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The Effect of Ceo₂ Addition on Transformation Temperatures and Wear Resistance of Cu-Al-Ni Shape Memory Alloys

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HIGHLIGHTS

- Enhance Transformations Temperature for SMAs by adding CeO₂ particles with different percentages.
- Enhance Wear behavior for SMAs BY adding CeO₂ particles with different percentages.
- Enhance Hardness for SMAs BY adding CeO₂ particles with different percentages.

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ABSTRACT

SMAs can switch from one crystallographic structure to another in response to temperature or stress stimuli. When SMAs are exposed to mechanical cyclic stress, they can absorb and discharge mechanical energy by experiencing a reversible hysteretic shape change. SMAs are widely used for sensing, actuation, impact absorption, and vibration damping. This work studied the effect of CeO₂ addition on the transformation temperature and wear resistance of Cu-Al-Ni SMAs. WhereCeO₂ was added at different percent's 0.5, 1, and 3 wt% to the base alloy, followed by casting and homogenization at 900°C. Some tests were carried out: Differential scanning calorimeter, Optical Microscope, Scanning Electron microscopy, Energy dispersion spectrometer, X-Ray Diffraction, and Wear and Hardness tests. OM and SEM tests reveal that both phases of martensite β and γ are found. Also, the additions of CeO₂ show a visible effect on phase formation and transformation temperatures. It was observed that increasing of CeO2 particles in Cu-based SMAs owing to improve interfacial bonding between matrix and reinforcement and also observed that the variants become thicker with increasing in percent. Additions of different percentages of cerium oxide increase the hardness of Cu-Al-Ni SMAs. Due to the addition of CeO2 particles, the sample's wear rate decreases compared to pure SMAs.

1. Introduction

The shape memory alloys (SMAs) can remember and restore their previous shape after substantial inelastic deformation through a phase transformation mechanism. The shape memory effect is the name for SMA's unique characteristic (SME). SMA phase change may be induced by either sending an electrical current through the structure or increasing the structure's temperature over its phase transformation temperature with an external heat source [1]. Cu-based SMAs are in high demand because of their low cost and excellent mechanical and shape memory capabilities. In various applications, copper-based alloys are an excellent replacement for Ni-Ti alloys, such as micro-electro-mechanical systems, valves, actuators, and dampers, since they are inexpensive and easy to manufacture and have a wide temperature range [2]. Cu-based alloys with the characteristics of high tensile strength and hardness require extensive research and development. Age hardening or adding dispersed particles to copper's matrix can significantly improve its mechanical strength. At high temperatures, age-hardenable copper alloys are prone to precipitate coarsening, significantly reducing their strength. This way, even when exposed to high temperatures, dispersionstrengthened copper preserves most of its properties. Insoluble in the copper matrix, dispersed particles including oxides, carbides, and borides are thermally stable at high temperatures. Dispersion-strengthened copper alloys that have been dispersionstrengthened are used to make copper-based matrix composites [3-6]. Wang et al. [7] had shown some oxides can exhibit exceptional shape memory effects and superelasticity at nano/microscales, as evidenced by recent development in shape memory oxide studies connected to small-scale approaches such as lowering grain boundaries, strain engineering, and use in the form of nanoscale thin films. The shape memory efficiency of the materials systems ZrO₂, BiFO₃, and VO₂ in bulk and small-scale is reviewed. Pandey et al. [8] investigated the effect of adding nano CeO2 to a well-known Cu-Al-Ni shape memory alloy as a grain pinner/refiner. They discovered a very uniform dispersion of CeO2 and that increasing CeO2 resulted in a superior martensitic structure. The inclusion of CeO₂ substantially impacts transition temperature and phase formation. In addition,

Pandey et al. [9] studied the addition of nano-CeO₂ and modified the morphologies of microstructures significantly. 0.1% CeO₂ addition was reported to have the highest mechanical (fracture stress=536 MPa and fracture strain=7.67%) and shape memory properties (95% shape recovery). The addition beyond this was detrimental to mechanical and shape memory properties. In one of the studies, Abolhasani et al. [10] reported employing a new way to boost the shape memory effect and its stability in Cu-Al-Ni SMAs by selective laser melting and partial reinforcement with alumina (Al₂O₃). Partially reinforced sections with 0.3 wt% and 0.9 wt% Al₂O₃ are compared to fully reinforced parts to define the effect of partial reinforcement. The shape memory effect is improved by enhancing mechanical anisotropy in the building direction, resulting in the buildup of localized residual compressive stresses above the reinforcing lines in the neutral plane, generated during selective laser melting. Furthermore, a unique phenomenon known as the "waistband effect" is seen in cyclic compressive operation, which results in incremental penetration of the reinforcing particles into the neutral plane during compression. These findings show that partial reinforcement with Al₂O₃ improves the shape memory effect of Cu-Al-Ni SMAs substantially. Xie Dong et al,[11] studied TiO₂ coatings with thicknesses of 100 and 50 nm on NiTi shape memory alloys were created using high-power impulse magnetron sputtering (SMA-NiTi). Under uniaxial strain, the NiTi substrate transitioned from B_2 to B_{19} martensitic phase. Surface relief appeared on the film after stress relaxation, but residual deformation in the NiTi persisted. After heating, a reverse B_{19}' to B_2 martensitic phase transition occurred on a Ni-Ti substrate, removing surface relief and residual deformation. The 50-nm film showed no breaking or spalling after tensile deformation at a maximum strain of 10%, and the shape was recovered by heating. In a similar line, the current work investigates the microstructure, hardness, phase precipitation, and transition temperature of-14 wt pct Al-4.5 wt pct Ni when various quantities of (CeO₂) Micro particles are introduced to CuAlNi SMAs. CuAlNi SMA was studied using X-ray diffraction (XRD), differential scanning calorimetry (DSC), scanning electron microscopy (SEM), optical microscopy (OM), as well as hardness and wear testing.

2. Experimental Study

2.1 Preparation of the Sample

Three alloys with varying CeO₂ compositions were created in a gas furnace at 1400°C with a protective environment of (Ar) Gas, as shown in Table1. To make the ingots, pure metals such as Cu (99.99 percent), Al (99.999 percent), Ni (99.9 percent), and CeO₂ (99.95 percent) were wrapped in Al foils and placed in a silicon carbide crucible. The produced ingots are cylindrical, with a diameter of 14 mm and a height of 117 mm. Homogenization was performed on all ingots in a Carbolite1500, England furnace at 1173 K (900 °C) for 30 minutes, followed by iced water quenching. A wire cut device EDM was also used to divide the samples into two groups of specimens. For microstructure testing, the first one has a 14mm diameter and a 5mm length. For the wear test, the second was machined by wire cutting to 14mm diameter and 30mm height.

Table 1:	Specimen	coding and	composition	of the materials
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Sample Composition	Sample Code	
Cu-14wt pctAl-4.5wt pct Ni	С	
Cu-14wt %Al-4.5wt%Ni +0.5 wt % CeO2	C1	
Cu-14wt %Al-4.5wt%Ni +1 wt % CeO2	C2	
Cu-14wt %Al-4.5wt%Ni +3 wt % CeO2	C3	

2.2 Xrd and Microstructure Tests

X-Ray Diffraction testing was used to detect the phases and compounds that precipitated in all of the alloys generated. The inspections were carried out with (ASCII-2Theta-Intensity (ASC)), (K-1=1.54060), and the supplied voltage and current were 40 KV and 30 mA, respectively. In addition, SEM was also utilized to evaluate the surface morphology and microstructure of the samples (SEM Tescan Mira3 XMU, Czech Republic).

2.3 Dsc Test

Differential scanning calorimetry is commonly used to examine the thermal properties of SMAs. A mechanical saw was used to cut the specimen, which was then placed in a tiny crucible weighing 100mg and heated for 10 minutes at 400°C. The DSC was carried out in the Materials Engineering department DSC using (DSC131EVO-SETARAM DSC Device). Also, the equilibrium temperature is calculated from Equation 1. According to Salzbrenner and Cohen, the equilibrium temperature is characterized by the following relationship [12]:

$$T^{0} = 0.5 (M_{s} + A_{f})$$
 (1)

$$Hysteresis = A_s - M_f$$
(2)

$$Spread = A_f - A_s \tag{3}$$

The martensite starts temperature is M_s , while the austenite finishing temperature is A_f . During heating and cooling cycles, the endothermic and exothermic peaks acquired from DSC data were combined as a function of temperature. Hysteresis from Equation 2 [13], and spread from Equation 3 [13].

2.4 Hardness and Wear Tests

All samples were subjected to a Vickers micro-hardness test using an HVS100 Microhardness tester with a force of 2.94N and a holding duration of 15 seconds. Because there may have been holes on the sample's surface that would alter the hardness assessment, two readings were collected for each sample. In the wear test, pin-on-disc equipment was utilized to examine the wear attributes of the composite. Each composite sample was put through a wear test with a continuous load of 15 N at a constant speed of 1 m/s for 5, 10, and 15 minutes at a constant sliding distance of 62.5 m using 180 mesh Al_2O_3 paper. The wear pin was cleaned in acetone and dried before being weighed on a micro-balance with a 0.1 mg sensitivity before wear testing. Adhesive wear equations are given by the following expressions:

$$V = K W L / H_V$$
(4)

W represents weight, L is the sliding distance, and K stands for wear coefficient, which is determined by the plasticity index, surface roughness, and friction coefficient [14].

3. Results and Discussion

A composite reinforcement's content and size are directly related to its microstructure and mechanical characteristics.

3.1 Dsc Test

Cerium dioxide additions for the three alloys produce well-defined peaks once more. The CeO₂+Base SMA transition temperature and equilibrium temperature findings are shown in Table 2. The kind and shape of the martensite phase and the order and stability at a degree of γ ' have a role in the transition temperature fluctuation. The forward (cooling) and reverse (heating) martensitic transformations are confirmed by the peaks in Figures 1(A, B, C, and D). The transformation went to a higher temperature for each sample for the three alloys. Thermal cycling up to 200°C is well recognized for causing changes in alloy structure and composition and physical and mechanical characteristics[15]. After a suitably long aging period at 200°C, an increase in the critical transition temperatures associated with a conversion of γ ' into β ' martensite was also identified. [16]. The DSC thermograms for the base alloy show transformation temperature in the domain between(-100-1700C), also shown in [17]. In terms of equilibrium temperature, the greatest value of T° was recorded at C₃ alloy with a 3 percent addition, which is connected to martensite phase precipitation and shape. Adding 3% CeO₂ to Mf boosts Mf to 171, according to Table 2. This is because of the quick decline in transformation temperature as grain size increases with CeO₂ concentration.

Table 2: Transition temperature for SMA + CeO₂ alloys

Alloy	As °C	Af °C	Ms°C	Mf°C	T°C	H°C	S°C	
С	130	165	125	104	145	26	35	
C_1	376	384	166	101	275	274	8	
C_2	285	356	172	141	264	144	70	
C3	356	394	210	171	302	184	38	



(A)



(B)







Figure 1: Displayed DSC Test for A. base alloy B. addition of 0.5 CeO_2 to the base alloy C . addition of $1\% \text{ CeO}_2$ to the base D .addition of $3\% \text{ CeO}_2$ to the base

3.2 X-Ray Diffraction (XRD) Analysis

Figure 2 A shows the XRD spectrum of Cu-Al-Ni materials with ratios Al (14%), Ni 4.5%, and Cu's rest ratio. The diffraction peaks correspond to the polycrystalline structure. The spectrum curve showed sharp and very fine peaks indicating good structure crystallization. The based alloy SMA for (Cu-Al-Ni) observed that the formation of both phases indicates the form of the both martensitic Cu-Al (β ') at angles (42°, 44°, and 64°) and Cu-Al-Ni (γ ') at angles (24°, 46°, and 74°) respectively. Figure (2) B illustrate the XRD charts for the SMA (Cu-Al-Ni) modified with 0.5% CeO₂ is found at angles (22°, 31°, and 51°). Also, the (β ') phase Cu-Al (41°, 43°, and 61°) and Cu-Al-Ni (γ ') are found at angles (25° and 75°) phase indices. Figure C illustrates the XRD charts for the SMA (Cu-Al-Ni) modified with 1% CeO₂ found at angles (31°, 39°, 49°, and 51°). Also, the (β ') phase Cu-Al (42°, 44°, and 64°) and Cu-Al-Ni (γ ') are found at angles (27° and 77°) phases, respectively. Finally, D showed the XRD show charts for the SMA (Cu-Al-Ni) modified with 3% CeO₂ found at angles (51°). Also, the (β ') phase Cu-Al (41° and 43°) and Cu-Al-Ni (γ ') are found at angles (51°). Also, the (β ') phase Cu-Al (41° and 43°) and Cu-Al-Ni (γ ') are found at angles (51°). Also, the (β ') phase Cu-Al (41° and 43°) and Cu-Al-Ni (γ ') are found at angles (51°). Also, the (β ') phase Cu-Al (41° and 43°) and Cu-Al-Ni (γ ') are found at angles (51°).





Figure 2: Displayed XRD Test for A .base alloy B. addition of 0.5 CeO₂ to the base alloy C. addition of 1% CeO₂ to the base D .addition of 3% CeO₂ to the base

3.3 Microstructure

The morphology of both martensitic Cu-Al(β) and Cu-Al-Ni(γ) is shown in Fig.3.A. These phases are β and γ , which are thermally generated martensites with different thicknesses and orientations. Between the γ phases, the γ phase developed as a plate-like phase, while the β phase formed as a needle-like phase. Figure 3B shows the morphology of martensite for the SMA (Cu-Al-Ni) modified with 0.5 percent CeO₂ discovered in the inhomogeneous distribution in the matrix, which gave both phases a fine grain size. Both phases suggest the form of both martensitic Cu-Al(β) and Cu-Al-Ni (γ). Figure3.C and D also showed the morphology of martensite phases β and γ . Also, in the figures b, c, and d displed as the reinforcing content increased from 0.5 to 3%, it was discovered that martensite variations differed in size and thickness. The variants become thicker, as observed at Abohasani[9] when they added alumina to CuAlNi SMAs. Figures 4 B, C, and D showed SEM images, which also confirm [9] and reveal that the increase of CeO₂ content increases the hardness of the samples. Figure 5 A, B, C, and D showed the EDS test for all the samples and the content of the elements.



Figure 3: Displayed OM Test for A. base alloy B. addition of 0.5 CeO2 to the base alloy C. addition of 1% CeO2 to the base D. addition of 3% CeO2 to the base



Figure 4: Displayed SEM test for A.base alloy B. addition of 0.5 CeO₂ to the base alloy C. addition of 1% CeO₂ to the base D .addition of 3% CeO₂ to the base

3.4 Hardness and Wear Test

The average of two measurements was used to calculate microhardness, as shown in Figure 6. The hardness rose by adding 0.5 percent, with a measurement of 475.9Hv compared to the basic SMA. However, with additional addition, the hardness decreased, as shown in the table, at a percentage of 1% weight percent CeO₂. The hardness was also observed to increase when adding 3% weight percent CeO₂. This is a little difference in hardness in certain cases, which may be attributable to the presence of various phases in the samples. This confirms what we discovered in [8]. The wear rates of the composite were lower than those of pure Cu-Al-Ni at different periods, as indicated in Figure 7. An increase in wear volume with increasing time at constant load may be seen in the graph. When CeO₂ 0.5 weight percent is added to the base SMAs, the wear rate of the sample decreases compared to pure base SMA. This is due to the integration of CeO₂ particles, which increases the hardness of the specimen. Also, observe that the weight loss for adding 0.5 weight percent, 3 weight percent CeO₂ is slightly different from the previous additions.



Figure 5: Displayed EDS Test for A .base alloy B. C_1 , C . C_2 , D C_3



Figure 6: Variation of Hardness with different CeO₂ additions



Figure 7: Variation of wear Volume rate with different CeO₂ additions

4. Conclusion

This article examined the effect of adding CeO₂. The following conclusions can be taken from the report results:

- 1) From transition temperature for SMA modified with CeO_2 addition, there is an increase in Ms and Mf and hysteresis and spread. Also, the greatest value of T^0 was found in the C_3 alloy, which is attributed to martensite phase precipitation and shape.
- 2) From SEM and OM images, both phases denote the martensitic Cu-Al (β) and Cu-Al-Ni(γ) phases, respectively. These phases are β ' and γ ', produced as thermally induced martensites with varying thicknesses and orientations. It was observed that increasing CeO₂ particles in bronze matrix owing to improved interfacial bonding between matrix and reinforcement and that the variants become thicker with increasing in percent.
- 3) From the hardness test, we can see that the additions of different percent of cerium oxide increase the hardness of Cu-Al-Ni SMAs, Where the highest hardness was in sample C₃. Furthermore, from the wear test, we can see that when CeO2 was 0.5 added to the base SMA, the wear rate of the sample decreased compared to pure SMA. This is due to the addition of CeO₂ particles, which decrease the wear rate of the specimen.

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Author contribution

All authors contributed equally to this work.

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Data availability statement

The data that support the findings of this study are available on request from the corresponding author.

Conflicts of interest

The authors declare that there is no conflict of interest.

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