purification of phytase enzyme from Saccharomyces cerevisiae, study of their kinetic properties and estimation of their molecular weights

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Abstract

The current study aimed to extract and purify the phytase enzyme from dried Saccharomyces cerevisiae yeast (of Iranian origin), determine its specific activity during the various purification stages, and study the optimum conditions for enzyme activity in terms of pH and temperature. The aim was to evaluate its efficiency and potential use in industrial or food applications. A single sample of dried Saccharomyces cerevisiae yeast of Iranian origin was obtained from the local market in Tikrit during September ۲۰۲۳ and used to extract and purify the phytase enzyme and study its specific activity and protein concentration. The purification process was carried out through several sequential stages, including preparation of the crude extract, precipitation with ammonium sulfate, ion exchange chromatography, and gel filtration chromatography. The results showed that the total activity of the crude extract was Yi, 919 enzyme units (IU) and the specific activity was o, YY IU/mg. After precipitation with ammonium sulfate, the total activity increased to \90, T IU and the specific activity to \,\T units/mg, with a 1,10-fold increase in enzyme purity. Following ion-exchange purification, the specific activity increased to \\\.\.\ units/mg, a \\.\\\\-fold purification ratio and a yield of \\\.\.\\\.\.\.\.\. Finally, the gel filtration stage significantly increased the enzyme purity, with the specific extract, with a total yield of YTY, A%. The study showed that the purified phytase enzyme had its highest activity at pH V, o, while the optimum temperature for its activity was The C, where enzyme activity gradually increased with increasing temperature up to this point and then began to decline. Conclusion: The study results demonstrate that Saccharomyces cerevisiae (Iranian type) is an effective source for phytase production, as it can be efficiently purified through successive stages to achieve high specific activity and a clear purity index. The purified enzyme also showed the best activity at nearly neutral acidity (pH \forall_{r} °) and an optimum temperature (\forall_{r} °C), indicating its suitability for industrial or food applications operating under these conditions.

Keywords: Saccharomyces cerevisiae, phytase, purification, kinetic properties

المستخلص:

هدفت الدراسة الحالية اإلى استخلاص وتنقية إنزيم الفايتاز من خميرة Saccharomyces cerevisiae المجففة (من منشأ إيراني)، وتحديد النشاط النوعي له خلال مراحل التنقية المختلفة، إضافة إلى دراسة الظروف المثلى لفعالية الإنزيم من حيث درجة الحموضة (pH) ودرجة الحرارة، وذلك بهدف تقييم كفاءته وإمكانية استخدامه في التطبيقات الصناعية أو الغذائية. تم الحصول على عينة واحدة من خميرة Saccharomyces cerevisiae المجففة من السوق المحلى في تكريت خلال شهر أيلول ٢٠٢٣، وهي من منشأ إيراني، واستخدمت لغرض استخلاص وتنقية إنزيم الفايتاز ودراسة نشاطه النوعي وتركيز البروتين. تمت عملية التنقية من خلال عدة مراحل متتابعة شملت: تحضير المستخلص الخام، الترسيب بكبريتات الأمونيوم، الكروماتوغرافيا بالتبادل الأيوني، وكروماتوغرافيا الترشيح الهلامي. أظهرت النتائج أن النشاط الكلى للمستخلص الخام بلغ ٩١٩,٩١٩ وحدة إنزيمية (IU) والنشاط النوعي ٥,٢٣ وحدة/ملغم. بعد الترسيب بكبريتات الأمونيوم ارتفع النشاط الكلي إلى ١٩٥,٣ والنشاط النوعي إلى ٦,٠٣ وحدة/ملغم، مع زيادة في نقاوة الإنزيم بمقدار ١,١٥ مرة. أما بعد التنقية بالتبادل الأيوني فقد ارتفع النشاط النوعي إلى ١٤,٨ وحدة/ملغم بنسبة تنقية ٢,٨٣ مرة، وبمردود ٢,٥٠٦٪. وأخيرا، أدت مرحلة الترشيح الهلامي إلى زيادة كبيرة في نقاوة الإنزيم حيث بلغ النشاط النوعي ٨, ٦٦ وحدة/ملغم، بمعدل تنقية ٥٧,٧ مرة مقارنة بالمستخلص الخام، وبمردود كلي بلغ ٨, ٢٣١.% . أظهرت الدراسة أن إنزيم الفايتاز المنقى يمتلك أعلى نشاط عند درجة pH تساوي ٥,٧، بينما كانت درجة الحرارة المثلى لنشاطه ٣٦°، حيث يزداد النشاط الإنزيمي تدريجيا مع ارتفاع درجة الحرارة حتى هذه النقطة ثم يبدأ بالانخفاض بعدها . الاستنتاج: تظهر نتائج الدراسة أن خميرة Saccharomyces cerevisiae (النوع الإيراني) تعد مصدرا فعالا لإنتاج إنزيم الفايتاز، إذ أمكن تنقيته بكفاءة عالية من خلال المراحل المتتابعة حتى الوصول إلى نشاط نوعى مرتفع ودليل نقاوة واضح. كما تبين أن الإنزيم المنقى يتميز بأفضل نشاط عند درجة حموضة متعادلة تقريبا (٣,٥) ودرجة حرارة مثلى (٣٦°C)، مما يشير إلى ملاءمته للتطبيقات الصناعية أو الغذائية التي تعمل ضمن هذه الظروف.

الكلمات المفتاحية: خميرة Saccharomyces cerevisiae ، فيتاز، تنقية، خصائص حركية

Introduction

Phytases are hydrolytic enzymes (phosphatases) that are part of the histidine acid phosphatase subfamily. They facilitate the hydrolysis of phytate phosphomonoester linkages (salts of myo-inositol hexakisphosphate) or myo-inositol 1, 7, 7, 4, 0, 7-hexakis dihydrogen phosphate (phytic acid), yielding

derivatives such as tetra, tri, di, and inositol monophosphate, in addition to inorganic phosphate (Pi) [1,7].

Phytase producing microorganisms include filamentous fungi of the genus Aspergillus. In various studies, these fungi were found to produce the most active extracellular enzyme with the most suitable characteristics of both pH and temperature stability. Hence, microorganisms of the genus Aspergillus, are the most used in the industrial production of this enzyme [7,٤]. Solid-state fermentation is a process commonly applied for the production of extracellular enzymes [9].

As a result of the favorable characteristics of phytase, as well as its practical application as an additive in the diets of non-ruminants, this enzyme has taken a position of great interest in biotechnological applications to reduce the phytate content of fodder and commercial foods [7]. Phytases used as additive must be effective in releasing phosphate from phytic acid and must demonstrate their effectiveness at the digestive tract level, as well as withstand the conditions of pH and temperature [7 , 5].

Materials and methods

A single sample of S. cerevisia dried yeast, which was taken from the local market in Tikrit in September Y.YY. It is an Iranian variety, with one replicate to test the specific activity and protein concentration during the extraction and precipitation phase. Local yeast is prepared using the common traditional methods, using only flour and water without any additives. A certain amount of flour, Yo grams, is placed in a glass bottle with a tight lid. The same amount of distilled water (Yo ml) is added to the flour and mixed well. The container is covered, leaving a small opening for air to enter for respiration. The mixture is left for Yt hours at room temperature (YooC). On the second day, Yo grams of flour and water are added, mixed, and left under the same conditions. On the fifth day, after all the additions are complete, the container is tightly closed and placed in the refrigerator at to C for later use.

Phytase Enzyme Extraction and Production

Weigh ' · · g of prepared baker's yeast and mix it with ° · ml of organic toluene solvent. Place the mixture in a water bath at ٤°°C for one hour. Leave the mixture at room temperature at ' °°C for ' - " hours, then add ' · · ml of cold distilled water. Place the mixture in a glass separating funnel and shake well for half an hour. Then, incubate at ٤°C for ' ^ hours. Filter the mixture, collect the separated aqueous phase, and centrifuge it in a refrigerated centrifuge at ' · · · · · rpm at ٤°C for ' · minutes. The filtrate was taken and the precipitate discarded. Phytase activity and protein concentration were measured.

Measuring the activity of the phytase enzyme from baker's yeast

•, 7 ml of the crude extract prepared was added to •, 7 ml of the basic solution and •, 0 ml of the phytase solution. Phytase activity was evaluated as described by [9].

Estimation of Total protein

The method of ['\'] was followed in estimating the amount of protein in the enzyme solution extracted from baker's yeast S. cerevisiae, as well as in all stages of extraction, purification and characterization. This method depends on the reaction of the basic copper solution (basic copper sulfate) present in the contents of the reagent solution with the nitrogen atoms present in the peptide bonds of the protein to produce a complex with a purple color.

Phytase Enzyme Purification

Ammonium Sulfate Precipitation:

Ammonium sulfate (SO[£] \(\cdot\)(NH[£]) was used in the crude enzyme extract at saturation levels ranging from \(\cdot\). These levels were tested individually to achieve the best saturation levels. These weights were added gradually to the extract in an ice bath under continuous stirring using a magnetic stirrer for an hour. The components were separated using a refrigerated centrifuge at \(\cdot\). The rpm at \(\xi\)°C for \(\cdot\) minutes. This process was repeated by taking the filtrate and adding ammonium sulfate to reach \(\cdot\). saturation under the same conditions as before. The filtrate was discarded, and the precipitate was dissolved in \(\cdot\)° ml of

•, \ M basic buffer at pH \(\). Absorbance readings were taken at \(\) \(\) nm to measure enzyme activity \(\) \(\) \(\).

Purification by membrane screening (dialysis)

The enzyme produced from the ammonium sulfate precipitation step was dialysis. The enzyme was placed in a dialysis bag, which allows the passage of substances less than '\(\gamma\) kDa. The bag was placed in a container containing basic buffer solution for '\(\gamma\) hours, which separated the saline solution from the enzyme, leaving the enzyme inside the bag. The solution was concentrated using sucrose. Enzyme activity, specificity, and protein concentration in the sample were determined. The sample was stored at -\(\gamma\) oC until further purification steps were performed[\(\gamma\)]

Ion-exchang chromatography

This purification was done in this way based on what was mentioned by [\\].

Sephadex G-Y..

Sephadex G- $^{\Upsilon}$ ·· column was prepared and filled according to the manufacturer's instructions (Pharmacia Fine chemicals). $^{\Upsilon}$ · g of Sephadex powder was suspended in $^{\Upsilon}$ ·· ml of distilled water to remove preservatives. The column was then heated in a water bath for $^{\circ}$ hours at $^{\Upsilon}$ · $^{\circ}$ C, then left to cool and washed twice with $^{\Upsilon}$ · $^{\Upsilon}$ · M tetraacetic acid-base buffer solution, pH $^{\Upsilon}$ · $^{\Upsilon}$. Degassing was performed using a laboratory vacuum pump. The gel was filled into a glass column with dimensions of $(^{\Upsilon}$ · $^{\circ}$ C $^{\Upsilon}$ · $^{\circ}$ C and the material was left to settle. The column was then calibrated using tetraacetic acid-base buffer solution at a flow rate of $^{\Upsilon}$ · ml/h and $^{\circ}$ ml per part $[^{\Upsilon}]$.

Kinetic study

Determining the pH

To determine the optimum pH for enzyme stability, buffer solutions with a concentration of ',' M and varying pHs ranging from ' to '' were prepared according to paragraph ($^{\circ}$ - $^{\circ}$ - $^{\circ}$). Fifty microliters of purified enzyme were added

Determining the optimum temperature for enzyme activity:

To determine the optimum temperature for enzyme activity, $\circ \cdot$ microliters of purified enzyme were added to test tubes containing \cdot , \cdot M basic buffer solution at pH \vee . The tubes were incubated for \cdot minutes at temperatures of \circ , \cdot , \cdot , \cdot , and \cdot \cdot \cdot C [9].

Result

The purification process of phytase enzyme from Saccharomyces cerevisiae was carried out through several sequential steps, including crude extract preparation, ammonium sulfate precipitation, ion-exchange chromatography, and gel filtration chromatography. The results of the purification steps are summarized in Table (1). The crude extract showed a total activity of \(\frac{1}{2}, \frac{1}{3}\) IU and a specific activity of \(\frac{1}{2}, \frac{1}{3}\) IU, while the specific activity increased to \(\frac{1}{3}, \frac{1}{3}\) U/mg. The yield at this stage was \(\frac{1}{3}, \frac{1}{3}\) o-fold increase in enzyme purity. Subsequent purification by ion-exchange chromatography resulted in a marked increase in specific activity to \(\frac{1}{2}, \hat{1}\) U/mg, corresponding to a \(\frac{1}{3}, \hat{1}\) of old purification, with recovery yield of \(\frac{1}{3}, \hat{1}\) U/mg and a \(\frac{1}{3}, \hat{2}\) U/mg and a \(\frac{1}{3}, \hat{2}\) o-fold purification compared to the crude extract. However, the overall recovery at this stage was \(\frac{1}{3}, \hat{2}\) of as shown in table (1) and Figure(1).

Filtration

Purificatio n Steps	Total Volum e mL	Activity IU/mL	Total Activit y	Protein g/mL	Total Protei n	Specific Activity U/mg	Recover y Yielld %	Fold of Purificati on Total Protein
Crude	١٣	0,777	V£,919	1,1•	1 £,٣	0,789.9	1,	1
Precepitati on	١٨	1.,00	190,7.	١,٨٠	٣٢,٤	7,• YVVV A	۱۸۸,۲۷	1,10
Ione- Exchange	٥	11,1000	09,70.	• , \ • • • •	٤	18,1170	7.0,77	۲,۸۳
Gel	٥	17,77	٦٦,٨٠٠	.,۲	١	77,// ***	771,77	17,70

Table (1) Purification of phytase enzyme from Saccharomyces cerevisiae

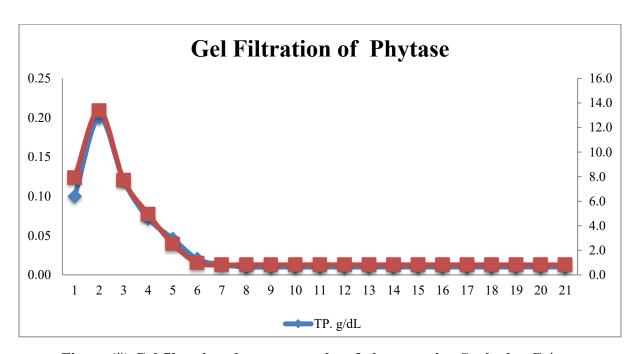


Figure (*) Gel filtration chromatography of phytase using Sephadex G-1 · · ·

The effect of the substrate concentration on the reaction rate of the partially purified phytase enzyme from Saccharomyces cerevisiae was studied. The optimal substrate concentration is used when estimating the enzyme activity in order to obtain the maximum speed. Therefore, different concentrations were used as shown in Figure (1).

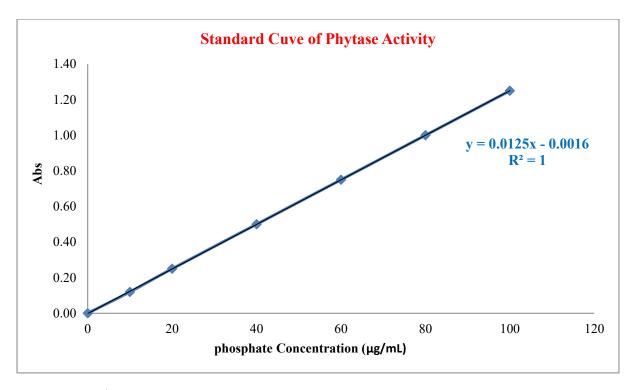


Figure (1) shows the Michaelis-Menten diagram showing the effect of substrate concentration on phytase enzyme activity.

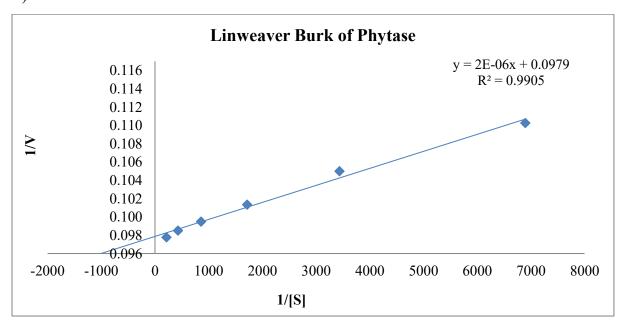


Figure (7) Lineweaver–Burk diagram of partially purified phytase enzyme

Different temperatures were tested to determine the optimum temperature for phytase activity ($^{7} \cdot -^{7} \cdot -^{5} \cdot -^{0} \cdot -^{7} \cdot ^{0}$ C). The optimum temperature was 77 °C. The results showed that increasing the temperature led to an increase in enzyme activity, followed by a decrease in their activity, as shown in the Figure.(7)

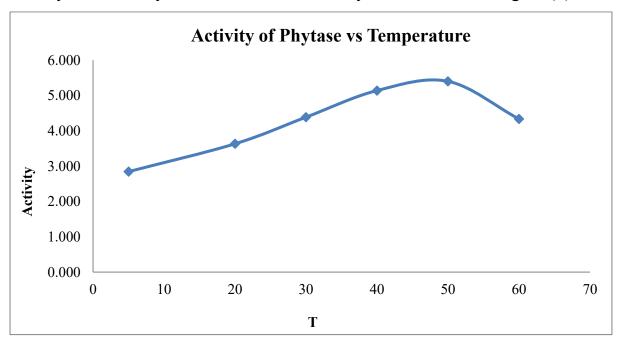


Figure (*) Effect of temperature on phytase enzyme activity

The different effects of pH on phytase activity were studied as shown in Figure ($^{\xi}$). For the pure enzyme, it was found that the highest enzyme activity was achieved at pH.($^{\vee}$, $^{\circ}$)

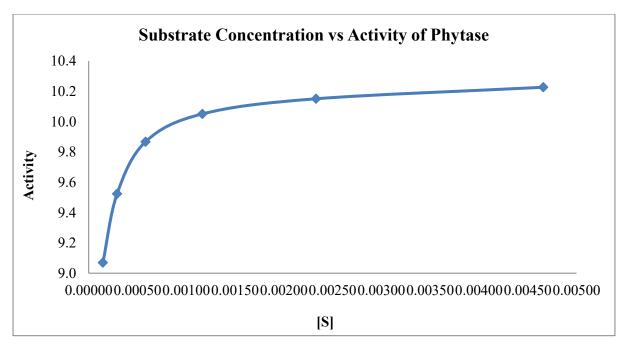


Figure (4) Effect of pH on phytase activity

Discussion

 removal of inhibitors or enzyme activation during partial purification, a phenomenon also reported by [17].

The final purification stage—gel filtration chromatography—achieved the highest enzyme purity, with a specific activity of T, A IU/mg and a Y, Vo-fold purification relative to the crude extract. Gel filtration separates proteins based on molecular weight, thus effectively eliminating low-molecular-weight contaminants and aggregates [YV]. Similar purification folds were obtained in previous studies purifying S. cerevisiae and Aspergillus niger phytases [YA], confirming the reliability of this multi-step approach.

Regarding enzyme characterization, the purified phytase exhibited maximum activity at pH V,o and TToC, aligning with the near-neutral pH range optimal for yeast phytases. This finding corresponds with earlier reports showing that S. cerevisiae phytase is active under slightly neutral to mildly acidic conditions (pH T,o-V,o), which is favorable for food and feed processing environments [19]. The optimal temperature of TToC also matches the mesophilic nature of S. cerevisiae, suggesting the enzyme's stability under physiological and moderate industrial temperatures [Y ·]. Beyond this temperature, the observed decline in enzyme activity may be attributed to conformational denaturation, as phytase's tertiary structure is sensitive to heat [17].

Phytase activity is highly sensitive to both pH and temperature, as both factors affect the enzyme's spatial structure and reaction rate. Studies indicate that most fungal phytases, particularly those extracted from Aspergillus niger, operate at their maximum efficiency in relatively acidic environments, with optimal activity recorded at pHs between [£], ^o and ^o, ^o [¹ ⁹, ⁷ ¹]. This property makes them suitable for use in the acidic environment of the human and animal stomachs, which explains their widespread adoption in the poultry and pig feed industries to improve phosphorus release from phytate.

Bacterial phytases, on the other hand, have demonstrated broader pH optimum ranges, with good activity observed in neutral and even slightly alkaline media

(pH ¬,·-¬,°) [¬]. This broad pH optimum range makes them an important choice for industrial applications requiring stability under diverse processing conditions, such as food production or the bioremediation of agricultural waste. Regarding temperature, most studies agree that the optimal activity of most fungal phytases is found at temperatures between ٤° and ¬·°C [¬¬]. However, the stability of activity at high temperatures varies depending on the source. Bacterial phytases, particularly those from the genus Bacillus, have been shown to retain a significant portion of their activity at ¬·-¬·°C, making them suitable for heat-processing feeds [¬¬]. In contrast, plant phytases experience a marked decline in activity at temperatures above ¬·°C due to thermal decomposition and loss of the enzyme's tertiary structure [¬² ½].

Conclusion

The results confirm that Saccharomyces cerevisiae is an efficient and eco-friendly source of phytase. The enzyme's biochemical properties—neutral pH and moderate temperature optima, make it highly suitable for use in food processing and feed applications to enhance mineral bioavailability and reduce antinutritional phytate content.

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