### Investigation The Effect of oxygen flow rate on the properties of photoconductive TiO<sub>2</sub>/Si UV detector

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#### **Abstract:**

UV photoconductive detector was fabricated by the deposition of  $TiO_2$  film on silicon by dc magnetron sputtering. Pure oxygen and Argon were used as the reactive and sputtering gases respectively. The gases were mixed prior to the admission in the sputtering chamber [oxygen flow 10, 20, 30, 40 and 50 s.m (stander centimeter cubic per minute) with constant argon flow]. The structure properties of  $TiO_2$  thin films were investigated by means of X-ray diffraction. X-ray diffraction (XRD) patterns showed three phases Rutile, Anatase and Brookite . The ratio between these phases changed and depended on oxygen flow. The deposited  $TiO_2$  film was coated by nanosheet of polyamind polymer to improve the photoresponsivity of the detector and to reduced the response time of the  $TiO_2$  UV detector to about 170  $\mu$ s. The final device was tested with high speed pulsed nitrogen laser.

The spectral responses of TiO<sub>2</sub>/Si junction were studied. The maximum value of responsivity occurred at wavelength equals to 385 nm. The responsivity, quantum efficiency and specific detectivity increased with increasing of the oxygen flow, while the noise equivalent power decreased with increasing the oxygen flow. It has been observed that the best spectral response occurs when the oxygen flow equal to 50 s.m and we can say that this value of oxygen flow is the optimum condition for TiO<sub>2</sub> photoconductive detector preparation.

**Keywords:** responsivity; quantum efficiency ;specific detectivity; polyamind polymer; noise equivalent power.

## استقصاء تاثير معدل تدفق الاوكسجين على خواص كاشف التوصيليه الضوئيه للأشعه فوق البنفسجيه لأوكسيد التيتانيوم/سيليكون

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الخلاصيه

تم تصتيع كاشف التوصيلية الضوئية بترسيب  ${
m TiO_2}$  على ركائز السيليكون بواسطة الترذيذ المغناطيسي المستمر . تم استخدام الأوكسجين النقي والأركون النقي كغاز متفاعل وغاز ترذيذ على التوالي. خلط غاز الأوكسجين بمعايير جريان ( ${
m S.m}$   ${
m S.m}$   ${
m S.m}$  (حجم مكعب لكل دقيقة) والأركون بجريان ثابت مقداره  ${
m S.m}$   ${
m S.m}$  دخولهما الى حجرة التغريغ .

وقد تم التحقيق الخصائص التركيبية لأغشية TiO2 الرقيقة عن طريق حيود الأشعة السينية أظهرت نتائج حيود الأشعة السينية (XRD) ثلاث أطوار وهي Rutile وAnatase وBrookite تغير نسب هذه الأطوار بالاعتماد على تغير تدفق الأوكسجين. تم تحسين الاستجابة الضوئية لأغشيه ثنائي أوكسيد التيتانيوم (2.24 A/W) بطلائها بطبقه نانويه من بوليمير نوع (بولي أمايند). أظهرت هذه الكواشف سرعه أستجابه عاليه لأشعه فوق البنفسجيه بزمن نهوض قدره 170 مايكروثانيه عند تعرضها الى ليزر نايتروجين نبضي عالى السرعه. درست الخصائص الطيفية للمفرق TiO2/Si ووجدنا أن أعظم قيمة للاستجابة الطيفية حدثت عند طول موجي

درست الخصائص الطيفية للمفرق TiO2/Si ووجدنا أن أعظم قيمة للاستجابة الطيفية حدثت عند طول موجي يساوي 385nm. أن الاستجابة والكشفية النوعية والكفاءة الكمية تزداد بزيادة تدفق الأوكسجين، في حين أن القدرة المكافئة للضوضاء تتناقص مع زيادة تدفق الأوكسجين. وقد لوحظ أن أفضل استجابة طيفية تحدث عندما يكون تدفق الأوكسجين مساوي ل 50 sccm وإن هذه القيمة هي المثلى لتحضير غشاء ال TiO2 ليعمل كنبيطة ضوئية.

الكلمات المفتاحية: الاستجابية ، الكفائة الكمية،الكشفية النوعية ، بوليمير بولى أمايند ، القدرة المكافئة للضوضاء .

#### 1-Introduction

Titanium dioxide (titania, TiO<sub>2</sub>) is one of the most widely used materials in thin films because it has numerous applications, such as solar cells, photocatalysts, gas sensors, optical coatings, ect<sup>[1-4]</sup>.

dioxide Titanium (TiO<sub>2</sub>)many excellent physical properties such as a high dielectric constant, a mechanical and chemical stability, as well as good insulating properties. Due its to high optical refractive index and transmittance in the visible range, is especially suitable material for optical coatings and protective layers for very largescale integrated circuits [5-8].

TiO<sub>2</sub> belongs to the family of transition metal oxides. There are three commonly known polymorphs of TiO<sub>2</sub> found in nature: rutile (tetragonal), anatase (tetragonal) and brookite (orthorhombic) [9].

TiO<sub>2</sub> is an effective ultraviolet (UV) radiation absorber, due to its ability to absorb ultraviolet (UV) radiation with energies equal to or

#### 2-Experimental Work 2-1 Preparation of TiO<sub>2</sub> Thin Film

In d.c magnetron sputtering a circular magnetron with 10 cm diameter and 400 gausses was used. The discharge characteristics have been controlled using variable d.c supply (5kVand 500 mA). Pure Ti (99.99%) with a diameter of 10 cm and 0.4 cm thickness has been used as a sputtering target; the flow meters were used to control the gases in vacuum chamber. The best distance between target and substrate was 4.2 cm, and

the substrate was set above the target. Glass plates and wafer silicon were used as substrate. greater than its band gap energy [10]

The detector based UV polycrystalline TiO<sub>2</sub> thin film shows low responsivity and long response time which is of the order of few minutes. Since the one dimension TiO<sub>2</sub> nanostructures are characterized by presence of deep level surface trap states, the TiO2 detector exhibits long lifetime of the photo carriers [11]. Despite a great deal of research on TiO2 UV detector. most of the research concentrated on the improvements of the micro mask electrodes, in order to enhance the performance  $TiO_2$ photoconductive the detectors [12-14].

The improvement of the photoresposivity of the TiO2 UV detectors was carried out by the surface treatment of the TiO2 thin film. The covering of the TiO<sub>2</sub> film surface with nanosheet of different of polymers has improved the detector performance [15,16]. In this work, a simple and reliable technique is used highly to fabricate photoconductive TiO<sub>2</sub> UV detector.

Before deposition, The growth chamber was evacuated to a base pressure of  $6 \times 10^{-4}$  pa for 30 min, then Ar (99.99%) and O<sub>2</sub> (99.99%) gases were introduced into the chamber as working gas. The chamber total pressure was fixed at 0.2 Pa during the whole deposition, and pre-sputtering process employed for 10 minutes to clean the target surface. A direct voltage (1000 V the d.c power was maintained at 100W) is between the cathode and anode. Sputtering is a removal of surface atoms due to particle bombardment which is caused by atomic collisions at the surface of a target.

The crystalline structure of 200 nm thickness  $\text{TiO}_2$  film deposited on

consisted

glass substrate was studied by X-ray diffraction using XRD-6000-Schemadzu system  $\lambda=1.54$ nm. Photodetector Characteristics (responsivity, quantum efficiency, and signal to noise ratio, noise equivalent power and detectivity) of  $\text{TiO}_2$  /Si detector was measured using the

# monochrometer in the range (200-900) nm, electrometer and power meter for measuring the radiation power for each wavelength and d.c power supply to supply bias voltage on each side of the detector.

is

which

system

#### **3- Result and Discussion:**

#### **3-1 Structure Characteristic**

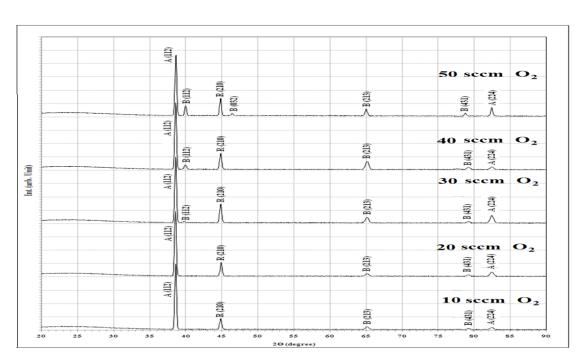
The XRD pattern of a TiO<sub>2</sub> film is illustrated in Fig. (1). It shows that all films of TiO<sub>2</sub> have multiphase Brookite (B), Anatase (A) and Rutile (R). When the oxygen flow (10 and 20 s.m) there are five peaks A (112), R(210), B(213), B(431) and A(224) at angles  $(2\theta=38.5^{\circ})$ ,  $(2\theta=46.8^{\circ})$ ,  $(2\theta=64.9^{\circ})$ ,  $(2\theta=79.3^{\circ})$ and  $(2\theta=82.2^{\circ})$  respectively. While when the oxygen flows (30 and 40 s.m), the peak B (112) grown at angle  $(2\theta=39.5^{\circ})$  and another peak B (032) grown in angle

 $(2\theta=46.1^{\circ})$  at the oxygen flow became 50s.m.

The grain size (estimated by Scherrer formula: (17)

$$D = \frac{0.9\lambda}{\beta \cos \theta} \dots (1)$$

From the data obtained, we can see decrease in grain size as oxygen flow increased. When the oxygen flow increased from 10 s.m to 50 s.m the grain size was reduced from (32.7 nm to 20.5 nm), (22.6 nm to 21nm) and (27.3 nm to 25nm) for Brookite, Anatase and Rutile respectively. (18)



Fig(1) XRD of TiO<sub>2</sub> thin films on glass substrate for different Oxygen flow rate.

#### 3-2 Spectral Measurements 3-2-1 Spectral Responsivity

The spectral responsivity of an optical detector is a measure of its response to radiation at a specified Wavelength. The responsivity for monochromatic light of wavelength incident normally is given by [19].

$$R_{\lambda} = \frac{I_{ph}}{P_{in}}$$
 or  $R_{\lambda} = \frac{V}{P_{in}}$ ....(2)

Fig.(2) Shows the wavelength dependent of spectral responsivity  $(R_{\lambda})$  for  $TiO_2$  thin films at room temperature under different oxygen flow rate 10, 30 and 50 s.m . It is clear

that a maximum responsivity appears at the ultraviolet region and its peak is at about 385 nm. When the oxygen flow rate increase from 10 to 50 s.m the responsivity increasing from 0.03 to 0.07 A/W (at 385nm). This is attributed to the increasing roughness with increasing of oxygen flow rate which results to increase absorption coefficient (decrease Transmission) and then more power of the incident radiation is absorption in thin film which are generate electron-hole pairs<sup>[20]</sup>.

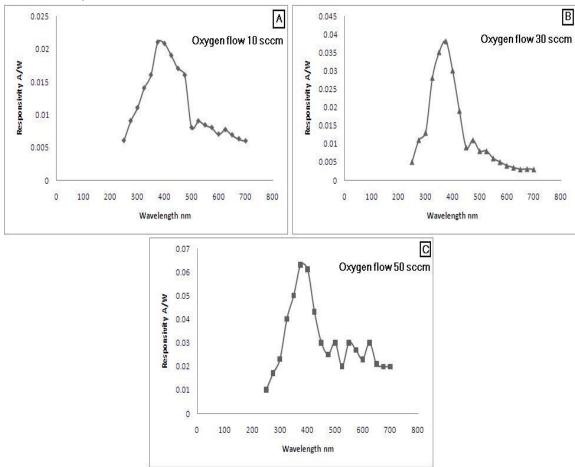


Fig.(2)The variation of spectral responsivity as a function of wavelength for Oxygen flow rate.

#### 3-2-2 Quantum Efficiency

The quantum efficiency was determined as a function of wavelength for TiO<sub>2</sub>/Si junction; it can be expressed [19].

$$\eta = (I_{ph}/e).(h\upsilon/p_{in})....(3)$$

efficiency was achieved at 385 nm with a value of 7.4% for TiO<sub>2</sub>/Si at Oxygen flow equal 10 s.m and increased to 25% at Oxygen flow equal

50 s.m. The reasons are mentioned earlier in the discussion of the spectral responsivity.

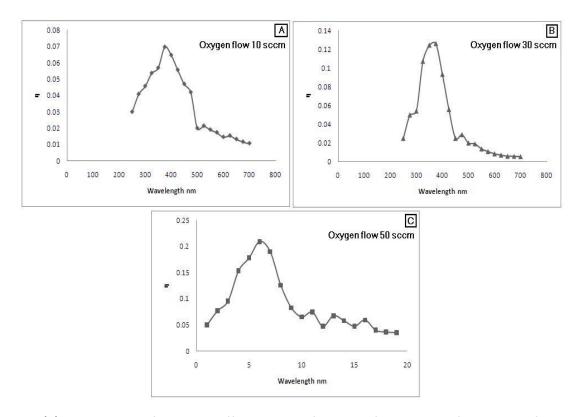


Fig.(3)The variation of quantum efficiency as a function of wavelength for TiO<sub>2</sub> thin films prepared at different Oxygen flow rate.

#### 3-2-3 Noise Equivalent Power

The Noise Equivalent Power (NEP) is a concept often used to quantify the sensitivity of a detector or the power generated by a source of noise on a detector NEP defined as [19]:

 $NEP \!\!=\!\! I_n \!/ \; R_{\lambda} \quad Watt \; ..... \qquad (5)$ 

The variation of NEP as a function of wavelength under Oxygen flow rates 10, 30 and 50 s.m is shown in Fig. (4).

In general, the minimum NEP occurs when  $R_{\lambda}$  has the maximum value. From these figures, we can notice that NEP decreases with increasing of Oxygen flow. It decreases from 2.4  $\times 10^{-8}$  (Watt) to  $0.35\times 10^{-8}$  (Watt) when Oxygen flow rate varies from 10 s.m to 50 s.m, and we relate this to the defects

(voids) which decrease with increasing Oxygen flow rate, and thus decrease noise current.

#### 3-2-4 Specific Detectivity

The specific detectivity (D\*) of the TiO<sub>2</sub> thin films were measured at room temperature as a function of incident light wavelength, it is defined as <sup>[19]</sup>:

D\*=D  $(A \Delta F)^{1/2}$  cm.  $Hz^{1/2}/Watt...$  (6) or

The response time of the fabricated  $TiO_2$  UV detector was tested with nitrogen laser of 0.3 ns pulse duration and  $50\mu j$  energy.

It can be noticed from the traced pulse shape that the rise time was of the order of  $170\mu s$  and the fall time was about  $750\mu s$ .

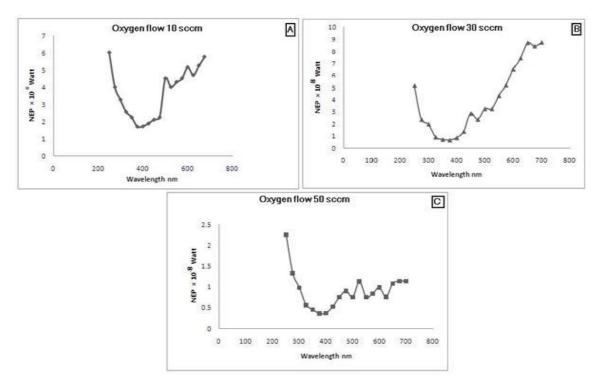


Fig.(4)The variation of NEP as a function of wavelength for TiO<sub>2</sub> thin films prepared at different Oxygen flows.

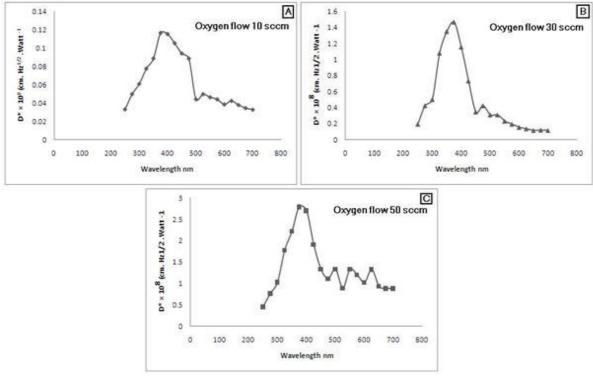


Fig. (5)The variation of Specific Detectivity as a function of wavelength for TiO<sub>2</sub> thin films prepared at different Oxygen flows

The slow decay time is due to slow escape of holes from the tarps. . Hole captured into the hole traps may be emitted back into the valance band according to a time constant depends on energy separation between the corresponding hole traps and the edge of the valance band. The deep traps

have slow response and this may explain the slow fall time of the output pulse as shown in Fig .(6). The long tail accompanied with the pulse is due to sample heating by the high repetitions rate  $N_2$  laser illuminate the TiO<sub>2</sub> photoconductive detector <sup>[21]</sup>.

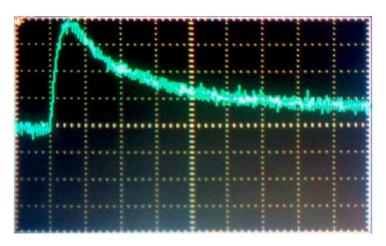


Fig.(6) The photoresponse time of fabricated TiO<sub>2</sub> UV detector to the nitrogen laser. The time base on x-

#### 4- Conclusion:

The conclusions of this work The XRD analysis showed that TiO2 thin mixture (Rutile, Anatase and Brookite). Dependence on oxygen flow rate. Spectral Measurements show that the responsivity, specific detectivity and quantum efficiency increase with increasing of oxygen flow rate, while noise equivalent power decreases with

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films was polycrystalline in sputtering in oxide mode and have three phases increasing oxygen flow. The functionalization of the  $TiO_2$  film surface by polyamide nylon highly improved the photoresponse time reaches to  $170\mu s$  which is six orders of magnitudes faster than that of  $TiO_2$  photoconductive detector.

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