EFFECT OF GIBBERELLIC ACID, CHELATED IRON COCONUT LIQUID IN SOME FLOWERING GROWTH TRAITS AND THE VOLATILE OIL YIELD FOR Jasminum sambac (L.) Ait PLANT.

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ABSTRACT

The study was conducted during the two growth seasons (2015-2016 and 2016-2017) at the production nursery (Saran) belonging to the Research Station, College of Agriculture, University of Basrah, Qarmat Ali city. The study aimed to investigate the effect of spraying Gibberellic Acid (GA3), chelated Iron and Coconut liquid on some flowering growth indicators and volatile oil yield of Jasminum sambac (L.) Ait. The study included 27 factorial treatments, It included combinations of three factors, namely, Gibberellic acid with three concentrations of (0, 150, 300 mg. L⁻¹), the chelated Iron with three concentrations of (0, 50, 100 mg. L⁻¹) and Coconut liquid with three concentrations of (0, 15, 30%). The Complete Randomized Block Design was used, with three replicates for each treatment. Thus, the number of experimental units is 81 experimental units, with a rate of 6 plants per experimental unit. So, the number of plants used in the experiment is 486 plants. The averages were Compared according to the Revised Least Significant Difference (R.L.S.D.) Test at probability level of 0.05. The following are the main results:

- 1- Spraying plants with Gibberellic acid at a concentration of (300 mg. L⁻¹) led to a significant increase in the number of flowers and total flower yield and the volatile oil yield and the fresh weight for flowers, in the first season and when spraying it with a concentration of (150 or 300 mg. L⁻¹) led to a significant increase in the percentage of volatile oil for the two seasons of study, and fresh weight for flowers in the second season.
- 2- Jasminum sambac (L.) Ait plant responded to spraying with chelated iron, spraying it with a concentration of (100 mg. L⁻¹) led to a significant increase in the number of flowers, the fresh weight of the flowers, total flower yield, the percentage of volatile oil and the volatile oil yield for the two study seasons.
- 3- The spraying of plants with coconuts at a concentration of 30% had a significant effect on the number of flowers, total flowers yield and volatile oil yield for the two study seasons, When spraying it with a concentration of (15% or 30%) led to a significant increase in the percentage of volatile oil for the two study seasons.
- 4- Bi-interactions and triple interactions had a significant effect for most flowering growth indicators and volatile oil yield for the two study seasons.

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تأثير حامض الجبرليك والحديد المخلبي وسائل جوز الهند في بعض صفات النمو الزهري وحاصل الزيت العطري Jasminum sambac (L.) Ait . الطيار لنبات الرازقي

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الخلاصة

أجريت الدراسة خلال موسمي النمو 2015-2016 و2016-2017 في الظلة القماشية التابعة لمحطة أبحاث كلية الزراعة حجامعة البصرة حموقع كرمة على . استهدفت الدراسة معرفة تأثير الرش بحامض الجبرليك والحديد المخلبي وسائل جوز الهند في بعض مؤشرات النمو الزهري وحاصل الزيت العطري الطيار لنبات الرازقي Aiton (L.) Aiton ملعه تو لنها تعريبات الدراسة 27 معاملة عامليه شملت توليفات من ثلاث عوامل هي حامض الجبرليك بثلاث تراكيز (0 و 10 و 300) ملغم. لتر⁻¹ والحديد المخلبي بثلاث تراكيز (0 و 10 و 300)% استعمل تصميم القطاعات العشوائية بثلاث تراكيز (0 و 10 و 30 و 300)% استعمل تصميم القطاعات العشوائية الكاملة بتجربة عامليه بثلاث عوامل وبثلاث مكررات لكل عامل وبهذا تكون عدد الوحدات التجريبية 81 وحدة تجريبية وبواقع 6 الكاملة بتجربة عامليه بثلاث عوامل وبثلاث مكررات لكل عامل وبهذا تكون عدد الوحدات التجريبية 18 وحدة تجريبية وبواقع 6 نبتات لكل محدة تجريبية وبعذا يكون عدد النباتات المستخدمة بالتجربة 486 نبات . قورنت المتوسطات حسب اختبار اقل فرق معنوي نبتات لكل وحدة تجريبية و 300 وفيما يلي اهم النتائج التي تم التوصل اليها: أدى رش النباتات بحامض الجبرليك بتركيز 300 ملغم. لتر⁻¹ الف فرق معنوي معدل عند مستوى احتمالية و300 وفيما يلي اهم النتائج التي تم التوصل اليها: أدى رش النباتات بحامض الجريبية و 300 معنوي التر⁻¹ الى زيادة معنوي احتمالية و300 وفيما يلي اهم النتائج التي تم التوصل اليها: أدى رش النباتات بحامض الحريبية و 300 معنوي التر⁻¹ الى زيادة معنوية في عدد الإذهار و حاصل الاز هار الكلي وحاصل الزيت العطري الطيار والوزن الطري للأذهار ، في الموسم الأول و عند رشه بتركيز 300 ملغم. التر⁻¹ أدى الى زيادة معنوية في عدد الأذهار ، في الموسم الولي للأول و عند رشه بتركيز 300 ملغم. لتر⁻¹ أدى الى زيادة معنوية في النسبة المئوية للزيت الطيل لموسمي الدراسه، والوزن الطري للأول و عند رشه بتركيز 300 ملغم. لتر⁻¹ أدى الى زيادة معنوية في عدد الإذهار موالوزن الطري للأول و حد رشه بتركيز 300 ملغم. لتر⁻¹ ألى زيادة معنوية في عدد الأول في الموسمي الدراسه، كان لرش النداني الرازي الحمون الي بلدي والنسبة المئوية للزيت العطري الطيار وحاصل الزياد معنوي ألموي الطري للأزهار وحاصل الزيان و 30% مالي والمورن الطري للأول و ملار في والوزن الطري للأزهار وحاصل الزياد والوي لي معنوي في عدد الأدهار والوري الطري للأزهار وحاصل الزياد معنوي والغان وحدة ألى معنوي في عدد الأول في مالي والوزن الطري للأزهار وحاصل الزهار الحمو ي الطبي والموري والوي الطري للأزهار وحاصل الزهار الكلي والنسبة المؤية مريي و30% أدى م

1. INTRODUCTION

Jasminum sambac (L.) Ait follows the oleaceae family. The Jasminum genus contains about 300 species distributed in Asia, Africa, Europe and the region of Pacific Ocean [1]. The original habitat of the jasmine plant is the Mediterranean region, the Levant, North Africa, India and China. The interest with jasmine plant is attributed to its many benefits in the medical, aromatic and cosmetic fields. Its flowers contain a lot of essential oil, which is called jasmine oil, which contains almost 100 ingredients according to the environment in which it grows. The compounds responsible for the fragrance include benzyl acetate, benzyl alcohol and linalool [2]. Jasmine oil is considered one of the most expensive oils to be used in the manufacture of cosmetics, perfumes, soap and pharmaceutical industry [3]. The effect of spraying of Gibberellic acid is not limited to vegetative parts, but is also effective in influencing flowering. The research shows that the Gibberellic acid has an impact in the flowering of most plants, but the extent of their effect varies depending on the adding method and its quantity, as well as the influence of some external factors such as temperature and light and the development stage of the plant [4]. The [5] showed When she added the chelated iron to the Gardenia plant at concentrations of *بحث مستل من رسالة دكتوراه للباحث الاول

(40, 60, 80, 100 mg. flowerpot⁻¹) led to a significant increase in the number of flowers for the plant and the number of petals for flowers, especially when added at the concentration of (80, 100 mg.flowerpot⁻¹). The experiment aims to study the extent of the response of the Jasmine plant to spraying with both the Gibberellic acid, the chelated iron and the coconut liquid in some flowering growth indicators and volatile oil yield.

2. MATERIALS AND METHODS

The study was conducted at the production nursery (Saran) belonging to the Research Station, College of Agriculture, University of Basrah, Oarmat Ali city, and for two consecutive seasons from 1/11/2015 to 1/11/2017. The study included 27 factorial treatments, It included combinations of three factors, namely, Gibberellic acid with three concentrations of $(0, 150, 300 \text{ mg}. \text{ L}^{-1})$, the chelated Iron with three concentrations of (0, 50, 100 mg. L^{-1}) and Coconut liquid with three concentrations of (0, 15, 30%). The aqueous solutions were prepared for it and were sprayed on the plant three times between one spray and month. The another is one Complete Randomized Block Design was used, with three replicates for each treatment. Thus, the number of experimental units is 81 experimental units,

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with a rate of 6 plants per experimental unit. So, the number of plants used in the experiment is 486 plants. The experimental measurements included the number of flowers (flower.plant⁻¹), fresh weight for flowers (g) and flowers yield (flower), which is calculated according to the following equation: fresh weight for flowers × total number of flowers for the plant, and the percentage of the volatile oil which extracted by organic solvents method (Hussein, 1979), it was estimated according [6] to the following equation:

 $\frac{\text{The volatile oil }(g)}{\frac{\text{The weight of the oil sample }(g)}{\text{Sample weight }(g)}} \times 100$

The volatile oil yield, which was calculated according to the following equation:

The volatile oil yield (g) = the percentage of volatile oil \times total plant flowers.

3. RESULT AND DISCUSSION

The number of Flowers (flower.plant⁻¹)

Table (1) shows the plants that treated with Gibberellic acid at a concentration of (300 mg. L^{-1}) was significantly excelled in the number of their flowers for the two study seasons, which amounted of $(23.12, 22.88 \text{ flower. plant}^{-1})$, respectively, compared to the plants the control treatment that amounted of (19.39, 19.28 flower. plant⁻¹), respectively. As for the treating plants with chelated iron, the plants treated with $(100 \text{ mg. } \text{L}^{-1})$ of chelated iron was significantly excelled in the number of flowers and for the two study seasons, which amounted of (22.38, 21.93 flower. plant⁻¹), respectively, compared to the plants of the control treatment that amounted of $(20.40, 20.52 \text{ flower. plant}^{-1})$, respectively. As for spraying with coconut liquid, the plants treated with 30% coconut liquid was significantly excelled in the number of flowers and for the two study seasons, which amounted of $(23.49, 22.88 \text{ flower. plant}^{-1})$, respectively, compared to the plants of the control treatment that amounted of (19.39, 19.18 flower. plant⁻¹), respectively. The

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significant increase in the number of flowers for the plants when spraying it with coconut liquid may be explained by its significant effect in increasing the number of primary and secondary branches and thus increasing the number of flowers. Table (1) shows the significant effect for the plants treated with Gibberellic acid at a concentration of (300 mg. L^{-1}) and the chelated iron at a concentration of (100-mg.L⁻¹) gave the highest number of flowers and for the two study seasons amounted of (24.06, 23.64 flower. plant⁻¹), respectively, compared to the plants of the control treatment which amounted of (19.86, 18.79 flower. plant ¹), respectively. The bi-interaction between the spraving treatment with Gibberellic acid and the coconut liquid had a significant effect on the number of flowers, where the bi-interaction between the plants treated with Gibberellic acid at a concentration of (300 mg. L^{-1}) and the spraying with coconuts liquid at a concentration of 30% was significantly excelled in the increase the number of flowers and for the two study seasons, which amounted of (23.74, 24.02 flower.plant⁻¹), respectively, compared to the lowest number of flowers for the plants of the control treatment, which amounted of (17.43, 16.93 flower.plant⁻¹). As for the effect of the bi-interaction between the two spraying factors with the chelated iron and the coconut liquid was significant in the trait of the number of flowers, where the plants treated with chelated iron at a concentration of (100 mg.L^{-1}) and the coconut liquid at a concentration of 30% gave the highest number of flowers amounted of $(24.36, 23.59 \text{ flower.plant}^{-1})$ for the two study seasons, respectively, compared to the lowest number of flowers fromed by the control treatment plants amounted of (18.41 flower.plant^{$\overline{1}$}) for the first season and plants treated with chelated iron at a concentration of (50 mg.L^{-1}) only amounted of (18.56)flower.plant⁻¹) for the second season. As for the triple interaction, it had a significant effect in this trait, where the plants treated with the Gibberellic acid at the concentration of (300 $mg.L^{-1}$), the chelated iron at the concentration of $(100 \text{ mg}.\text{L}^{-1})$ and the Coconut liquid at the concentration of 30% by forming it the highest number of flowers amounted of (24.67 flower.plant⁻¹) compared to the lowest number of flowers recorded by plants that did not treating with any of the three solutions, which amounted to $(16.44 \text{ flower.plant}^{-1})$.

Table 1: Effect of the Spraying with Gibberellic Acid, chelated iron and the Coconut liquid and their interactions in the number of Flowers (flower.plant⁻¹) for the jasmine Plant for both Study Seasons (2015-2016, 2016-2017).

			Fi	rst seas	on	Second season				
0.11 11.		Coo	conut lic	juid	Interaction	Co	conut liq	uid	Interaction	
Gibberellic	Chelated			Î	(Gibberellic				(Gibberellic	
Acid $(1 + 1)$	Iron $(m \in \mathbf{I}^{-1})$	•	15	20	Acid ×	•	15	20	Acid ×	
$(mg.L^{-1})$	$(mg.L^{-1})$	0	15	30	Chelated	0	15	30	Chelated	
					Iron)				Iron)	
	0	17.40	19.50	22.67	19.86	16.44	18.25	21.67	18.79	
0	50	17.22	20.67	22.75	20.21	16.67	19.25	20.66	18.86	
	100	17.66	19.87	23.90	20.48	17.67	20.55	22.34	20.19	
	0	18.16	19.00	22.67	19.94	17.77	19.75	21.76	19.76	
150	50	19.78	21.66	23.67	21.70	18.36	20.67	22.72	20.58	
	100	20.60	22.76	24.50	22.62	20.65	20.45	24.75	21.95	
	0	19.66	21.75	22.80	21.40	21.67	22.66	24.75	23.03	
300	50	21.16	22.67	23.75	22.53	20.66	23.75	23.63	22.68	
	100	22.90	24.60	24.67	24.06	22.76	24.50	23.67	23.64	
R.L.S.I	0.05	1.322			0.763	1.297			0.749	
Average effect of		19.39	21.39	23.49		19.18	21.09	22.88		
Coconut	Coconut liquid		21.39	23.49		19.10	21.09	22.00		
R.L.S.I) 0.05		0.440			0.432				
					Average	Fffect	of inter	Average		
Effect of in	nteraction (Gibbere	llic acid	l and	effect of	Effect of interaction Gibberellic acid and coconut liquid			effect of	
	coconut	liquid			Gibberellic				Gibberellic	
					acid		-		acid	
0		17.43	20.01	23.11	20.18	16.93	19.35	21.56	19.28	
15		19.51	20.14	23.61	21.11	18.93	20.29	23.08	20.76	
30	-	21.24	23.01	23.74	22.66	21.69	23.64	24.02	23.12	
R.L.S.I	0.05		0.763		0.440		0.749		0.432	
		~		-	Average effect of	Effect	of inter	action	Average	
Effect of	Effect of interaction Chelated Iron and						ated Iro		effect of	
coconut liquid					Chelated		onut liq		Chelated	
			20.00	00.71	Iron		-		Iron	
0		18.41	20.08	22.71	20.40	18.63	20.22	22.73	20.52	
50		19.39	21.67	23.39	21.48	18.56	21.22	22.34	20.71	
		20.39	22.41	24.36	22.38	20.36	21.83	23.59	21.93	
R.L.S.D 0.05			0.763		0.440	0.749			0.432	

The fresh weight of flowers (g)

Table (2) indicates the superiority of the plants treated with Gibberellic acid at a concentration of (300 mg.L^{-1}) in fresh weight of the flowers, which amounted to (2.22 g) compared to the

plants of control treatment, which amounted to (1.99 g). As for the second season, the plants that treated with Gibberellic acid at the concentration of $(150 \text{ or } 300 \text{ mg.L}^{-1})$ has significantly excelled in the fresh weight of the flowers, which amounted to (2.13, 2.11 g),

respectively, compared to the plants of control treatment, which amounted to (1.98 g). The spraying of Gibberellic acid may be explained by the activation of cell division and its expansion by increasing the decomposed starch by increasing the elasticity of the wall and thus expanding the cells by increasing the cell's Osmosis content, thereby increasing their absorption of water [7]. As for plants treated with Chelated iron, the same table showed a significant superiority for the plants treated with (100 mg.L^{-1}) Chelated iron which the average fresh weight of their flowers amounted of (2.26 g) compared to the plants of the control treatments which recorded (1.97 g) in the first season, In the second season, there was a significant superiority of plants treated with Chelated iron with a concentration of (100 $mg.L^{-1}$) in fresh weight of their flowers, which amounted of (2.19 g) compared to the control plants, which recorded (1.95 g). As for the spraving with coconut liquid, the results showed the same table to the absence of a significant effect in this and both seasons of the study. Table (2) shows the significant effect for the interaction between the two study factors (the spraying with Gibberellic acid and Chelated iron) in the fresh weight of flowers. The plants of the first season that treated with Gibberellic acid at a concentration of (300 $mg.L^{-1}$) and Chelated iron with a concentration of (50 mg.L^{-1}) gave the highest fresh weight for its flowers amounted of (2.29 g), compared to the lowest fresh weight for the flowers of the control plants which amounted of (1.84 g), In the second season, the plants were treated with Gibberellic acid at concentration of (150 gm.L⁻ ¹) and the Chelated iron at concentration of (100 mg.L^{-1}) gave the highest fresh weight for the flowers, which amounted to (2.24 g)compared to the plants of the control treatment which amounted of (1.88 g). As for the biinteraction between the spraying with the Gibberellic acid and coconut liquid had a significant effect in this trait, The plants of the first season and the treating with Gibberellic acid at concentration of (300 mg.L^{-1}) and the

Coconut liquid with a concentration of 30% characterized by the highest fresh weight for the flowers amounted of (2.26 g) compared to the plants of the control treatment which amounted of (1.91 g). In the second season, the plants were treated with Gibberellic acid at concentration of $(150 \text{ gm}.\text{L}^{-1})$ and the coconut liquid at concentration of (30%) gave the highest fresh weight for the flowers, which amounted to (2.22 g) compared to the plants of the control treatment which amounted of (1.97 g). The bi-interaction between the spraying with Chelated iron and coconut liquid had a significant effect in this trait. The plants that were sprayed with Chelated iron at a concentration of (100 mg.L^{-1}) and the Coconut liquid with a concentration of 30% gave the highest fresh weight for the flowers amounted of (2.29 g) compared to the plants of the control treatment which gave the lowest fresh weight for the flowes amounted of (1.89 g) during the first season. The plants that were spraved with Chelated iron at a concentration of (100 mg.L^{-1}) and did not spray with the Coconut liquid with a concentration gave the highest fresh weight for the flowers amounted of (2.26 g) compared to 1.88 g for plants that did not spray any of the solutions. Table (2) shows the significant effect of the triple interaction in the fresh weight for the flowers, where the plants treated with (300 $mg.L^{-1}$) of Gibberellic acid and (50 $mg.L^{-1}$) of Chelated iron only and the plants treated with Chelated iron at a concentration of (100 mg.L^{-1}) an the Coconut liquid with a concentration of only 30% characterized by recording it the highest fresh weight for the flowers amounted of (2.32 g) compared to the lowest fresh weight of flowers recorded by the control plants, which amounted of (1.82 g) in the first season, In the second season, the same table showed that the plants treated with (150 mg.L⁻¹) of Gibberellic acid, (100 mg.L^{-1}) of Chelated iron and 30% of Coconut Liquid was excelled from the rest of the plants by giving it the highest fresh weight for its flowers amounted of (2.30 g) compared to plants not treated with any of the three solutions, which amounted of (1.83 g).

Table 2: Effect of the Spraying with Gibberellic Acid, chelated iron and the Coconut liquid and theirinteractions in the fresh weight for the Flowers (g) for the jasmine Plant for both Study Seasons (2015-2016, 2016-2017).

			I	First se	eason		Se	cond sea	ason
Gibberellic Chelated		Coc	onut li	quid	Interaction	Co	conut lic	Interaction	
Acid (mg. L^{-}					(Gibberellic				(Gibberellic
1)	$(mg.L^{-1})$	0	15	30	Acid ×	0	15	30	Acid ×
	_				Chelated Iron)				Chelated Iron)
	0	1.82	1.87	1.83	1.84	1.83	1.96	1.86	1.88
0	50	1.78	1.85	2.07	1.90	1.84	1.92	1.96	1.91
	100	2.13	2.26	2.32	2.24	2.25	2.15	2.08	2.16
	0	1.87	1.99	2.06	1.97	1.87	1.95	2.09	1.97
150	50	2.20	2.25	2.27	2.24	2.09	2.21	2.27	2.19
	100	2.23	2.27	2.28	2.26	2.27	2.15	2.30	2.24
	0	1.97	2.06	2.25	2.09	1.95	2.02	2.06	2.01
300	50	2.32	2.31	2.25	2.29	2.06	2.22	2.17	2.15
	100	2.27	2.30	2.28	2.28	2.26	2.07	2.15	2.16
R.L.S.I	0.05		0.647		0.373	0.201		0.116	
Average of	Average effect of		2.13	2.18		2.05	2.07	2.10	
Coconut	Coconut liquid		2.13	2.10		2.05	2.07	2.10	
R.L.S.I	0.05		NS			NS			
Effect of inte	eraction Gil coconut lic		ic acid	l and	Average effect of Gibberellic	Effect of interaction Gibberellic acid			Average effect of Gibberellic
	coconut ne	luiu			acid and coconut liquid				acid
0		1.91	1.99	2.07	1.99	1.97	2.01	1.97	1.98
15		2.10	2.17	2.20	2.16	2.08	2.10	2.22	2.13
30	0	2.19	2.22	2.26	2.22	2.09	2.10	2.13	2.11
R.L.S.I) 0.05		0.373		0.215		0.116		0.067
Effort of in	toraction C	holotor	IIron	and	Average	Effect	of inter	action	Average
Effect of interaction Chelated Iron and coconut liquid					effect of	Chela	ated Iro	n and	effect of
coconut nquiu					Chelated Iron		onut liq		Chelated Iron
0	0		1.97	2.05	1.97	1.88	1.98	2.00	1.95
50		2.10	2.14	2.20	2.15	2.00	2.12	2.13	2.08
10		2.21	2.28	2.29	2.26	2.26	2.12	2.18	2.19
R.L.S.D 0.05			0.373		0.215	0.116			0.067

Total yield of flowers (g)

Table (3) indicates the superiority of plants treated with Gibberellic acid at a concentration of (300 mg.L⁻¹) in the flowers yield and for both study seasons, which amounted to (50.52, 48.73 g) significantly compared to the plants of the control which recorded (40.42, 38.31g). The significant increase in total flowers yield due to spraying with Gibberellic acid was attributed to

its positive role in increasing the number of primary and secondary branches, respectively, and increasing the efficiency of photosynthesis. These led to an increase in the number of the formed flowers as shown in Table (1). As for the spraying of plants with Chelated Iron, the table showed the superiority of plants treated with Chelated iron at a concentration of (100 mg. L^{-1}) significantly in the flowers yield and

for both study seasons, which amounted of (50.69, 47.97 g) compared to the control treatment which recorded (40.36, 40.27 g), As for the plants treated with coconut liquid, the plants treated with 30% of coconut liquid was significantly excelled in the flowers yield and for both study seasons, which amounted of (51.29, 48.28 g) respectively, compared to the control plants which amounted of (40.35, 39.59 g). The reason may be explained by the significant effect of coconut liquid in increasing the number of flowers as shown in Table (1), thus increasing the total yield of flowers. The results of Table (3) showed the effect of the interaction between the two study factors of spraying with (Gibberellic acid and Chelated Iron) in the total yield of flowers, where the plants treated with Gibberellic acid at a concentration of (300 mg.L^{-1}) and the Chelated Iron at concentration of (100 mg.L^{-1}) were characterized by giving it the highest total yield of flowers for both study seasons amounted of (54.94, 51.02 g), respectively, compared to the lowest total yield of flowers for the control plants, which amounted to (36.57, 35.39 g), respectively. The bi-interaction for the two study factors, spraying with Gibberellic acid and coconut liquid had a significant effect in this trait, where the plants treated with Gibberellic acid at a concentration of (300 $mg.L^{-1}$) and Coconut liquid with concentration of 30% gave the highest total yield of flowers amounted of (53.69 g)

compared to (33.3 g) resulted from the control plants during the first season, In the second season was the excellence for the plants that sprayed with a Gibberellic acid at a concentration of (150 mg.L⁻¹) and the Coconut liquid with a concentration of 30% by giving it the highest yield of flowers amounted of (51.39 g) compared to (33.56 g) for the control plants. As for the effect of the bi-interaction between the two spraying factors with Chelated Iron and coconut liquid was also significant in this trait. The plants treated with the Chelated iron at concentrated of $(100 \text{ mg}.\text{L}^{-1})$ and Coconut liquid with a concentration of 30% gave the highest yield of flowers for both study seasons amounted of (55.85, 51.44 g), respectively, compared to (34.79, 35.19 g) for the control plants. Table (3) indicates that the triple interaction has a significant effect on the yield of flowers, where the plants treated with (300 $mg.L^{-1}$) of Gibberellic acid, (100 mg. L^{-1}) of Chelated iron and 15% coconut liquid recorded the highest yield of flowers amounted of 56.58 g compared to the lowest yield of flowers for plants treated with Chelated iron with a concentration of (50 mg.L^{-1}) only which amounted of (30.65 g) in the first season, In the second season, the plants treated with (150 mg. L^{-1}) of Gibberellic acid, (100 mg. L^{-1}) of Chelated iron and 30% of Coconut Liquid has excelled by giving it the highest yield of flowers amounted of 56.97 g compared to (30.08 g) for the control plants.

Table 3: Effect of the Spraying with Gibberellic Acid, chelated iron and the Coconut liquid and theirinteractions in the total yield of flowers (g) for the jasmine Plant for both Study Seasons (2015-2016,
2016-2017).

			Fi	rst seas	on		Sec	ond seas	son
Gibberellic	Chelated	Coconut liquid			Interaction	Co	conut liq	uid	Interaction
Acid	Iron				(Gibberellic				(Gibberellic
(mg.L^{-1})	(mg.L^{-1})	0	15	30	Acid ×	0	15	30	Acid ×
(mg.L)	(ing.L)	U	15	50	Chelated	U	15	50	Chelated
					Iron)				Iron)
	0	1.82	1.87	1.83	1.84	1.83	1.96	1.86	1.88
0	50	1.78	1.85	2.07	1.90	1.84	1.92	1.96	1.91
	100	2.13	2.26	2.32	2.24	2.25	2.15	2.08	2.16
	0	1.87	1.99	2.06	1.97	1.87	1.95	2.09	1.97
150	50	2.20	2.25	2.27	2.24	2.09	2.21	2.27	2.19
	100	2.23	2.27	2.28	2.26	2.27	2.15	2.30	2.24
	0	1.97	2.06	2.25	2.09	1.95	2.02	2.06	2.01
300	50	2.32	2.31	2.25	2.29	2.06	2.22	2.17	2.15
	100	2.27	2.30	2.28	2.28	2.26	2.07	2.15	2.16
R.L.S.I	0.05		1.172		0.677	1.733			1.000
Average	Average effect of		45.73	51.29		39.59	43.86	48.28	
Coconut	Coconut liquid		43.75	51.29		39.39	45.80	40.20	
R.L.S.I) 0.05		0.391				0.578		
					Average	Effoot	ofintor	action	Average
Effect of in	nteraction (Gibbere	llic acid	l and	effect of Gibberellic acid and				effect of
	coconut	liquid			Gibberellic		onut liq	Gibberellic	
		I		I	acid		-		acid
0		33.31	39.87	48.09	40.42	33.56	38.96	42.42	38.31
15		41.14	46.07	52.10	46.44	39.79	42.88	51.39	44.69
30		46.60	51.25	53.69	50.52	45.42	49.74	51.04	48.73
R.L.S.I) 0.05		0.677		0.391	1.000			0.578
					Average	Effect	of inter	action	Average
Effect of	interaction		ed Iron	and	effect of		ated Iro		effect of
coconut liquid					Chelated		onut liq		Chelated
					Iron		-	-	Iron
0		34.79	39.69	46.61	40.36	35.19	40.02	45.61	40.27
50		41.09	46.45	51.42	46.32	37.56	45.11	47.80	43.49
10		45.18	51.05	55.85	50.69	46.02	46.45	51.44	47.97
R.L.S.D 0.05			0.677		0.391	1.000			0.578

The percentage of volatile oil (%)

Table (4) indicates the superiority of plants treated with Gibberellic acid at a concentration of (150 mg.L^{-1}) in the percentage of volatile oil and for both study seasons, which amounted to (0.629, 0.640 %) and (0.638, 0.654%) respectively, significantly compared to the

control plants which amounted of (40.42, 38.31g). This may be due to the role of gibberellic acid in the genes production in the cell chromosomes, thus stimulating mDNA and after formation Some enzymes, which represent primary products, are composed of byproducts, such as volatile oil [8]. This result agrees with

the result of [9] on the Pelargonium graveolens plant. As for the spraying with Chelated iron in this trait has had a significant effect where the plants treated with Chelated iron at a concentration of (100 mg.L⁻¹) for the first season was significant in the percentage of its volatile oil, which amounted of (0.644%)compared to the control plants, where the percentage of its volatile oil amounted of (0.604, 0.613%). During the second season, there was no significant difference between the plants that sprayed with Chelated iron with a concentration of (50 or 100 mg.L⁻¹), in the percentage of its volatile oil amounted of (639, 0.652 %), respectively. However, it was significantly excelled in this trait on the control plants which gave 0.611%. The effect of the spraving with coconut liquid at a concentration of (15% or 30%) was significantly excelled for both seasons by giving it the highest percentage of volatile oil which amounted of (0.628, 0.639%, 0.630, 0.644 %), respectively on the percentage of volatile oil for the control plants which amounted of (0.604, 0.620%). The biinteraction between the two study factors (the spraving with Gibberellic acid and Chelated Iron) had a significant effect on this trait, where the plants treated with Gibberellic acid at a concentration of (300 mg.L⁻¹) and the Chelated Iron at concentration of (100 mg.L^{-1}) were by giving it the highest characterized percentage of volatile oil for both study seasons amounted of (0.668, 0.671 %), respectively, compared to the lowest percentage of volatile oil in the plants that sprayed with (50 mg.L^{-1}) of Chelated Iron and did not spraying with the Gibberellic acid for the first season which amounted to 0.557% and plants that did not spraying with any of the solutions for the second season, which amounted to 0.597%. The bi-interaction between the spraying with the Gibberellic acid and the coconut liquid had a significant effect in this trait, where the plants that sprayed with the Gibberellic acid at a concentration of (300 mg.L⁻¹) and Coconut liquid with a concentration of 30% recorded the

highest percentage of volatile oil and for both study seasons, which amounted of (0.667, 0.661), respectively, compared to the lowest percentage of volatile oil recorded by the plants that were spraved with coconut liquid at a concentration of 30% for the first season which amounted of 0.591% and the control plants for the second season, which amounted to 0.581%. As for the effect of the bi-interaction between the spraying with the Chelated iron and the coconut liquid, it was significant in this trait, the plants that sprayed with Chelated iron at a concentration of (100 mg. L^{-1}) and Coconut liquid at a concentration of 15% for the first season and the plants that sprayed with Chelated iron at a concentration of (100 mg. L⁻ ¹) and Coconut liquid at a concentration of 30%in the second season recorded the highest percentage of volatile oil, which amounted of (0.666, 0.660%), respectively, compared to the lowest percentage of volatile oil recorded by the control plants and for both study seasons which amounted of (0.571, 0.591 %), respectively. The triple interaction between the three study factors spraying with the Gibberellic acid and the coconut liquid had a significant effect in this trait, where the plants that sprayed with the Gibberellic acid at a concentration of (150 mg. L^{-1}), Chelated iron at a concentration of (100) mg. L^{-1}) and Coconut liquid at a concentration of 30% recorded the highest percentage of volatile oil and for the first seasons of the study, which amounted of (0.678 %), respectively, compared to the lowest percentage of volatile oil recorded by the plants that sprayed with (300 mg.L^{-1}) of Gibberellic acid, and did not spraying with Chelated iron and coconut liquid, which amounted to 0.525, In the second season, the plants that sprayed with Gibberellic acid at a concentration of (300 mg.L^{-1}) , the Chelated iron at a concentration of (100 mg. L⁻ ¹) and Coconut liquid with a concentration of 15% recorded the highest percentage of volatile oil amounted of 0.674% compared to 0.574% recorded by the control plants.

Table 4: Effect of the Spraying with Gibberellic Acid, chelated iron and the Coconut liquid and theirinteractions in the percentage of volatile oil (%) for the jasmine Plant for both Study Seasons (2015-2016, 2016-2017).

			Fi	rst seas	on		Sec	ond seas	son
Gibberellic	Chelated	Coconut liquid			Interaction	Co	conut liq	uid	Interaction
Acid	Iron				(Gibberellic				(Gibberellic
$(mg.L^{-1})$	$(mg.L^{-1})$	0	15	30	Acid ×	0	15	30	Acid ×
(Ing.L)	(ing.L)	U	15	30	Chelated	U	15	30	Chelated
					Iron)				Iron)
	0	0.592	0.615	0.636	0.614	0.574	0.621	0.597	0.597
0	50	0.580	0.533	0.558	0.557	0.583	0.594	0.635	0.604
	100	0.605	0.663	0.578	0.615	0.587	0.632	0.651	0.623
	0	0.597	0.654	0.658	0.636	0.590	0.622	0.617	0.610
150	50	0.609	0.633	0.562	0.601	0.650	0.647	0.652	0.650
	100	0.605	0.662	0.678	0.648	0.661	0.664	0.658	0.661
	0	0.525	0.586	0.662	0.591	0.610	0.623	0.647	0.627
300	50	0.654	0.637	0.670	0.654	0.654	0.671	0.667	0.664
	100	0.665	0.672	0.668	0.668	0.668	0.674	0.670	0.671
R.L.S.I	R.L.S.D 0.05				0.024	0.044			0.023
Average	Average effect of		0.629	0.630		0.620	0.639	0.644	
Coconut	Coconut liquid		0.628	0.030		0.620	0.039	0.044	
R.L.S.I) 0.05		0.016			0.018			
					Average	Effect	ofintor	action	Average
Effect of in	nteraction (Gibbere	llic acid	l and	effect of	Effect of interaction Gibberellic acid and			effect of
	coconut	liquid			Gibberellic	erellic coconut liquid			Gibberellic
					acid			acid	
0		0.592	0.603	0.591	0.595	0.581	0.616	0.628	0.608
15		0.604	0.650	0.632	0.629	0.634	0.644	0.642	0.640
30		0.615	0.632	0.667	0.638	0.644	0.656	0.661	0.654
R.L.S.I) 0.05		0.024		0.016	0.023			0.018
						Effect	of inter	action	Average
Effect of	Effect of interaction Chelated Iron and				effect of				effect of
coconut liquid					Chelated	Chelated Iron and coconut liquid			Chelated
					Iron			-	Iron
0		0.571	0.618	0.651	0.613	0.591	0.622	0.620	0.611
50		0.614	0.601	0.597	0.604	0.629	0.637	0.651	0.639
10		0.625	0.666	0.641	0.644	0.639	0.657	0.660	0.652
R.L.S.I) 0.05		0.024		0.016		0.023		0.018

Volatile oil yield (g)

Table (5) indicates the superiority of plants treated with Gibberellic acid at a concentration of (300 mg.L^{-1}) in the volatile oil yield and for both study seasons, which amounted to (0.325, 0.319 g), respectively, compared to the control plants which amounted of (0.241, 0.234 g),

respectively. It may be attributed to the fact that the Gibberellic acid led to a significant increase in the percentage of volatile oil, in the number of primary and secondary branches, thus a significant increase in the total yield of flowers for the plant and all this leads to increase in the plant yield from the volatile oil, in addition to the role of Gibberellic acid in elongation of cells and its expansion, including oily glands, and this is agreed with [10] on the plant Chrysanthemum [11] on the Pelargonium graveolens plant. As for the spraving with Chelated iron, where the plants treated with Chelated iron at a concentration of (100 mg.L^{-1}) for both seasons, was significantly excelled in the volatile oil yield, which amounted of (0.327, 0.314 g) compared to the control plants, where the percentage of its volatile oil amounted of (0.249, 0.247 g), respectively. As for the treatment of plants with coconut liquid, the same table showed the superiority of the plants treated with coconut liquid at a concentration of 30% in Volatile oil yield, amounted of (0.347,which 0.312 g) significantly and for both study seasons on the control plants which amounted of (0.245, 0.248 g), respectively. The significant increase in the oil yield when sprayed with coconut liquid may be explained by its significant effect on increasing the total yield of flowers for the plant as shown in Table (5). As for the biinteraction between spraying with the Gibberellic acid and Chelated iron had a significant effect, the plants treated with Gibberellic acid at a concentration of (300 mg. L^{-1}) and the Chelated iron at a concentration of (100 mg. L^{-1}) recorded the highest Volatile oil yield and for both study season s, which amounted to (0.367, 0.342 g) compared to the lowest Volatile oil yield recorded by plants treated with (50 mg. L^{-1}) of Chelated iron, which was not treated with Gibberellic acid, which amounted to (0.215 g) for the first season and the control plants, which amounted to (0.212 g) for the second season. As for the biinteraction between the spraying with the

Gibberellic acid and the coconut liquid had a significant effect where the plants treated with $(300 \text{ mg. } \text{L}^{-1})$ Gibberellic acid and 30% of coconut liquid gave the highest Volatile oil vield and for both study seasons, which amounted to (0.358, 0.338 g), respectively, compared to the control plants, which gave the lowest Volatile oil yield amounted of (0.198, 0.195 g), respectively. The bi-interaction between the spraying with Chelated iron and coconut liquid had a significant effect on the Volatile oil yield and for both study seasons, where the plants treated with (100 mg. L-1) of Chelated iron, and 30% of coconut liquid was significantly excelled, which recorded the highest Volatile oil yield amounted of (0.358, 0.340 g), respectively, compared with the lowest of the Volatile oil vield, which amounted of (0.198, 0.209), respectively. The results of the table show the significant effect of the triple interaction between the three study factors in the Volatile oil yield, where the plants treated with the Gibberellic acid at a concentration of (300 mg. L⁻¹), Chelated iron at a concentration of (100 mg. L^{-1}) and the Coconut liquid at a concentration of 15% recorded the highest Volatile oil yield amounted of (0.380 g) compared to the lowest Volatile oil yield formed by the plants treated with $(50 \text{ mg. } \text{L}^{-1})$ of Chelated iron only amounted of (0.178 g) in the first season. In the second season, the plants were treated with $(150 \text{ mg. } \text{L}^{-1})$ of Gibberellic acid, $(100 \text{ mg. } \text{L}^{-1})$ of Chelated iron and 30% coconut liquid which recorded the highest Volatile oil yield amounted of (0.375 g) compared to the lowest Volatile oil yield gave it the control plants, which amounted to 0.173 g.

Table 5: Effect of the Spraying with Gibberellic Acid, chelated iron and the Coconut liquid and their interactions in the Volatile oil yield (g) for the jasmine Plant for both Study Seasons (2015-2016, 2016-2017).

			Fi	rst seas	2017).	Second season				
		Coo	conut lic		Interaction	Coconut liquid			Interaction	
Gibberellic					(Gibberellic				(Gibberellic	
Acid	Iron				Acid ×				Acid ×	
$(mg.L^{-1})$	$(mg.L^{-1})$	0	15	30	Chelated	0	15	30	Chelated	
					Iron)				Iron)	
	0	0.187	0.224	0.269	0.227	0.173	0.222	0.241	0.212	
0	50	0.178	0.203	0.263	0.215	0.180	0.219	0.257	0.219	
	100	0.228	0.298	0.320	0.282	0.233	0.279	0.303	0.272	
	0	0.203	0.247	0.306	0.252	0.196	0.240	0.281	0.239	
150	50	0.265	0.308	0.302	0.292	0.249	0.296	0.337	0.294	
	100	0.278	0.342	0.379	0.333	0.310	0.295	0.375	0.327	
	0	0.203	0.263	0.340	0.269	0.258	0.285	0.330	0.291	
300	50	0.231	0.336	0.358	0.338	0.278	0.347	0.342	0.322	
	100	0.346	0.380	0.376	0.367	0.344	0.342	0.341	0.342	
R.L.S.I	0.05	0.063			0.036	0.043			0.023	
Average	Average effect of		0.303	0.347		0.248	0.281	0.312		
Coconut	Coconut liquid		0.505	0.547		0.248	0.281	0.512		
R.L.S.I) 0.05		0.021			0.018				
					Average	Ffoot	ofintor	action	Average	
Effect of in	nteraction (llic acid	l and	effect of Gibberellic acid and				effect of	
	coconut	liquid			Gibberellic		onut liq		Gibberellic	
					acid		-		acid	
0		0.198	0.242	0.284	0.241	0.195	0.240	0.267	0.234	
15		0.249	0.299	0.329	0.292	0.252	0.277	0.331	0.287	
		0.290	0.326	0.358	0.325	0.293	0.325	0.338	0.319	
R.L.S.I) 0.05		0.036		0.021	0.023			0.018	
				_	Average	Effect	of inter	action	Average	
Effect of	Effect of interaction Chelated Iron and					effect of Chelated Iron and			effect of	
coconut liquid					Chelated		onut liq		Chelated	
					Iron		-		Iron	
0		0.198	0.245	0.305	0.249	0.209	0.249	0.284	0.247	
50		0.255	0.282	0.308	0.282	0.236	0.287	0.312	0.278	
10 D L G L		0.284	0.340	0.358	0.327	0.296	0.305	0.340	0.314	
R.L.S.D 0.05			0.036		0.021	0.023			0.018	

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