Inter-Band Transitions and Urbach Tails of Chlorine Doped Tin Monosulfide

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Abstract

Thin films of SnS and chlorine doped SnS were prepared utilizing chemical spray pyrolysis.

The electronic transition was found to be direct and indirect for all the samples under investigation. The Urbach tail values were increased as the doping percentage increase while the energy gap takes the behavior of reverse. The single oscillator model proposed by Wimple and Didomoico were applied to the obtained sample which certainly obey single oscillator model.

الانتقالات الحزمية وذيول اورباخ لأغشية كبريتيد القصدير المشوبة بالفلور

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الخلاصة

حضرت أغشية رقيقة من كبريتيد القصدير وكبريتيد القصدير المشوب بالكلور باستخدام التحلل الكيميائي الحراري.

لقد وجد بان الانتقالات الالكترونية هي من النوع المباشر وغير المباشر ولكافة النماذج قيد الدر اسة.

ان قيم ذيول اورباخ تزداد بازدياد نسبة التشويب بينما تسلك فجوة الطاقة سلوك عكسي, طبق نموذج المتذبذب الاحادي المقترح من قبل Didomenico, Wemple على كافة النماذج المحضرة, لقد وجد بان كافة النتائج تتفق وبشكل مؤكد مع نموذج المتنبذب الاحادي.

Introduction

Recently much attention has been paid for ZnS thin film, since it is a promising material for low cost photovoltaic conversion of solar cell, its conductivity is controllable by using various dopants such as Al, Ag, and Cl, tin sulfide exists in a variety of phases such as SnS, Sn₂S₃,SnS₄ and SnS₂, due to bonding characteristics of tin and sulfur, it can have p or n type conduction (an excess of tin changes the type of conductivity from p to n type). SnS is reported to have a direct band gap and indirect band gap depends on the conditions of preparation, according to Prince-Loferski diagrams, a high conversion efficiency of about 25% can be obtained using SnS in photovoltaic devices, in addition it is also useful as heated mirrors in solar control coatings ^[1-6].

Sns thin films could be prepared using different techniques such as chemical deposition, thermal evaporation, chemical bath deposition, co-evaporation plasma-enhanced, chemical vapor deposition and spray pyrolysis ^[7-12].

In the present work we show the results of the effect of chlorine doping on the inter-band transition of SnS and the results of the dispersion parameters deduced from wimple and Didomenico theory.

Experimental details

The optical energy band gap (E_g^{opt}) and the type of optical transition was analyzed using the equation foe the near-edge absorption proposed by Bardeen et al ^[13]:

 $(\alpha h \upsilon)^{r} = A (h \upsilon - E_{g}^{opt}) \qquad (1)$

Where α is the absorption coefficient, hv is the photon energy, E_g^{opt} is the optical energy gap, A constant and r is characterizes the transition process we can see that r = 2 for direct allowed and r=1/2 for indiect forbidden transition.

Fig. (1) Shows the curves of $(\alpha h \upsilon)^2$ versus (h υ) for SnS and SnS:Cl. The curve has a good straight line fit with higher energy rang above the absorption edge. The direct energy gap E_g have been calculated as 1.35 eV for pure and (1.32, 1.30, 1.27) eV for doped SnS with 1%, 2%, 3% chlorine respectively the value of pure SnS is in good agreement with the reported data on single crystal and polycrystalline films grown by other techniques ^[14-17].

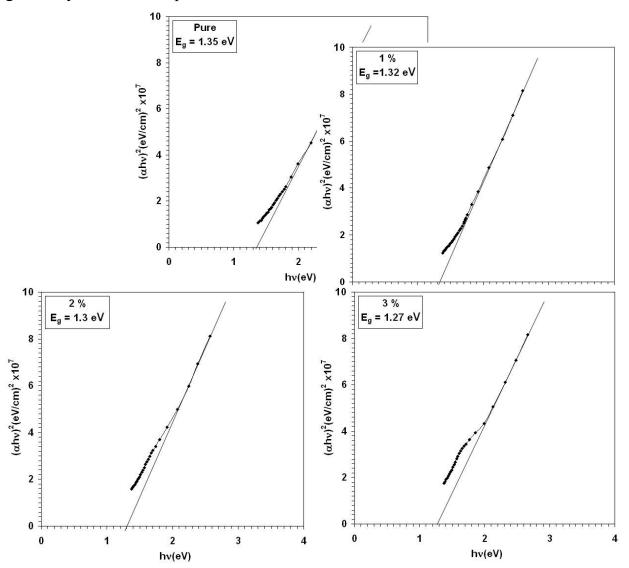
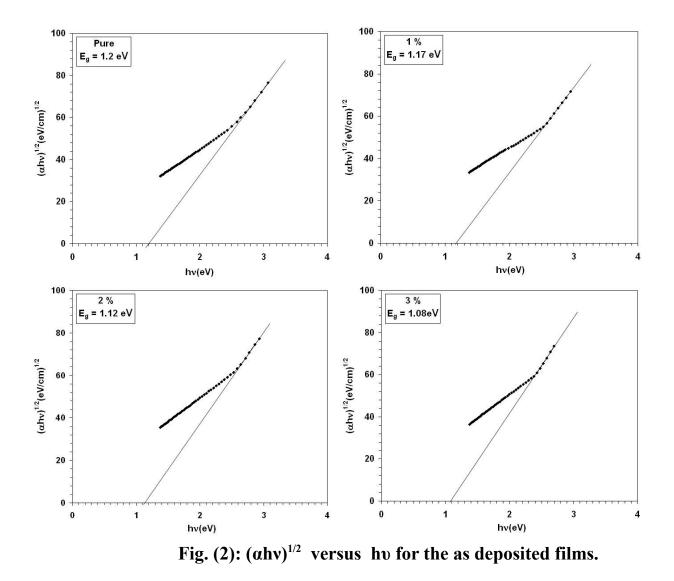


Fig. (1): $(\alpha h\nu)^2$ versus $h\nu$ for the as deposited films. Fig. (2) Illustrate the dependence of $(\alpha h\nu)^2$ versus of photon energy($h\nu$).

It can be shown from the fig, that the value of undoped SnS is 1.2 eV, which is in good agreement with the literature ^[4.18].



It is known that the absorption coefficient near the band edge shows an exponential dependence on photon energy described by the following relation ^[19]:

$$\alpha = \alpha_{o} \exp\left(\frac{hv}{E_{U}}\right) \dots (2)$$

Where (α_o) is constant, (hv) is the photon energy, and E_U is the Urbach energy which corresponds to the width of the band tail and can be evaluated as the width of the localized state. Thus the plot of ln α versus hv should be linear and Urbach can be obtained from the slope. Fig (3) illustrates ln α versus hv, Urbach energy was calculated from the reciprocal gradient of the linear portion of these curves, the width of band tail i.e Urbach energy increases as the chlorine percentage increase (370-500) meV.

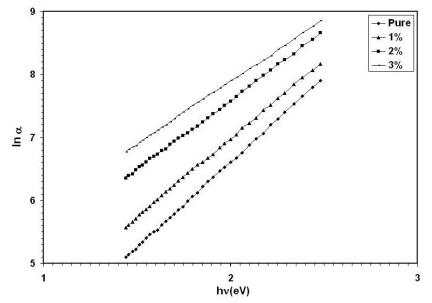


Fig. (3): Illustrate ln α versus hv.

Sampla	Ed	Eo	11	$oldsymbol{\mathcal{E}}_{_{\infty}}$	M ₋₁	M-3	S ₀ x10 ¹³	λο	Eu
Sample	(eV)	(eV)	n _{lwl}	$\boldsymbol{\upsilon}_{\infty}$		(eV) ⁻²	m ⁻²	nm	meV
Pure	9.87	3.6	1.93	3.725	2.74	0.211	3.12	290	370
1%	10.87	3.48	2.03	4.121	3.12	0.257	3.38	304	400
2%	11.18	3.22	2.11	4.452	3.47	0.334	3.42	307	454
3%	12.3	3.198	2.20	4.840	3.84	0.375	3.75	320	500

Table (1) the optical parameters of PMMA and copper sulfate doped PMMA.

Urbach energy values change inversely with the optical band gap of films, the decrease in E_g is attributed to the increase of disorder of the material occurred by doping.

This increase leads to a redistribution of states from band to tail, thus allows for a greater number of possible bands to tail and tail to tail transitions ^[20].

The refractive index is a significant factor in optical communication and in designing devices for spectral dispersion and the refractive index dispersion data below the inter- band absorption edge are important for technological applications of the optical materials.

The dispersion data of the refractive index can be described by a single-oscillator model ^[21]:

$$n^{2} = 1 + \frac{E_{o}E_{d}}{E_{o}^{2} - (hv)^{2}}$$
 (3) ------

Where n is the refractive index, E_o is oscillator excitation energy for electronic transitions and E_d is the dispersion energy

Which is the measure of the strength inter-band optical transition, this model describes the dielectric response of transitions below the optical gap.

The values of E_o and E_d is for inter-band optical transitions are calculated from Fig. (4) for all the samples under investigation and their values were tabulated in table (1) from these values, the long wavelength limit of refractive index n_{twt} can be determined using the expression:

$$n_{1wl} = \sqrt{1 + \frac{E_d}{E}}$$
 (4) -----

Their values were also tabulated in table (1) these values of the refractive index are within the rang found (1.5-4.6) in the literure ^[22-24], the value of the static dielectric constant $\varepsilon_{\infty} = n_{_{bd}}^2$ was calculated and their value was listed in Table (1), as can be seen that ε_{∞} increases as the doping percentage increases.

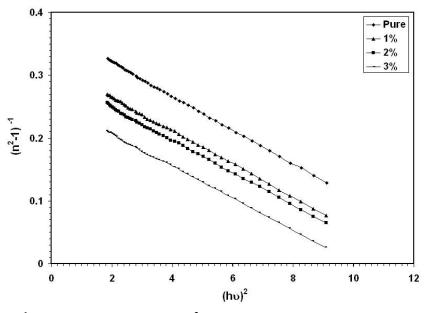


Fig. (4) $(n^2 - 1)^{-1}$ as a function of $(hv)^2$.

The single-oscillator parameters E_o and E_d is related to the imaginary of ε_i of the complex dielectric constant. The ε_i parameter includes the_desired response information about electronic and optical properties of the optical material thus, the fact that the moments of ε_i spectrum are determined is very important for the optical application of the optical material studied. The M₋₁ and M₋₃ moments of the optical spectrum can be obtained from the following relations ^[25]:

$$E_{o}^{2} = \frac{M_{-1}}{M_{-3}} \quad (5) -----$$
$$E_{d}^{2} = \frac{M_{-1}^{3}}{M_{-3}} \quad (6) ------$$

Their values were listed in table (1) according to the single oscillator model proposed by DiDomenico and wimple ^[26]:

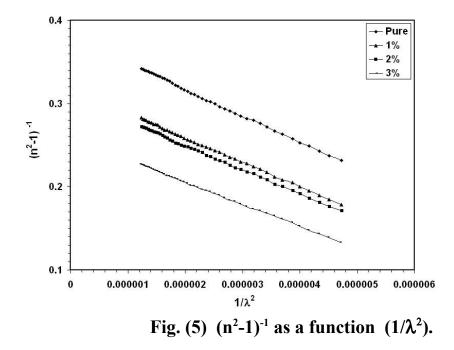
$$n^{2} - 1 = \frac{S_{o}\lambda_{o}^{2}}{1 - (\lambda_{o}/\lambda)^{2}}$$
 (7) ------

Where λ_{\circ} is an average oscillator wavelength, S_{\circ} is the average oscillator strength and λ_{\circ} is the wavelength of the incident light, the values S_{\circ} and λ_{\circ} were tabulated in table (1).

The linear relation between $(n^2-1)^{-1}$ and $(1/\lambda^2)$ is expected from equation (7).

 $(n^2-1)^{-1}$ of the as deposited thin films as a function of $(1/\lambda^2)$ is shown in the inset of

Fig. (5) Proves the validity of the above model, the values were listed in Table (1).



Conclusions

SnS and SnS:Cl which were prepared by chemical spray pyrolysis shows direct and indirect transitions.

The energy gap for both transitions were decreasing as the doping percentage increase, while the Urbach tail vales were increased by doping.

We conclude also that the refractive index dispersion curve obey the single oscillator model.

<u>References</u>

- [1] B. Ghosh, M. Das, P. Banerjee, and S. Das, Fabrication of Vacuum Evaporated SnS/CdS Heterojunction for PV Applications, Solar Energy Material & Solar Cells, Vol. 92 (2008) 1099-1104.
- [2] E. Guneri, C. Ulutas, F. Kimizigul, G. Altindemir, F. Code and C. Gumus, Effect of Deposition Time on Structural, Electrical and Optical Proerties of SnS Thin Films Deposited by Chemical Bath Deposition, Applied Surface Science, Vol. 257 (2010) 1189-1195.
- [3] B. G. Jeyaprakash, R. Ashok Kumar, K. Kesavan and A. Amalarani, Structural and Optical Characterization of Spray Deposited SnS Thin Films,
 - Journal of American Science, Vol. 6 (3) (2010) 22-26.
- [4] A. Tanusevski and D. Poelman, Optical and Photoconductive Properties of SnS Thin Films Prepared by Electron Beam Evaporation, Solar Energy

Materials & Solar Cells, Vol. 80 (2003) 297-303.

[5] K. T. R. Reddy, P.P. Reddy, P. K. Dalta and R. W. Miles, Formation of Polycrystalline SnS Layers by a Two Step Process, Thin Solid Films, Vol.

- [6] K. T. R. Reddy and P.P. Reddy, Structural Studies on SnS Films Grown by A Two Stage Process, Materials Letters, Vol. 56 (2002) 108-111.
- [7] D. Avellaneda, G. Delgado, M. T. S. Nair and P. K Nair, Structural and Chemical Transformations In SnS Thin Films Used in Chemically

[8] Deposited Photovoltaic Cells, Thin Solid Films, Vol. 515 (200) 5771-5776.
[8] M. Devika, N. K. Reddy, K. Ramesh, K. R. Gumnaselhar, , E. S. R. Gopal and K. T. R. Reddy, Influence of Annealing on Physical Properties of Evaporated SnS Films, Semicond. Sci. Technol., Vol. 21 (2006) 1125-1131.
[9] E. Turan, M. Kul, A. S. Aybek and M. Zor, Structure and Properties of SnS

Semiconductor Films Produced by Chemical Bath Deposition, J. Phys. D:

^{403-404 (2002) 116-119.}

[10]	Appl. Phys., Vol. 42 (2009) 245408 (6pp) C. Cifuentes, M. Botero, C. Calderon and G. Gordillo, Optical and
[10]	Structural Studies on SnS Films Grown By Co-Evaporation, Brazilian
[11]	Journal of Physics, Vol. 36(3B) (2006) 1046-1049. A. S. Jurez, A. T. Silver and A. Ortiz, Fabrication Of SnS _{2b} /Sns
	Heterojunction Thin Films Diodes By Plasma-Enhanced Chemical Vapor
[12]	Deposition, Thin Solid Films, Vol. 480-481 (2005) 452-456. M. C. Rodriguez, H. Martinez, A. S. Juarez, J. C. Alvarez, A. T. Silver and
	N. E. Calixto, Structural, Optical and Electrical Properties of Tin Sulfide
	Thin Films Grown by Spray Pyrolysis, Thin Solid Films, Vol. 517 (2009)
	2497-2499.
[13]	G. H. Yue, D. L. Penh, P. X. Yan, L. S. Wang and X. H. Luo, Structural and
	Optical Properties of SnS Thin Film Prepared by Pulse Electro deposition, J.
[14]	of Alloys and Compounds, Vol. 468 (1-2) (2009) 254-257. S. C. Ray, M. K. Karanjai and D. D. Gupta, Structure and Photo-
	Conductive Properties of Dip Deposited SnS and SnS ₂ Thin Films and Their
	Conversion to Tin Dioxide by Annealing in Air, Thin Solid Films, Vol. 350
	(199) 72-78.
[15]	B. Thangaraju and P. Kaliannan, Spray Pyrolytic Deposition and
	Characterization of SnS And SnS ₂ Thin Films, J. Phys. D. Appl. Phys., Vol.
	33 (2000) 1054-1061).
[16]	C. An, K. Tang, G. Shen, C. Wang, Q. Yang, B. Hai and Y. Qin, Growth Of
	Betlt-Like SnS Crystals From-Ethylenediamine Solution, Journal of Crystal
[17]	Growth, Vol. 244 (3-4) (2002) 333-338. A. Tanuesevski, Optical and Photoelectric Properties of SnS Thin Films
[*']	
[18]	Prepared by Chemical Bath Deposition, Vol. 18 (6) (2003) 501. J. B. Johnson, H. Jones, B. S. Latham, J. D. Parker, R. D. Engelken and C.
	Barber, Optimization of Photo Conductivity in Vacuum-Evaporated Tin
	Sulfide Thin Films, Semiconductor Science and Technology Vol. 14 (6)
	(1999) 501-507.

[19]	F. Vrbach, The Long Wavelength Edge ff Photographic Sensitivity and of
[20]	the Electronic Absorption of Solids, Phys. Rev., Vol. 92 (1953) 1324-1332. S. K. Oleary, S. Zukotynski and J. M. Perz,Disorder and Optical Absorption
	In Amorphous Silicon and Amorphous Germanium, Journal of Non-
[21]	Crystalline Solids, Vol. 210 (2-3) (1997) 249-253. S. H. Wimple And M. Didomenico, Behavior of The Electronic Dielectric
	Constant In Covalent and Ionic Material, Phys. Rev. B., Vol. 3 (1971) 1338-
[22]	1345. N. K. Reddy and K. T. R. Reddy, Optical Behavior of Sprayed Tin Sulphide
[23]	Thin Films, Materials Research Bulletin, Vol. 41(2006) 414-422. H. Nozaki, M. Onoda, M. Sekita and T. Wada, Variation of Lattice
	Dimensions in Epitaxial SnS Films on MgO (001), Journal of Solid State
[24]	Chemistry, Vol. 178 (1) (2005) 245-252. M. Deopura, C. K. Ullal, B. Temelkuran and Y. Fink, Dielectric
	Omnidirectional Visible Reflector Optics Letters, Vol. 26 (15)(2001) 1197-
[25]	119. S. H. Wimple and M. Didomenico, Optical Dispersion and The Structure of
[26]	Solids, Phys. Rev. Lett., Vol. 23 (1969)1156-1160. M. Didomenico and S. H. Wimple, Oxygen-Octahedra Ferroelectrics:
	Theory of Electro-Optical and Nonlinear Optical Effects, J. Appl. Phys., Vol.
	40 (1969) 720.