) show δ:6.8-8.3(m , 7H , Ar-H) , 9.7 (m , 1H , NH) ,9.3 (m , 1H,CH),3.5(m,6H, CH₃) , 9.2(m,1H,OH). The¹³C-NMR(DMSO) spectrum data of compound (3) show δ:158.01 (C10) , 156.5 (C2) ,155.03(C14) , 153.6(C13) , 149.8(C9,C8),139.9(C11), 130.1(C5,C6) , 127.8 (C16),121.2 (C7,C4) , 117 (C15) , 112.1(C12), 66.5(C17) .

Table (1) infrared spectrum data for(1,2,3) compounds

Comp.	vAr-H arom.	vN-H	vC=N	vC-H aleph.	vOH	vArOCH ₃ CO
1	3055	3309	1650	2923	-	-
2	3055	3379	1681	-		
3	3062	3363	1681	2877-2931	3349	1257

compound (4)

3-(1H-benzo[d]imidazol-2-yl)-2-(4-(dimethylamino)phenyl)imidazolidin-4-one (4) as yellow solid, yield 65.7% ,R_f =0.8 , M.P(186)^oC.

The infrared spectrum data of compound (4)show absorption at (3050)cm⁻¹ for (Ar-H),(- 3150) cm⁻¹ (N-H),(- 1681)cm⁻¹(C=N),and show band at(2970)cm⁻¹ for (C-H)CH₃, and(1690) cm⁻¹ for(C=O) .

compound (5) 3-(1H-benzo[d]imidazol-2-yl)-2-(furan-2-yl)imidazolidin-4-one

as Purple solid, yield 65% ,R_f =0.56 , M.P(109)^oC.

The infrared spectrum data of compound (5)show absorption at (3070)cm⁻¹ for (Ar-

H),(3363) cm⁻¹ (N-H),(1635)cm⁻¹(C=N),(1712) cm⁻¹ for(C=O), and show band at (1218) for (C-O).

compound(6)

3-(1H-benzo[d] imidazol-2-yl)-2-(4-hydroxy-3-methoxyphenyl)imidazolidin-4-oneas Purple solid, yield 66.6% ,R_f =0.82 , M.P(102)^oC.

The infrared spectrum data of compound (6) show absorption at (3070) cm⁻¹ for (Ar-H),(- 3163) cm⁻¹ (N-H),(1681)cm⁻¹(C=N),and show band at (2970) for (C-H)CH₃,(1226) cm⁻¹ for (C-O) C-OH ,(1234) cm⁻¹ (C-O) Ph-O-CH₃),(1702) cm⁻¹ for(C=O).

Table (2) infrared spectrum data for imidazolidin-4-one derivatives compounds (4-6) cm⁻¹

Comp.	vAr-H	vN-H	vC=N	vC-H aleph.	vC=O	vOH	vArOCH ₃ CO
4	3050	3150	1681	2970	1690		
5	3070	3363	1635	-	1712		
6	3070	3163	1681	2970	1690	3346	1234

compound (7)

3-(1H-benzo[d]imidazol-2-yl)-2-(4-(dimethylamino)phenyl)thiazolidin-4-one as yellow solid, yield 76.6% ,Rf =0.65 , M.P(123)^oC.

The infrared spectrum data of compound (4) show absorption at (3070)cm⁻¹ for (Ar-H), (3332) cm⁻¹ (N-H), (1666)cm⁻¹(C=N), and show band at(2916)cm⁻¹ for (C-H)CH₃, and(1681) cm⁻¹ for(C=O) .

ompound (8) 3-(1H-benzo[d]imidazol-2-yl)-2-(furan-2-yl)imidazolidin-4-one

as yellow solid, yield 98% ,Rf =0.64 , M.P(170)^oC.

The infrared spectrum data of compound (8) show absorption at (3070)cm⁻¹ for (Ar-H),(3332) cm⁻¹ (N-H),(1650)cm⁻¹(C=N),(1712) cm⁻¹ for(C=O), and show band at (1218) for (C-O).

compound (9)

3-(1H-benzo[d] imidazol-2-yl)-2-(4-hydroxy-3-methoxyphenyl) thiazolidin-4-one as white serum, yield 76.6% ,Rf =0.7 , M.P.

The infrared spectrum data of compound (6) show absorption at (3070) cm⁻¹ for (Ar-H), (3379)cm⁻¹ (N-H) ,(1681)cm⁻¹(C=N),and show band at (2962) for (C-H)CH₃,(1208) cm⁻¹ for (C-O) C-OH ,(1026) cm⁻¹ (C-O) Ph-O-CH₃),(1712) cm⁻¹ for (C=O)and (3502)cm⁻¹ for(OH) .

The ¹H-NMR(DMSO) spectrum data of compound (9) show δ:6.7-7.3(m , 7H , Ar-H) , 9.4 (s , 1H , NH) ,3.7 (m , 1H,CH) , 3.4-3.6(s,2H,CH₂),3.1 (m,6H, CH₃) , 5.1(m,1H,OH).

Table(3)infrared spectrum data for imidazolidin-4-one derivatives compounds (7-9) cm⁻¹

Comp.	vAr-H	vN-H	vC=N	vC-H aleph.	vC=O	vOH	vArOCH ₃ CO	vC-O
7	3070	3332	1666	2916	1681	-	-	-
8	3070	3332	1650	-	1712	-	-	1026
9	3070	3379	1618	2962	1712	3502	1265	1026

compound (10)

N-(1H-benzo[d] imidazol-2-yl)-2-chloroacetamide as Green solid, yield 54.5 ,Rf =0.56 , M.P(198)^oC.

The infrared spectrum data of compound (10) show absorption at (3078) cm⁻¹ for (Ar-H),(1620) cm⁻¹ (C=N),and show new band at (2877-2993) for (C-H)CH₂ ,

(1689) cm⁻¹ for(C=O) and (740) Cm⁻¹ for (C-Cl),(3147)Cm⁻¹ for (N-H) , (1560) Cm⁻¹ for (C=C)

The¹H-NMR(DMSO) spectrum data of compound (10) show δ:6.2-7.4(m , 4H , Ar-H) , 11 (s , 1H , NH) , 3.35(m,2H,CH₂), 4.3(m,1H,NH).

The ^{13}C -NMR(DMSO) spectrum data of compound (10) show δ :148.5 (C10) , 136.8 (C2) , 123.04 (C9,C8) , 45.07(C11) , 121.9(C5,C6), 113.02(C4,C7) .

compound (11)

N5-(1H-benzo[d] imidazol-2-yl) oxazole-2,5-diamine as orang solid, yield =92% , Rf = 0.5 , M.P(266) $^{\circ}\text{C}$.

The infrared spectrum data of compound (11) show absorption at (3070) cm^{-1} for (Ar-H),(3379) cm^{-1} (N-H) overlap with absorption of NH₂ and show band at (1272)for C-O , (1650) cm^{-1} for (C=N) , (1604) cm^{-1} for (C=C)

The ^1H -NMR(DMSO) spectrum data of compound (11) show δ :6.3-7.4(m , 5H , Ar-H) , 11.1 (s , 1H , NH) , 5.5(m,2H,NH₂) , 4.1(m,1H,NH).

compound (12)

N5-(1H-benzo[d]imidazol-2-yl) thiazole -2,5-diamine as white solid, yield =72.7% ,Rf=0.57 , M.P(70) $^{\circ}\text{C}$.

The infrared spectrum data of compound (12) show absorption at (3070) cm^{-1} for (Ar-H),(3147) cm^{-1} (N-H),(3350) cm^{-1} for(N-H) for NH₂,(1666) cm^{-1} for (C=C) , (1681) cm^{-1} for (C=N)

compound(13)

N5N-(1H-benzo[d]imidazol-2-yl)-2-(2-carbamothioylhydrazinyl) acetamide as Brown solid, yield =67.79% ,Rf=0.54 , M.P(189) $^{\circ}\text{C}$.

The infrared spectrum data of compound (13) show absorption at (3070) cm^{-1} for (Ar-H),(3132) cm^{-1} for (N-H) imidazol and show new band at (3360) cm^{-1} for NH₂,(1257) for C=S , (2923) for (C-H)CH₂ , (1697)for (C=O) , ,(1620) cm^{-1} for (C=C) , (1681) cm^{-1} for (C=N)

compound(14)

2-((1H-benzo[d]imidazol-2-yl)amino)-N-(1H-benzo[d]imidazol-2-yl) acetamide as Green solid, yield =75.8% ,Rf =0.6 , M.P(162) $^{\circ}\text{C}$.

The infrared spectrum data of compound (14) show absorption at (3001) cm^{-1} for (Ar-H),(3147) cm^{-1} (N-H),and show new band at (1689) cm^{-1} for C=O,(1218) cm^{-1} for C-O , (2877-2939) cm^{-1} for (C-H)CH₂,(1620) cm^{-1} for (C=C) , (1635) cm^{-1} for (C=N)

compound(15)

N-(1H-benzo [d]imidazol-2-yl)-2-(pyridin-2-ylamino)acetamide as Brown solid, yield =70% ,Rf =0.34 , M.P(189) $^{\circ}\text{C}$.

The infrared spectrum data of compound (14) show absorption at (3078) cm^{-1} for (Ar-H),(3178) cm^{-1} (N-H)and show new band at (1689) cm^{-1} for C=O,(1226) cm^{-1} for C-O ,(2931) cm^{-1} for (C-H)CH₂,(1592) cm^{-1} for (C=C) , (1650) cm^{-1} for (C=N)).

Table(4)infrared spectrum data for(10-15) compounds cm^{-1}

Comp.	vAr-H	vN-H Imidazol	vC-H aleph.	vC=O	NH ₂	C=S	vC-Cl
10	3078	3147	2939	1689	-	-	740
11	3078		-	-	3379	-	-
12	3070	3147			3350		
13	3070	3132	2923		3360	1257	
14	3001	3147	2939	1689			
15	3078	3178	2931	1689			

Table(5):- Analytical and physical data of compounds .

No.	Molecular formula	Color	M.P°C	Yield%	R _f
1	C ₁₆ H ₁₆ N ₄ (264.325)	yellow	263	90.86	0.87
2	C ₁₂ H ₉ N ₃ O (211.219)	Brown	174	63	0.39
3	C ₁₅ H ₁₃ N ₃ O ₂ (267.283)	orang	81	35%	0.66
4	C ₁₈ H ₁₉ N ₅ O (321.37)	yellow	186	65.7	0.8
5	C ₁₄ H ₁₂ N ₄ O ₂ (268.807)	purple	109	65	0.56
6	C ₁₇ H ₁₆ N ₄ O ₃ (324.334)	purple	102	66.6	0.82
7	C ₁₈ H ₁₈ N ₄ OS (338.42)	yellow	123	76.6	0.65
8	C ₁₄ H ₁₁ N ₃ O ₂ S (285.32)	yellow	170	98	0.64
9	C ₁₇ H ₁₅ N ₃ O ₃ S (341.38)	white	seram	76.6	0.7
10	C ₉ H ₈ N ₃ OCl (209.632)	Green	198	54.5	0.56
11	C ₁₀ H ₉ N ₅ O (215.211)	orang	266	92	0.5
12	C ₁₀ H ₉ N ₅ S (231.277)	white	70	72.7	0.57
13	C ₁₀ H ₁₂ N ₆ OS (264.079)	Brown	seram	67.79	0.54
14	C ₁₆ H ₁₄ N ₆ O (306.322)	Green	162	75.8	0.6
15	C ₁₄ H ₁₃ N ₅ O (267.286)	Brown	189	70	0.34

Reference

- 1-R. Wali, K. Dhamij, Vandan, Md J. Akhtar and HS. Lamba; *IJP CBS.*; 2012, **2(3)**, 293-298.
- 2- Davinder Kumar and keshav sharma ; *IJPR, July.*; 2012,V.3, Issue-2.
- 3- BNB. Vaidehi, K. Gnana Deepika, RV. Satya, RR. Bangaramma, R. Harish Kumar, Y. Ratna Sudha and T. Ravi Kumar , *IJR PC.*; 2012, **2(2)** .

- 4- Butrus H. Nabeel and Saeed T. Farah ; *Res.J.Chem.Sci.,June.*; 2012, **2(6)**, 43-49.
- 5- Mehdi Forouzani and Hassan Ghasemnejad-Bosra ; *J. Chem.*; 2012, **9(3)**, 1064-1069 .
- 6- Vishvanath D. Patil1, Gole Medha, Mhatre Shramesha, Jaiswal Aarti ; *Der Chem. Sinica.*; 2010, **1(2)**, 125-129.
- 7- Simone Budow, Mariola Kozlowska, Agata Gorska, Zygmunt Kazimierczuk, Henning Eickmeier, Paolo La CollaGilles Gosselin, and Frank Seela; *ARKIVOC.*; 2009, **(iii)**, 225-250.
- 8- A. Jamal Abdul Nasser, A.Idhayadhull, R.surendra kumar, and J.SELVIN ; *J. Chem.*; 2010, **7(4)**, 1320-1325 .
- 9- Maira GR Pitta, Andréa CA Silva, Juliana Kelle AL Neves, Poliana G Silva, João I Irmão, Elizabeth Malagueño, José V Santana, Maria CA Lima, Suely L Galdino, Ivan R Pitta, Mônica CPA Albuquerque, *Mem Inst Oswaldo Cruz, Rio de Janeiro.*; 2006, **101(Suppl. I)**, 313-316.
- 10- Nuno Vale, Miguel Prudencio, Catarina A. Marques, Margaret S. Collins ,) Jiri Gut, F atima Nogueira, Joana Matos, Philip J. Rosenthal, Melanie T. Cushion, Virgi'lio E. do Ros_ario, Maria M. Mota, Rui Moreira,3 and Paula Gomes, *J. Med. Chem.*; 2009, **52**, 7800 –7807.
- 11- G. Omprakash, Y. Anjaneyulu, N. Siva Subramanian, M.ramadevi and G. vijayakshmi ; *Int. J. Chem. Sci.*;2010, **8(2)**,783-790.
- 12- Rekhas S , Shantharam U , Rekhas S.et al *.IRJP.*; 2011,**2(9)**,81-84.
- 13- P. Sudhir Kumar, Debasis Mishr, Goutam Ghos and Chandra S. Panda ; *Rasayan, J. Chem.*; 2010,**3**, (3), 600-606.
- 14- Cherkupally Sanjeeva Reddy, Gaddam Rajesh Kumar,Macherla Vani Devil and Adki Nagaraj ; *Acta Chim. Slov.*; 2011,**58**, 576–581.
- 15-Arvind k. Singh, Geeta Mishra and Kshitiz Jyoti ; *J.APS.*;2011,**01(05)**, 44-49.
- 16-Balvant S. Singh, Hyacintha R. Lobo, Dipak V. Pinjari, Krishna J. Jarag a, Aniruddha B. Pandit , Ganapati S. Shankarling ; *Ultrasonics Sonochem.*;2013,**20**, 633–639.
- 17- Danilo Davyt and Gloria ; *Mar. Drugs.*; 2010),**8**,2755-2780 .
- 18- Ibrahim Evren Kırız, Yusuf Sert, Mustafa Saçmacı , Ertan Sahin, Ismail Yıldırım, Fatih Ucun ; *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy.*;2013,**114**, 491–501.
- 19- Andrew J. Phillips,Yoshikazu Uto,Peter Wipf,Michael J. Reno, and David R. Williams; *Org. Lett.*; 2000 , **2(8)**.
- 20- Panneer Selvam,T, P. P. Radhika1, S. Janagaraj1,A. Siva Kumar; *Researchin Biotchnology , J.chem.*; 2011,**2 (3)**,50-57.
- 21- Shaaban K. Mohamed, A. A. Abdelhamida, Walid Omarac, Abdel- Aal M. Jabere and Mustafa Albayatif , *JCPRC.*; 2013,**5(1)**19-31.
- 22- Slobodan Sukdolak, Slavica Soluji, Nenad Vukovl, Nedeljko Manojlov and Ljubomir Krsti; *J.Serb. Chem.Soc.*;2004,**69(5)**,319–326.
- 23- Nazar Trotsko, Maria Doboszi and Ewajagieo-W'jtoowicz , *Acta Poloniae Pharmaceutican Drug Research.*;2007,64(3), 227n231